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**Importance of stable carbon isotopes for  
studying dynamics of methanogenesis in  
rivers**

Ph.D. Thesis

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I, Adam Bednařík, thereby declare that I wrote the Ph.D. thesis myself using results of my own work or collaborative work of me and colleagues and with help of other publication resources which are properly cited.

Olomouc, July 2019

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### **Abstract**

Methane is one of the most important greenhouse gases. Despite of recent studies pointing out important contribution of running waters to natural methane emissions to the atmosphere, data concerning the methane sources in rivers are very scarce. This thesis deals with methane dynamic in river ecosystems with special emphasis on an effect of river impoundments on methane related processes. Beside the methane concentrations, oxidation, production and emission to the atmosphere, the changes in contribution of two main methanogenic pathways (hydrogenotrophic and acetoclastic) to the total methane production were determined using the stable carbon isotopes analysis. We found hotspots of the sediment methane production in a river continuum, which are connected with the local driving factors including mainly the existence of artificial barriers as weirs. Changes in rate of individual components of river methane dynamic were further examined in cascade of three weirs and river reaches between them. We found that river impoundments affect the sediment processes in several ways, including changes of the sediment characteristics (fine sediment fraction, higher sediment carbon content), enhanced microbial activities in the sediment (methane production and oxidation), ebullition of methane, and different contributions of hydrogenotrophic methanogens to the released methane. Thus, many parameters found for weir impoundments resemble observations for lake systems. Moreover, remarkable spatial variability in sediment methane production was demonstrated in cross-section profile of the one studied impoundment. Presented studies point to only the part of the samples could be activated for methane production despite of presence of methanogens (most probably due to substrate limitation). This suggest that the observed variability of the microbial activities as well as the resulting methane concentrations in the water column are only indirectly linked to the presence of different microbial guilds, but rather affected by their activity. Altogether our results confirm that the methane dynamics in a river system show a high local variability and that multiple measurements are needed to characterize the sources and fates of the methane. Obtained results might be further used for better estimates of importance of rivers in a global methane budget.

## Abstrakt

Metan je spolu s oxidem uhličitým a oxidem dusným řazen mezi nejvýznamnější skleníkové plyny. Navzdory mnoha současným studiím vyzdvihujícím význam říčních ekosystémů jako přírodního zdroje emisí metanu do atmosféry, jsou poznatky týkající se zdrojů metanu v řekách nedostatečné. Předložená práce se zabývá dynamikou metanu v říčních ekosystémech se zvláštním důrazem na vliv umělých příčných bariér (jezů) na procesy spojené s koloběhem metanu. Kromě koncentrací metanu, jeho oxidace, produkce a emisí do atmosféry byl determinován příspěvek dvou hlavních metabolických cest vzniku metanu do jeho celkové produkce, a to s využitím analýzy obsahu stabilních izotopů uhlíku. Na základě provedených měření v říčním kontinuu řeky Labe byly detekovány místa s nezvykle vysokou produkcí metanu v sedimentech, které odpovídaly změnám v lokálních faktorech prostředí spojených především s existencí příčných bariér na vodním toku. Další práce se proto blíže zaměřila na porovnání změn v jednotlivých složkách dynamiky metanu v kaskádě tří jezů a říčních úsecích mezi nimi. Bylo zjištěno, že jezy vyvolávají řadu změn v sedimentačních procesech (usazování jemnější frakce, vyšší obsah organického uhlíku), které se následně projevují v procesech spojených s koloběhem metanu, jako je zvýšená mikrobiální aktivita (produkce a oxidace metanu v sedimentech), vysoký podíl ebulice na celkové emisi do atmosféry (uvolňování metanu ve formě bublin) a rozdílný poměr metabolických cest vzniku metanu. To spolu s ostatními charakteristikami nadjezí vypovídá o tom, že mnoho procesů probíhajících v sedimentech jezových zdrží včetně tvorby bublin a metabolismu uhlíku je lépe srovnatelných s prostředím sedimentů jezer než s říčními sedimenty. Mimo to byla zjištěna také značná variabilita v produkci metanu uvnitř vybraného nadjezí. Další výsledky ukázali, že jen část vzorků inkubovaných v anoxických podmínkách produkuje metan, přestože v nich bylo detekováno srovnatelné množství metanogenních archaea jako v aktivních vzorcích, což bylo s největší pravděpodobností dáno nedostatkem vhodného substrátu. Toto zjištění naznačuje, že pozorovaná variabilita v mikrobiální aktivitě stejně jako výsledné koncentrace metanu ve vodě jsou jen nepřímou řízeny přítomností určitého mikrobiálního společenstva, ale jsou spíše ovlivněny jeho aktivitou. Naše studie tak potvrzuje, že dynamika metanu v říčních ekosystémech vykazuje vysokou prostorovou variabilitu a z toho důvodu lze charakterizovat zdroje metanu a jeho další osud v ekosystému jen s využitím velmi komplexních měření. Získaná data mohou mimo jiné posloužit i jako

cenný údaj pro zpřesnění odhadů významu říčních systémů v bilanci metanu v rámci vnitrozemských vod a v kontextu globální dynamiky metanu.

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## **List of publications**

The thesis is based on the following papers:

### **Paper I:**

**Bednařík A**, Blaser M, Matoušů A, Tušer M, Chaudhary PP, Šimek K, Rulík M (2019): Sediment methane dynamics along the Elbe River. *Limnologica* (*in review*)

### **Paper II:**

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## **1. Introduction**

### **1.1. Atmospheric methane cycle**

Methane (CH<sub>4</sub>) is one of the most potent greenhouse gases with a global warming potential ~28 times higher than carbon dioxide (CO<sub>2</sub>) over time horizon of 100 years and represents about 15 % of the anthropogenic greenhouse effect (IPCC 2013). From the mid-Holocene to about 300 years ago atmospheric methane concentration rose steadily by about 25 % (Brook et al. 2000). With human population increase and industrialization, methane concentration is now about 250 % higher than it was in the preindustrial age (Etheridge et al. 1998). Current atmospheric methane concentration is 1858 ppb (Dlugokencky 2018).

While the sinks of the methane in the environment are quite clear (reaction with hydroxyl radicals in the troposphere or oxidation by methanotrophic bacteria), the sources of methane are much more diverse. Among main atmospheric methane sources are included wetlands, ruminants, termites, oceans, freshwater sediments, landfills, biomass burning and fossil methane released during fossil fuel extraction (Wuebbles and Hayhoe, 2002). The most of the methane is produced microbiologically, while contribution of freshwater habitats (wetlands, rice fields) creates ~33 % of the annual atmospheric methane flux (Conrad 2009). Moreover, recent studies estimate that annual methane emission from fluvial ecosystems is equivalent to 20-50 % of lake or wetland effluxes (Stanley et al. 2016). Hence, climate driven fluctuations of methane emissions from natural wetlands are the main drivers of the global inter-annual variability of methane emissions (high confidence), with a smaller contribution from the variability in emissions from biomass burning emissions during high fire years (IPCC 2013). Our incomplete understanding of the global methane budget is in part due to the difficulty in quantifying emissions from all of the diverse methane sources (Saarnio et al. 2009, Bastviken et al. 2011).

Isotopic composition of atmospheric methane is useful for recognition of individual methane sources, when the methane originating from pyrogenic sources reaches values more close to the original source (organic matter) than in case of biogenic methane, which is characterised by high fractionation associated with metabolic processes during its production (Chanton et al. 2005).

The 2000-year methane concentrations together with stable carbon isotopes of methane, which have been derived from ice cores, revealed remarkable correlations between fluctuation of contribution of individual methane sources to the atmosphere

and changes in human population growth, eventually climate change (White et al. 2007). The study of White et al. (2007) demonstrates, that knowledge of  $\delta^{13}\text{CH}_4$  allows to reveal important changes in the methane budget that are hidden in the simple methane concentration record. Thus, knowledge of the isotopic composition of source of methane emitted from natural and anthropogenic systems is helpful for developing a global budget for methane sources and sinks (Chanton et al. 2005)

## **1.2. Methane dynamic in lotic ecosystems**

### **1.2.1. Components of methane dynamics in rivers**

First comprehensive study summarizing the methane ecology in running waters is a review by Stanley et al. (2016). In short, methane dynamic in lotic ecosystems consists of (1) production of the methane within hyporheic sediments, which are place of anaerobic metabolism and formation of methane, (2) subsequent diffusion of methane to the surface water, where the methane is (3) transported downstream or (4) emitted to the atmosphere. The methane is also a subject of significant (5) oxidation by methane oxidizing bacteria during its transport in lotic ecosystems. Methanotrophs are often found at the anoxic/oxic interface of various habitats including freshwater sediments, where they consume the methane arising from methanogenesis and are thus able to reduce the most of the potential methane flux to surface water (Segers 1998, Trimmer et al. 2010). Moreover, recent research implies that anaerobic oxidation of methane (AOM) might occur in freshwater sediments via denitrification (Ettwig et al. 2010, Norði and Thamdrup 2014), via sulfate reduction (Beal et al. 2011; Norði et al. 2013), and via iron reduction (Norði et al. 2013). It was found that AOM can be widespread in freshwater lake sediments and accounts for one third of the mean total methane produced in surface and near-surface lake sediments (Martinez-Cruz et al. 2018). The most recent study indicates that AOM activity can be important in the reduced sandy riverbeds, where the nitrite-dependent and nitrate-dependent AOM are the dominant AOM pathways (Shen et al. 2018).

This brief description can be further completed by ebullition (escape of methane from sediments directly to the atmosphere in form of bubbles). This process bypasses the importance of oxidation by methanotrophic bacteria and hence, ebullition of sediment gas bubbles is an important transport process accounting up to 60 % of the total methane emissions from the fluvial ecosystem (Wilcock and Sorrel 2008).

Last but not least, methane has been recently recognised as a potentially important carbon and energy source for freshwater food webs due to conversion of methane to microbial biomass by methane oxidation bacteria, which can be highly productive (Jones and Grey 2011). In rivers, grazing methane-oxidizing bacteria could provide the caddis larvae (genuses *Agapetus* and *Silo*) up to 30 % of their carbon (Trimmer et al. 2009).

Important elements modifying the methane dynamics in lotic ecosystems due to accumulation of sediments and organic matter are artificial impoundments (e.g. Maeck

et al. 2013). Moreover, it was found that smaller impoundments have greater sediment accumulation rates per unit area than the large ones, while small impoundments create a significant part of the total area of impoundments (Downing et al. 2006, Downing et al. 2008). These facts further influenced direction of this thesis.

Two main parts of methane dynamics in river ecosystems, methane inputs and outputs, are reviewed with more detail in next two subchapters (1.2.2. and 1.2.3.).

### **1.2.2. Methane inputs to river ecosystem**

Despite of significant contribution of methane emissions from rivers to global methane flux, there is a paucity of data concerning sources of methane to these freshwater ecosystems (IPCC 2013, Bastviken et al. 2011). Recent studies pointing out supersaturation of river water by methane and quantifying emissions from various inland waters deal with methane origin only marginally (e.g. Middleburg et al. 2002, Saarnio et al. 2009, Anthony et al. 2012). Basically, there are three main sources of methane to surface water of fluvial ecosystems: drainage of surrounding methane rich habitats, groundwater input and river sediments, while exact role of each of these sources is not yet quantified in the overall river C budget. Moreover, water inflows enriched in methane from wastewater treatment plants are significant source of methane in the human influenced rivers (Alshboul et al. 2016).

Some studies suggest that an increased concentration of methane in rivers comes from the drainage of the surrounding wetlands. For instance, drainage of methane rich peatland and riparian soils were recognized as a main source of dissolved methane in first-order stream (Hope et al. 2004). Wetlands as an important source of methane to the adjoining rivers were recognized for example by Borges et al. (2015) in the African rivers, when methane was positively related to wetland fraction of the catchment surface. Similarly, spatial heterogeneity of methane concentration in Zambezi River was found to be mainly determined by the connectivity with surrounded floodplains and wetlands (Teodoru et al. 2015).

Groundwater methane input is highly influenced by type of the aquifers and rate of surface-subsurface water exchange. Although methane-rich groundwater was found to significantly contribute to surface water methane concentration (Jones and Mulholland 1998), groundwater with low methane concentration in the shallow and unconfined aquifers (mostly fractured limestones) did not contribute significantly to the high methane levels in surface water of a River Meuse basin (Borges et al. 2018) as well

as groundwater from a chalky areas in UK had marginal effect on the total methane emissions (Goody and Darling 2005). All studies point out, that high heterogeneity in methane production and hydrologic exchange between groundwater and surface water results in high spatial variation in methane input from this source to the rivers.

More likely, methane concentration in surface water is rather a result of both the groundwater discharge together with a stream-bed methanogenesis (Atkins et al. 2017, Call et al. 2018). Previous studies showed that methane in hyporheic zone creates substantial part of interstitial DOC and also high methanogenic and methanotrophic biomass were detected through the vertical profile of these sediments (Boulton et al. 1998, Fischer et al. 2005, Hlaváčová et al. 2005, Buriánková et al. 2012, Brablcová et al. 2015). However, role of these microorganisms in direct methane supply to surface water and to whole ecosystem production and respiration remains poorly known (Stanley et al. 2016). Moreover methane input from river sediments is usually calculated from concentration differences corrected by effects of porosity and tortuosity, but only rarely directly measured (De Angelis and Scranton 1993, Huttunen et al. 2006, Sollberger et al. 2014).

Generally, methane production and consumption in river sediments is affected by a number of factors including gradients of dissolved oxygen, temperature, organic matter or sediment deposition (Findlay 1995, Fischer et al. 2005). First of all, methane production potential is significantly affected by organic matter content in sediment (e.g. Conrad et al. 2011, Comer-Warner et al. 2018). Methane production is therefore associated mainly with local patches and deposits of fine sediments, where organic matter is accumulated and dissolved oxygen occurs at low level (Sanders et al. 2007, Baulch et al. 2011). From this point of view, land-use can play important role, as the concentrations of dissolved methane in rivers increased with fraction of agriculture in the catchment owing to a larger delivery of organic matter (Borges et al. 2018) as well as sedimentation rate of organic matter is increased in river impoundments (e.g. Maeck et al. 2013). Increasing temperature can further accelerate the methane production in the river sediments (e.g. Yang 1998, Fey et al. 2004, Wilkinson et al. 2015), which leads to increasing concentration of dissolved methane in surface water during the summer months (Middleburg et al. 2002, Yang et al. 2012) and during the low water level (Borges et al. 2018). Recently, non-linearity and threshold responses of streambed methane production were observed with increased temperature implying the more complex estimation of methane fluxes in future (Comer-Warner et al. 2018). On the

other hand, the calculated kinetic and temperature responses showed that with increasing temperature, methane oxidation has potential to respond rapidly to increasing methane production and thus mitigating efflux of methane diffusing through the anoxic–oxic sediment layer (Shelley et al. 2015). Combination of these multiple controls results in a high spatiotemporal heterogeneity of methane fluxes through the sediment–water interface and their quantification and extrapolation is rather complicated (Bednařík et al. 2015). Nevertheless, methane concentration in surface sediment layer (0–10 cm) appears to be the most important for diffusion of methane through the sediment–water interface (Sollberger et al. 2014).

Usually, deeper sediment layers in streams and rivers are characterized by lower oxygen level and higher methane concentration, which can be oxidized by methanotrophs in the surface sediments or even deep in the sediments (Rulík et al. 2013). Despite of the theoretically better conditions for methanogens in deeper sediment layers of rivers, an analysis of vertical distribution of methanogens show that the methanogenic community in methane-emitting river sediments is relatively stable in absolute numbers along a vertical profile (irrespective of the methane production) not only on the level of total archaea and total methanogens but also on the level of the three dominant methanogenic orders (Mach et al. 2015, Chaudhary et al. 2017). Moreover, it was found that initial steps of organic matter degradation, which are catalysed by hydrolytic and fermenting bacteria, are rate-limiting for methane production and the decrease in bacterial numbers reflects a similar decrease in methane production (Chan et al. 2005). These findings suggest that the sediment methanogenic potential is not only limited by the presence of the different methanogens but also more likely regulated by environmental factors (e.g. substrate availability, metabolic activity of the microbial community) as well as the activity of certain members of the methanogenic community.

### **1.2.3. Methane emissions from river ecosystem**

The largest and most ecologically significant pathways of methane efflux from natural environments to the atmosphere are diffusion, ebullition (escape of methane in gas bubbles directly from the sediments) and passage through vascular plants.

Plant transport is the dominant route for methane release mainly from wetlands dominated by emergent aquatic plants, while it is marginal in the fluvial ecosystems (Bridgham et al. 2013). Nevertheless, macrophytes provide labile organic matter to

microbes (root exudates and root decay) and plant-mediated gas transport to the atmosphere reduces opportunities for methane oxidation at the sediment surface or water column (Yavitt and Knapp 1998, Bhullar et al. 2014). Moreover, vascular plants can trap and accumulate the fine and organic rich sediments that support local methane production in streams (Sanders et al. 2007).

In natural streams and rivers both methane diffusion and ebullition were recognized to contribute significantly to the total methane emissions from these ecosystems. However, compared to other natural ecosystems, very scarce data were available for estimation of their global significance and thus importance of lotic ecosystems contribution to global methane emissions from natural environments were overlooked in the past (Saarnio et al. 2009, Bastviken et al. 2011). Fortunately, a number of studies dealing with quantification of methane emissions from fluvial waters were published since that time.

Methane diffusion from rivers to the atmosphere results from widely occurred oversaturation of surface waters compare to atmospheric equilibrium. Rate of methane diffusion through the water-air interface (gas exchange velocity) is influenced mainly by concentration gradient between dissolved methane in surface water and in an ambient air, flow velocity, water turbulence, temperature, eventually wind speed (Jahne and Haubecker 1998, Natchimuthu et al. 2014, McGinnis et al. 2016, Noss et al. 2018). The methane diffusive fluxes from stream and rivers can reach mean value  $8.22 \pm 25.50 \text{ mmol m}^{-2} \text{ d}^{-1}$  based on literature reviewed by Stanley et al. (2016). It is significantly lower per unit of area compared to lake methane fluxes (including ebullition) with mean  $33 \text{ mmol m}^{-2} \text{ d}^{-1}$  estimated by Bastviken et al. (2004).

Recently, methane ebullition was found to participate significantly on a total methane emission from streams and rivers with wide range of contribution from 0 to 80 % and with the remarkable spatial and temporal heterogeneity of this process (Wilcock and Sorrel 2008, Baulch et al. 2011, Crawford et al. 2014, Spawn et al. 2015). The methane ebullition results from very low solubility of methane in freshwaters (saturation in freshwater is about  $1.6 \text{ mol m}^{-3}$  at  $20 \text{ }^\circ\text{C}$ , i.e.  $\sim 27$  times lower solubility than  $\text{CO}_2$ ) and therefore high concentrations of methane lead to production of bubbles and loss to the atmosphere by ebullition (Yammamoto et al. 1976, Sander 2015). Moreover, methane ebullition (as well as plant-mediated fluxes) allows methane leaving a river to bypass zones of aerobic oxidation.

The proportion of methane transported by ebullition is mainly driven by local conditions in rivers, such as sediment accumulation, low water depth, high organic carbon content, high temperature and low oxygen penetration, allowing the development of high methane production and increasing the probability of ebullition occurrence (Baulch et al. 2011). In natural rivers, above described changes can be reached in debris dams (Lancaster and Grant 2006), sediment accumulation due to vegetation (Sanders et al. 2007) or by channel modifications (damming) by beavers (Lazar et al. 2014). For instance, the methane emissions from beaver pond measured by Ford and Naiman (1988) even reached 33-fold higher values than from adjacent river reaches.

In man-altered rivers, artificial impoundments reduced flow water velocity and thus increase water residence time that allows organic matter sedimentation and development of anoxic conditions suitable for methane production (e. g. Maeck et al. 2013, Wilkinson et al. 2015, Crawford et al. 2016). The river impoundments (reservoirs, dams, weirs) have been recognized as significant source of methane emissions to the atmosphere, while they overlap emissions observed in natural lakes (St. Louis et al. 2000). Existence of these artificial barriers play important role in resulted contribution of different methane evasion pathways. Molecular diffusion is usually dominant pathway in rivers (with exceptions described above), while ebullitive emissions are the dominant way for methane emissions from the surface of tropical reservoirs, and it is less significant way for methane emission at the air–water interface in the temperate reservoirs, where the diffusive fluxes are prevailing (Yang et al. 2014).

Impoundments affect also downstream river reaches, where the river water is enriched with methane from the increased methane production in reservoirs, while the most of this methane is degassed at the spillways or the turbine outflows (Abril et al. 2005). This additional release of methane can create the dominant part of total methane emissions and should be further considered in assessments of methane emissions from reservoirs (Li and Zhang 2014, Kemenes et al. 2016).



### 1.3. Use of stable isotope analysis for determination of methanogenic pathways

#### 1.3.1. Biological methane production and stable carbon isotopes

Biogenic methanogenesis is the terminal step in carbon flow in many anaerobic habitats and hence plays important role in the carbon cycle (Zinder 1993). Generally, methanogenic decomposition of organic matter requires microbial consortia of at least three interacting metabolic groups. In the first step fermentative bacteria degrades polymers to H<sub>2</sub>, CO<sub>2</sub>, formate, acetate and higher volatile fatty acids. In the second step acetogenic bacteria oxidizes the higher acids to acetate and H<sub>2</sub> or formate. Then, the mentioned substrates are utilized by methane producers, which use two main methanogenic pathways – hydrogenotrophic and acetotrophic methanogenesis (Ferry 1993):



Other pathways of methane formation exist (formate or methanol as substrates), but usually create only marginally part of the total methane production (Conrad and Claus 2005, Demirel and Scherer 2008):

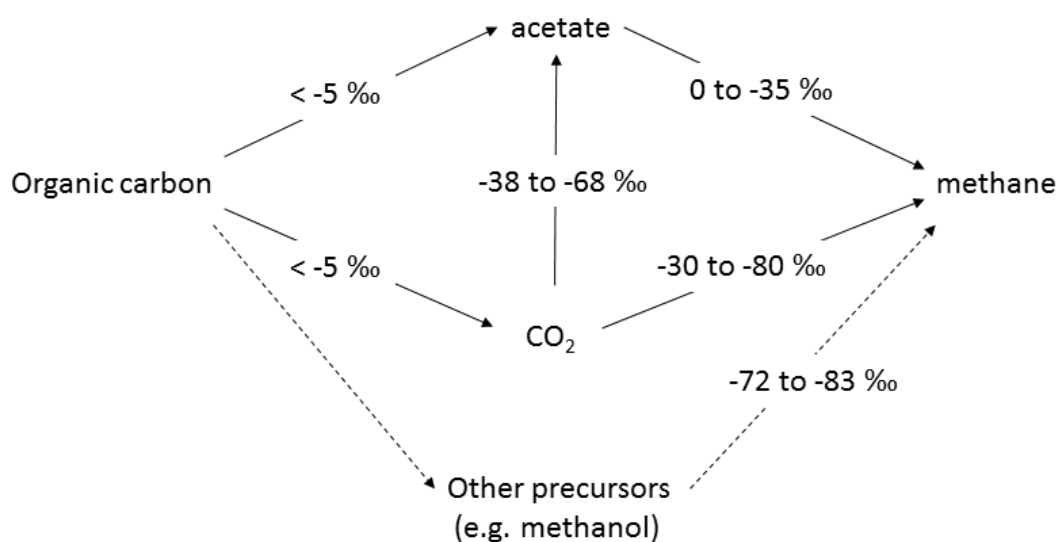


Based on the accumulation patterns of carbon dioxide and methane, the reduction process of organic matter might be further divided into three distinct phases: (1) an initial reduction phase during which most of the inorganic electron acceptors are depleted and carbon dioxide production is at its maximum, (2) a methanogenic phase during which methane production is initiated and reached its highest rate, and (3) a steady state phase with constant production rates of methane and carbon dioxide (Yao et al. 1999).

Whole process of organic matter degradation and methane formation is accompanied by isotopic fractionation. Both light and heavy carbon stable isotopes participate in chemical, biological or geochemical reactions freely but there are strong differences in the rate at which they react. The heavier isotopes (<sup>13</sup>C) react more slowly than lighter isotopes (<sup>12</sup>C) which lead to isotopic separation or fractionation between the

source and product (Fry 2006). Hence, the stable carbon isotopic composition of methane ( $\delta^{13}\text{C}\text{-CH}_4$ ) in a system is dependent on the mechanisms and rates of methane production and consumption and can thus be useful for the study of methane cycling (Whiticar 1986).

An important characteristic of biogenic methane produced in sediments is that with respect to its carbon stable isotope composition it is remarkably  $^{13}\text{C}$ -depleted ( $\delta^{13}\text{C}$  typically around -60 to -80‰; Conrad et al. 2007) compared with either allochthonous terrestrial plant detritus ( $\delta^{13}\text{C}$  value from C3 plants around -27 ‰; Peterson & Fry 1987) or phytoplankton ( $\delta^{13}\text{C}$  typically around -25 to -35 ‰; Vuorio et al. 2006). Isotopic fractionation during organic matter decomposition leading to the high  $^{13}\text{C}$  depletion of biologically produced methane can be further illustrated by Figure 1.



**Fig. 1:** Scheme of carbon flow and carbon isotope fractionation during the methane production (adapted from Conrad et al. 2014, Blaser and Conrad 2016, and using of additional literature values from: Krzycki et al. 1987, Krüger et al. 2002, Fey et al. 2004, Conrad 2005, Penger et al. 2012, Blaser et al. 2013, Ye et al. 2014).

In addition, bacterial consumption of methane is also associated with kinetic isotope effects, when enrich the residual methane in the heavier isotopes ( $^{13}\text{C}$ ). Carbon fractionation factors related to methane oxidation are generally less than 10 ‰ (Whiticar 1999, Bastviken et al. 2002), but methane oxidation enrichment factor from

18.6 to 21.4 ‰ was observed in boreal reservoirs (Venkiteswaran and Schiff 2005). Methanotrophs therefore appear to be less selective between the lighter and heavier carbon isotopes than the methanogens (Whiticar 1999).

### 1.3.2. Acetotrophic vs. hydrogenotrophic methanogenesis

Methanogens, which are able to convert acetate to methane, belong to genus *Methanosaeta* and *Methanosarcina*. Hydrogenotrophic methanogens belong to orders Methanobacteriales, Methanococcales, Methanomicrobiales and several genera of the Methanosarcinales (Zinder 1993, Demirel and Scherer 2008). Most of these methanogens are commonly found in river sediments (Brablcová et al. 2013, Buriánková et al. 2013) except a Methanococcales order, which was isolated essentially from marine and coastal environments (Garcia et al. 2000). Nevertheless, the composition of methanogenic community does not have to reflect the contribution of individual methanogenic pathways, as well as abundance of the methanogens does not reflect the methanogenic potential of the sediments (Conrad et al. 2011, Chaudhary et al. 2017).

The  $\delta^{13}\text{C}$  of methane produced only by acetotrophic methanogenesis vary between -27 and -60 ‰, while the  $\delta^{13}\text{C}$  of hydrogenotrophically produced methane vary between -45 and -90 ‰ (Conrad 2005). The difference between the two pathways of methane formation leading to the diverse  $\delta^{13}\text{C}$  values is given by diverse strength of fractionation during different metabolic pathways. The data clearly show that the fractionation factors found for  $\text{H}_2/\text{CO}_2$ -dependent methanogenesis ( $\alpha_{\text{mc}}$ ) are always significantly larger than the fractionation factors found for acetate dependent methanogenesis ( $\alpha_{\text{ma}}$ ), while the  $\alpha_{\text{ma}}$  range between 1.000 and 1.021 and  $\alpha_{\text{mc}}$  measured in various cultures of  $\text{CO}_2$ -reducing methanogens range between 1.031 and 1.077 (Krüger et al. 2002, Fey et al. 2004, Conrad 2005). While the fractionation factors can differ according to environmental conditions (e.g. substrate concentration, different microbial communities), both resulted values additionally depend on  $\delta^{13}\text{C}$  of organic matter or related processes during its degradation, which affect the  $\delta^{13}\text{C}$  of the carbon substrate ( $\text{CO}_2$  or acetate). Both pathways contribute to methane pool, while resulted  $\delta^{13}\text{C}$  of methane in the environment is given by different portion of these pathways,  $\delta^{13}\text{C}$  of methane precursors, eventually by associated processes as oxidation or diffusion. Recent findings regarding the isotopic analysis of methane in various freshwater ecosystems are summarized in Table 1.

**Tab. 1:** Overview of literature values regarding the isotopic analysis of methane in sediments of inland waters

Site	Ecosystem type	$\delta^{13}\text{C}$ of $\text{CH}_4$ (‰)	$f_{\text{mc}}$ (%)*	$\text{CH}_4$ production potential ( $\text{nmol h}^{-1} \text{gDW}^{-1}$ )	$\delta^{13}\text{C}_{\text{org}}$ (‰)	Reference
inflows of Lake Biwa, Japan	river	-64 to -47	n.s.	n.s.	n.s.	Murase et al. 2003
White Oak River, North Carolina	river	-70.8 to -65.2	18 to 42	n.s.	n.s.	Avery and Martens 1999
three streams in eastern Amazonia	river	-75.1 to -52.7	n.s.	n.s.	-29.7 to -22.8	Moura et al. 2008
five rivers in USA	river	-56.6 to -36..	n.s.	n.s.	n.s.	Sansone et al. 1999
Sitka Stream	river	-98.6 to -48.2	26 to 51	0-40	-26.7 to -25.8	Mach et al. 2015
Elbe River	river	-71.1 to -54.1	52 to 78	0 to 26.9	-27.2 to -0.4	Bednařik et al. submitted (this study)
Morava River	river	-63.9 to -52.5	37 to 89	0 to 83.3	-28.5 to -26.2	Bednařik et al. 2017 (this study)
River Itchen, U.K.	river	-58	33	22-80	n.s.	Shelley et al. 2015
16 tropical lakes, Brazil	lake	-94.5 to -57.7	50 to 90	$4.0 \pm 3.8$	-32.8 to -25.2	Conrad et al. 2011
Amazonian oxbow lakes, Brazilia	lake	n.s.	>50	0 to 130	-30 to -26	Conrad et al. 2014
Lake Dagow, Germany	lake	-65 to -50	35 to 60	250 to 300	$-30.1 \pm 0.05$	Conrad et al. 2009
two Amazonian clear-water lakes	lake	-70 to -55	53 to 63	45.5 and 63.3	-32	Conrad et al. 2010
Lake Constance, Germany	lake	-85 to -57	n.s.	n.s.	n.s.	Faber 1996
Lake Piaseczno, Poland	lake	-63.1 to -47.5	n.s.	n.s.	n.s.	Jedrysek 1995
Lake Biwa, Japan	lake	-80 to -61	>50	n.s.	-25.6 to -24.7	Murase and Sugimoto 2001
Würmsee, Germany	lake	-61 to - 52	30	n.s.	n.s.	Woltemate et al. 1984
paddy field, Vercelli, Italy	rice field	-80 to -50	33	n.s.	$-26.7 \pm 0.39$	Conrad et al. 2002
paddy field, Vercelli, Italy	rice field	n.s.	33	n.s.	$-26 \pm 0.3$	Fey et al. 2004
paddy field, Vercelli, Italy	rice field	-60 to -40	>50	n.s.	-26.5	Kruger et al. 2002
Uruguay	rice field	n.s.	25 to 42	25 to 35	-21.9 to -17.2	Scavino et al. 2013
swamp forest, Florida, USA	wetland	-63.2 to -37.1	n.s.	n.s.	n.s.	Happel et al. 1994
The Point Pelee Marsh, Ontario, Canada	wetland	-72.3 to -48.2	n.s.	n.s.	-25	Hornibrook et al. 1997
Mizorogaike pond, Kyoto, Japan	wetland	-76.7 to -52.8	n.s.	n.s.	n.s.	Sugimoto and Fujita 2006
Lakkasuo mire complex, Finland	peatland	n.s.	>40	14.9 to 209.7	n.s.	Galand et al. 2005
Lakkasuo mire complex, Finland	peatland	-89 to -58	>40	15 to 210	-27.4 to -26.5	Galand et al. 2010
permafrost region in Siberia (67°N)	peatland	n.s.	30	179.2	n.s.	Metje and Frenzel 2007

\*  $f_{\text{mc}}$  = part of hydrogenotrophically produced methane

n.s. = not specified

Generally, the most of the methane produced in nature originates from acetate, however, the relative amounts of methane produced from the methyl group of acetate or reduction of CO<sub>2</sub> can vary depending on the presence of other metabolic groups of anaerobes and the environment (Ferry 1993). Hydrogen should theoretically account for 33 % of total methanogenesis when carbohydrates or similar forms of organic matter are degraded (Conrad 1999). Many methanogenic environments show both much lower and much higher contributions of H<sub>2</sub> to methane production than is considered normal.

The lower contributions of hydrogenotrophic methanogenesis can be relatively easily explained by the contribution of homoacetogenesis. Homoacetogenic microorganisms oxidize H<sub>2</sub> or formate and reduce CO<sub>2</sub> to acetate. This process increases the amount of methane derived from acetate and the importance of methanogenic acetotrophs (Ferry 1993):



Despite of general consideration, that homoacetogenesis is thermodynamically unfavourable in many anaerobic environments and thus its importance is minimal, recent studies shows the significant occurrence of the homoacetogenesis in river sediments or peatland soils (Ye et al. 2014, Mach et al. 2015). Acetogenic bacteria can effectively compete with hydrogenotrophic methanogens (up to ten time faster consumption of H<sub>2</sub> than methanogens) and therefore, acetogens may play an important role in regulating acetate dynamics, methane production, and carbon cycling (Ye et al. 2014). The isotopic effect related to homoacetogenesis (acetyl-CoA pathway) is significantly stronger compared to fermentation and thus leads to depletion in δ<sup>13</sup>C of acetate-methyl relative to the soil organic carbon (Blaser et al. 2013). Due to this difference in fractionation between these two acetogenic pathways it was calculated, that homoacetogenesis can contribute up to 40 % of the acetate production in river sediments (Mach et al. 2015).

The mechanisms behind higher contributions of hydrogenotrophically produced methane are mostly unclear despite the fact that dominance of hydrogenotrophic methanogenesis is not unusual for freshwater ecosystems (Krüger et al 2002, Galand et al. 2010, Conrad et al. 2010, Conrad et al. 2014). In methanogenic environments H<sub>2</sub> is rapidly turned over, because its concentration is given by the simultaneous production by fermenting plus syntrophic bacteria and consumption by methanogenic archaea

(Conrad 2002). Conceivable explanations for dominance of hydrogenotrophic methanogenesis in freshwater sediments have been defined by Conrad et al. (2009) and are further discussed for example in Bednařík et al. 2017 (Paper II). Briefly, the most plausible explanation based on several studies is that organic matter in the sediments is incompletely degraded with the preferential production of H<sub>2</sub> and the accumulation of residual organic substances having a higher oxidation state than the original organic carbon (Conrad et al. 2011). End of the reduction process of organic matter characterised by steady state phase with constant production rates of methane and carbon dioxide can result in different rate of methane and carbon dioxide production, when for example oxidation status of soil organic matter changes during decomposition (Yao and Conrad 2000). Different path of organic matter degradation with use of organic compounds as oxidants (thus CO<sub>2</sub> rather than methane is the major degradation product) leading to prevalence of hydrogenotrophic methanogenesis was documented in the oligotrophic environments by Galand et al. (2010).

Moreover, changes in primary productivity, temperature, and hydrology may affect methanogenesis by altering the short-term supply of labile organic substrates to fermentative bacteria in the shallow subsurface. The predominance of methanogenic pathway appears to be determined primarily by the availability of labile substrates and hence the degree of decomposition of organic debris. This may result in prevailing acetoclastic methanogenesis in shallow subsurface, while the methane in deeper and older sediments layers is produced mainly hydrogenotrophically (Hornibrook et al. 1997, Conrad et al. 2009). However, distribution of organic matter in running waters is dependent on many factors such as stream velocity or river bed topography and do not show such a degree of stratification (Malard et al. 2002).

Temperature was recognized as an important factor effecting the stable isotopic composition of produced methane, while the methane was enriched in <sup>13</sup>C during the warmer months (Avery and Martens 1999). However, the temperature does not affect the rate of individual fractionation factors during methanogenesis, but increasing temperature was found to induce a change in the microbial community composition or the local substrate concentration and consequently in a change of the methanogenic pathway (Penger et al 2014). The methane produced from acetate reached 85 %, 67 % and 0 % of the total methane production at 10 °C, 20 °C and 50 °C, respectively (Conrad 2002). So, the methanogens are dominated by different species at different temperatures and the different population structures obviously affect the pathway and/or

rate of methane formation. However, methane production is not only affected by the direct methane producers themselves, but also by other microbial populations that influence the availability of methanogenic substrates. Hence, the change in community and pathway induced by temperature must not be mistaken as a direct temperature effect on the fractionation of stable carbon isotopes (Penger et al. 2014).

### 1.3.3. Experimental determination of particular methanogenic pathways contribution to total methane production

The relative contribution of the two main methanogenic pathways to total methane production can be calculated due to the sufficient difference in isotopic fractionation during both the hydrogenotrophic and acetoclastic methanogenesis. Rough estimation derived only from  $\delta^{13}\text{C}$  of methane and  $\delta^{13}\text{C}$  of  $\text{CO}_2$  can be calculated as apparent fractionation factor ( $\alpha_C$ ):

$$\alpha_C = \frac{\delta^{13}\text{C}-\text{CH}_4+1000}{\delta^{13}\text{C}-\text{CO}_2+1000} \quad (6)$$

Resulted values of  $\alpha_C$  higher than 1.055 are characteristic for prevailing hydrogenotrophic methanogenesis, while the values lower than 1.055 indicate prevailing acetoclastic methanogenesis (Whiticar et al. 1986, Whiticar and Faber 1986, Whiticar 1999). In addition, apparent fractionation factor is equal to fractionation factor for hydrogenotrophic methanogenesis, if the acetoclastic methanogenesis is not operating and methane is exclusively produced from  $\text{CO}_2$  reduction (Conrad et al. 2009).

It is possible to obtain more accurate (in order of percentages) contributions of individual methanogenic pathways to total methane production, if the further stable isotopic signatures and isotopic fractionation factors are known for the examined environmental system:  $\delta^{13}\text{C}-\text{CH}_4$ ,  $\delta^{13}\text{C}-\text{CO}_2$ ,  $\delta^{13}\text{C}$  of acetate-methyl, fractionation factor for the reduction of  $\text{CO}_2$  ( $\alpha_{mc}$ ) to methane and acetate-methyl to methane ( $\alpha_{ma}$ ) (Conrad 2005). However, experimental determination of fractionation factors in environmental samples is difficult, since either the hydrogenotrophic or the acetoclastic methanogenic pathway must be suppressed in order to determine the isotope fractionation by one of the two pathways specifically. For that reason, it was important milestone, when the methyl fluoride ( $\text{CH}_3\text{F}$ ) was reported to be a specific inhibitor of acetoclastic

methanogens, while hydrogenotrophic methanogens were not affected (Janssen and Frenzel 1997) and thus allowing the determination of  $\delta^{13}\text{C}\text{-CH}_4$  specifically produced from  $\text{CO}_2$  reduction ( $\delta_{\text{mc}}$ ) and of fractionation factor for hydrogenotrophic methanogenesis ( $\alpha_{\text{mc}}$ ). Acetate, which is then no longer consumed in inhibited samples, accumulates and allows determination of the isotopic signatures of the fermentatively produced acetate.

Based on inhibition treatment, direct calculation of contribution of methanogenic pathways using the residual methane production (comparison of methane production in the inhibited and uninhibited sample under same conditions) seems to be the simplest way. However, it was found that small part of hydrogenotrophic methanogenesis can be inhibited during incubation with  $\text{CH}_3\text{F}$  in addition to acetoclastic methanogenesis. Extent of the  $\text{H}_2/\text{CO}_2$ -dependent methanogenesis inhibition depends on  $\text{CH}_3\text{F}$  concentration (Conrad and Klose 1999). If the hydrogenotrophic methanogenesis is inhibited a little by  $\text{CH}_3\text{F}$ , residual methane production of the samples incubated with  $\text{CH}_3\text{F}$  can be lower than expected values corresponding with part of hydrogenotrophic methanogenesis determined by isotope analysis. Thus, it is more robust to calculate contribution of hydrogenotrophic methanogenesis from the isotopic mass balance, which should not or little be affected, when part of the hydrogenotrophic methanogenesis is inhibited (Conrad et al. 2009).

In order to determine resulted contribution of particular methane production pathways to total methane production, the calculation usually has two more assumptions: (1)  $\delta^{13}\text{C}$  of acetate is similar to  $\delta^{13}\text{C}$  of organic matter and (2)  $\delta^{13}\text{C}$  of acetoclastically produced methane is similar to  $\delta^{13}\text{C}$  of acetate. Ad 1: Isotopic enrichment factors during the degradation of organic matter to acetate is much smaller than those during hydrogenotrophic and acetoclastic methanogenesis or the intramolecular difference in  $\delta^{13}\text{C}$  between the carboxyl and methyl group of acetate (Conrad et al. 2014). So, when direct measurement of  $\delta^{13}\text{C}$  of acetate is not available, it can be assumed, that fractionation during fermentation between organic carbon and acetate-methyl, from which methane is formed, is not significant (Blair et al. 1987, Sugimoto and Wada 1993, Conrad et al. 2009), while the  $\delta^{13}\text{C}$  of acetate carboxyl group is usually depleted in  $^{13}\text{C}$  (Conrad et al. 2009, Conrad et al. 2010). However, possible exchange reactions of the carboxyl group of acetate with the  $\text{CO}_2$  pool during incubation may subsequently result to significantly heavier ( $^{13}\text{C}$  enriched) acetate (DeGraaf et al. 1996). Thus, the higher  $\delta^{13}\text{C}$  of acetate than the  $\delta^{13}\text{C}$  of organic carbon



can be theoretically caused by analysis of  $\delta^{13}\text{C}$  of total acetate, which combines the  $\delta^{13}\text{C}$  of methyl and carboxyl group of acetate. On the other hand, chemolithotrophic production of acetate (homoacetogenesis) leads to significant depletion of acetate-methyl in  $^{13}\text{C}$  due to the strong isotopic effect connected with this metabolic pathway as described above (Blaser et al. 2013).

Ad 2: Preference of the certain metabolic pathway for the lighter carbon isotope ( $^{12}\text{C}$ ), fractionation, is not sufficiently expressed in the substrate limited conditions (Fry 2007). Considering the almost complete consumption of acetate during incubation of samples in steady state conditions, very low or zero fractionation for methane from acetate can be assumed (Sugimoto and Wada 1993). Based on literature, fractionation factor for formation of methane from acetate is usually considered in the range of 1.007–1.035 (Gelwicks et al. 1994, Hornibrook et al. 2000, Penning 2006, Govert and Conrad 2009).

## **2. Aims of dissertation thesis**

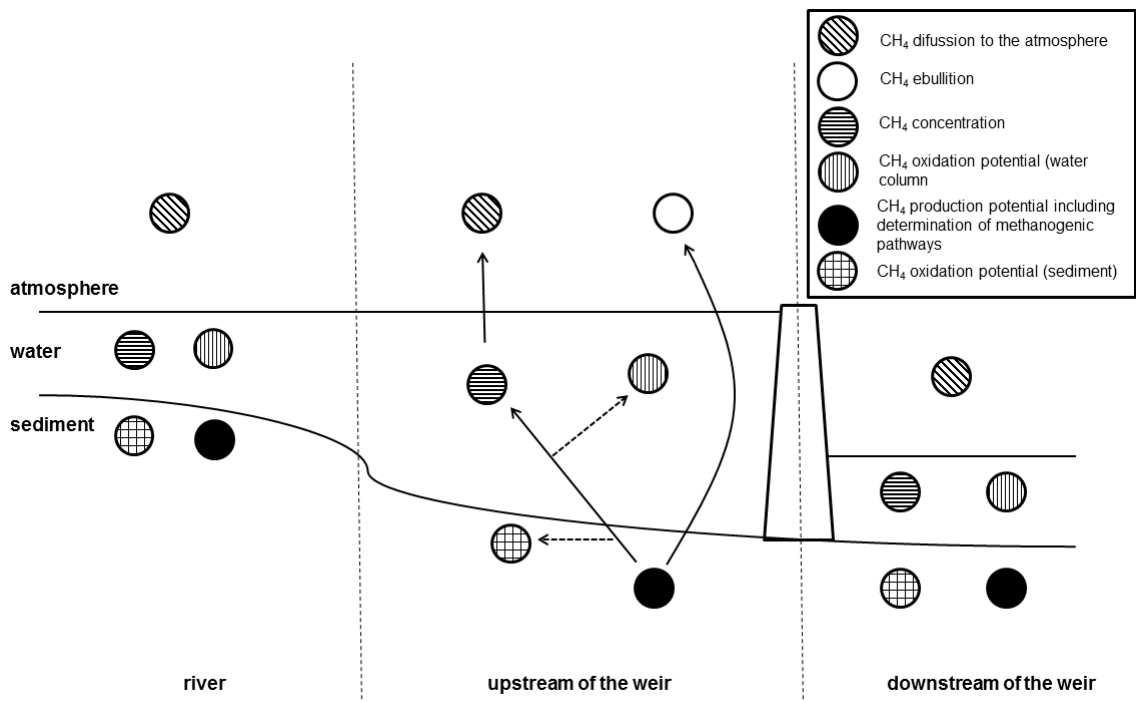
The aims of this thesis were:

- 1) To characterize the methane production and oxidation potential of the river sediments in longitudinal profile of the Elbe River and to reveal the contribution of individual methanogenic pathways to the total methane production using the stable carbon isotope analysis.
- 2) To compare the rate of the methane related processes (methane production, oxidation, emissions to the atmosphere) between weir impoundments and free flowing river sections of Morava River. Part of it was also quantification of methane ebullition and extent of methane degassing in the spillways.
- 3) To determine the proportion of methane production pathways in the sediments of the examined weir impoundments and river sections.
- 4) To characterize the spatial variability of methane production and consumption by the river sediments including the proportion of the methanogenic pathways in the cross-section profile of weir impoundment.

### 3. Material and methods

The samples collection and field measurements have taken place at the 11 locations along the Elbe River from river km 8 to river km 948 and in the cascade of three weirs in the 16 km long stretch of Morava River. More detailed characterization of studied sites is included in the Paper I and Paper II, respectively.

Determination of individual components of methane dynamic in river demands involvement of many different methods. Measured components of the methane dynamic are schematically shown in Figure 2.



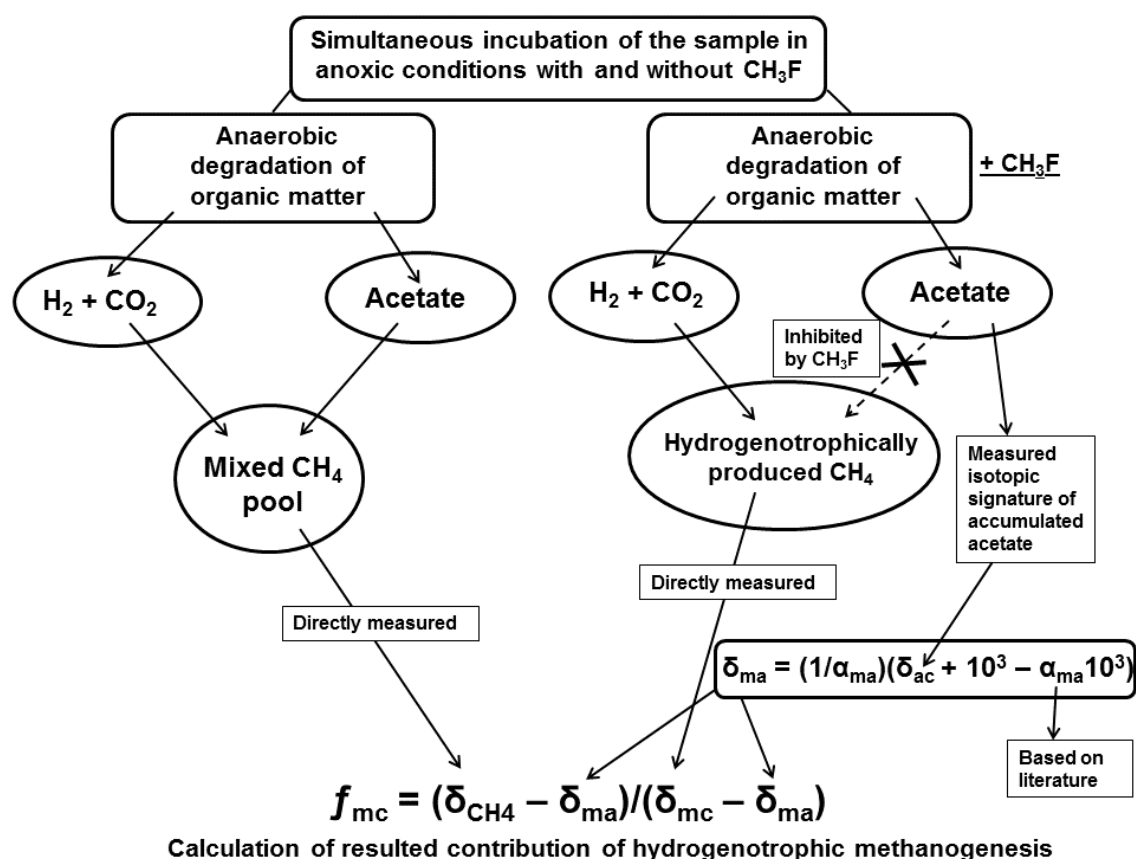
**Fig. 2:** Measured components of the methane dynamic in the impounded river

Individual methods are described in detail in the attached papers. This chapter therefore contains only an overview of the methods used for this thesis. They are briefly summarized here:

Methane production and oxidation potentials of sediments were measured during incubation experiments in the laboratory. Sediment for measurement of methane production potential was incubated in anoxic conditions (headspace of bottle was flushed with nitrogen) approximately one month. Gas samples from headspace were taken repeatedly during the course of incubation (4-6 weeks) and analysed for concentrations of methane. The rate of methane production was calculated from the

slope of the linear regression given by the graph of methane concentration increase over time. Sediment for measurement of methane oxidation potential was incubated under the oxic conditions (ambient air in a headspace) with addition of methane. Potential methane oxidation rates were obtained from the slope of the methane concentration decrease over time.

Determination of the contribution of individual methanogenic pathways to total methane production was carried out using the stable carbon isotope measurements. This method is described in more detail in Paper II and schematically illustrated in Figure 3. It is based on effect of methylfluorid ( $\text{CH}_3\text{F}$ ), which completely inhibits acetate-dependent methanogenesis. Methane was then exclusively produced by hydrogenotrophic methanogenesis and thus allowed determination of the fractionation factors specific for this methanogenic pathway.



**Fig. 3:** Scheme of the method for determination of the contribution of two main methanogenic pathways to the total methane production using analysis of the stable carbon isotopes (original Adam Bednařik)

Methane emissions from the surface water to the atmosphere were detected by three different methods fully described in Paper II. First, methane emissions across the air-water interface were directly measured by a floating chamber method. Second, methane diffusion fluxes (i.e. without contribution of the ebullition) to the atmosphere were determined using calculations derived from recent studies and based on the gas transfer velocity and the methane concentration gradient between the river water and the atmosphere (Striegl et al. 2012, McGinnis et al. 2014, Borges et al. 2015, Bodmer et al. 2016). Third, ebullition measurements were carried out using submerged gas funnel traps. Moreover, degassing at weirs was estimated on the basis of methane concentration differences and water discharge.

From comparison of first and second method for measurement of methane emissions in our conditions followed, that we did not observe the increased methane emissions caused by the additional induced turbulence arising from application of anchored chambers described by Lorke et al. (2015). Diffusive fluxes calculated from gas transfer velocity and directly measured emissions by chambers were not different in sites with the marginal contribution of the ebullition.

## **4. Main results and discussion**

### **4.1. Sediment methane dynamics along the Elbe River (Paper I)**

The first study was motivated by a paucity of data on methane dynamic in large rivers, especially those focused on methane-related processes in sediments. Therefore, we have investigated the spatial variability of methane production and consumption by sediments of Elbe River including the differentiation of the methane production pathways (acetoclastic vs. hydrogenotrophic) using the natural abundance of stable carbon isotopes. Additionally, we determined the diversity of the methanogenic communities.

The methane production was detected in six sediment samples (from 11 of total) along the Elbe River, while the methanotrophy was found in all examined sediment samples. The methane production and oxidation differed considerably in the river longitudinal profile without any clear trend and without any correlation with other studied environmental parameters. Moreover, the *mcr-A* and *pmo-A* gene copy numbers (genes showing the presence of methanogens and methanotrophs, respectively) were similar and quite stable among the sediment samples. It follows that while it was found hotspots of the measured methane processes, the molecular data showed no spatial characteristics. This observation confirms the more general pattern that the microbial abundance and community patterns only rarely correlate with their activity (Mach et al. 2015, Chaudhary et al. 2017).

On the other hand, several samples of sediments, which did not show any detectable methane production despite of incubation in anoxic conditions and presence of methanogens, were probably limited by substrate (organic carbon) or by the availability of alternative electron acceptors (dissolved  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Fe}^{3+}$ ).

Incubation experiments with isotopic analyses of  $\text{CO}_2$  and methane revealed that the hydrogenotrophic pathway of methane formation ( $\text{CO}_2$  reduction) was dominant for all examined methane productive sites accounting for 52 to 78 % of total methane release, thus implying the most probably the incomplete degradation of organic matter (see chapter 1.3.2.).

The following research was therefore more focused on detailed characterization of methane dynamic in these hot-spots of methane related processes.

#### **4.2. Effect of weir impoundments on methane dynamics in a river (Paper II)**

In order to understand the role of small weir impoundments in river methane dynamics, we measured methane concentration, methane oxidation in the water column as well as sediments, total methane emissions to the atmosphere (diffusion and ebullition) and sediment methane production in three weir impoundments and river reaches between them. Generally, reduced water velocity upstream of dams leads to higher sediment accumulation rates and burial of organic carbon, which subsequently allows development of anoxic conditions suitable for methane production (e.g. Abril et al. 2005; Maeck et al. 2013).

Indeed, we found that river methane dynamics might be highly influenced by weirs, especially by increased methane production and consumption by sediments, followed by increasing methane emissions to the atmosphere. Both methane production and oxidation potential of sediments were higher upstream of the weirs compared to downstream of the weirs or usual river reaches. The total methane emissions to the atmosphere reached the highest values upstream of the weirs, while the ebullition accounted for ~96 % of the total methane emissions. Methane consumption in the sediments together with the microbial methane oxidation in the water column substantially contributed to the methane removal from surface water. Thus, the contribution of the ebullition to the methane emissions in these shallow impoundments was enhanced by bypassing microbial methane oxidation, compared to relatively slow diffusion fluxes. Overall, methane emissions from average weir impoundment can reach up to 42 times higher values than from the river section of an equal area.

In spite of such high emission fluxes including further methane release by degassing in the spillways of the weirs and high methane oxidation, considerable 7.5 times increased of methane concentration in the surface water was observed in the 16 km long examined section, pointing to important methane sources in such a short river reach.

The contribution of  $H_2/CO_2$ -dependent methanogenesis to total produced methane tended to be higher for sediments upstream of the weirs, compared to the sediments from river sections or downstream of the weirs. More precisely, hydrogenotrophic methanogenesis contributed 37 to 89 % of the total methane production and was dominant (more than 50 %) in sediments upstream of the weirs, while acetoclastic methanogenesis was probably prevailing in remaining sediments.

Observed predominance of hydrogenotrophic methanogenesis most probably explained by incomplete degradation of organic matter in the sediments with the preferential production of H<sub>2</sub> and the accumulation of residual organic substances having a higher oxidation state than the original organic carbon (Conrad et al. 2011) is frequently reported from lake ecosystems (Murase and Sugimoto 2001, Conrad et al. 2009, Conrad et al. 2014).

These all findings together with other sediment characteristics show that methane related processes upstream of the weirs are more comparable with lake sediments than river sediments.

#### **4.3. Methane formation and consumption by sediments in the cross-channel profile of the impoundment (Paper III)**

Previous results encouraged us to have a further insight to the spatial variability of methane production within the weir impoundments. For this purpose, sediments upstream of the weir were sampled in the cross-channel profile and in two different sediment depths. Methane production and oxidation potentials including the contribution of individual methanogenic pathways to total methane production were measured as before.

Samples from the surface sediment layer (0-10 cm) reached higher methane production than sediments from the deeper layer (10-20 cm) during the incubation experiments. The methane oxidation potential of sediments showed the same spatial pattern as observed for the methane production.

We hypothesized that more uniform sedimentation in cross-channel profiles compared to river due to the overall decrease of the flow velocity upstream of the weir will result in more identical rates of methane-related processes through the transect. Instead, we found frequently observed pattern in lakes and rivers, that littoral zones are main sites of the methanogenic activity. Increased methane production in the littoral sediments of lakes is likely caused by greater availability of labile organic matter from the aquatic vegetation and by higher temperatures in summer months, which in turn support higher methane production rates (Bussmann 2005, Murase et al. 2005, Hofmann et al. 2010), while higher methane production in riparian habitats of rivers is probably connected with low water exchange between surface water and sediments, thus, the low oxygen supply to the sediment (Malard et al. 2002, Fischer et al. 2005).



In our study, sediments near the bank zones and in the mid-channel were characterised by the highest organic carbon content (6.9 %) as well the highest methanogenic activity ( $2.5 \text{ mmol g}^{-1} \text{ DW d}^{-1}$ ).

Stable carbon isotopes analysis, used for determination of individual methanogenic pathways, confirms our previous findings that the methane production is dominated by  $\text{H}_2/\text{CO}_2$  dependent methanogenesis upstream of the weir. However, it was more evident in the surface sediment layer (0-10 cm), while the proportion of acetoclastic and hydrogenotrophic methanogenesis in deeper sediment layer (10-20 cm) was more balanced. This slight shift in the contribution of methanogenic pathways was not caused by the lability or availability of organic substrate because the similarity of these parameters between examined sediment layers, so real causes stayed unclear based on available data.

## 5. Conclusions

Altogether results presented in this study confirm that river sediments are important place of anaerobic degradation of organic matter with methane as a final product. Moreover, the methane dynamics in a river system show a high local variability, indicating that multiple measurements are needed to characterize the sources and fates of the methane. The methane dynamics along the river continuum is dramatically impacted by the building of small impoundments, which contribute significantly to the total methane production and its subsequent emission disproportionately to their area. Therefore, sampling carried out regardless of occurrence of small impoundments can considerably underestimate methane emissions to atmosphere from lotic ecosystems and the importance of these barriers for river methane dynamic. We found out that the most productive sites in the impounded river zones are littoral sediments as was previously reported for different freshwater habitats including lakes and rivers. Modifications of methane related processes in impounded river zones are reflected also by different contribution of individual metabolic pathways of methane production compared to usual river sections.

Recent studies clearly show that rivers may emit considerable amount of methane to the atmosphere. Hence, rivers should be included into the future estimations and models of a global methane budget. Namely, high spatial variability of methane related processes is remarkable in rivers compared to other ecosystems and deserves to be considered in greater details in the further studies. In any case, it is necessary to consider the sampling location of particular water habitats, because sediment samples taken only in the particular zones of water habitats can significantly misrepresent the further extrapolation of obtained results. Considering the considerable sediment methane production upstream of the weirs, more studies focusing on quantification of direct methane fluxes from sediments to surface water and conducted in-situ would also be of great importance.

Finally, I do hope that findings presented here contribute to improving our knowledge about the sources and fate of methane in river ecosystems.

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## **7. Attached publications**

### **7.1. Sediment methane dynamics along the Elbe River**

**Bednařík A**, Blaser M, Matoušů A, Tušer M, Chaudhary PP, Šimek K, Rulík M (2019):  
Sediment methane dynamics along the Elbe River. *Limnologica (in review)*

1 **Sediment methane dynamics along the Elbe River**

2

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20

21 **Abstract**

22 Methane (CH<sub>4</sub>) is an important atmospheric trace gas mostly released from wet anoxic soils  
23 and sediments. While many studies have focused on relatively homogenous environments like  
24 rice fields and lake sediments, the changing contribution of heterogeneous sediments e.g.  
25 along the longitudinal profile of a rivers has not been covered very frequently. Here we  
26 investigated sediment samples from 11 locations of the Elbe River. Sediments were incubated  
27 to measure methanogenic/methanotrophic potentials and contribution along individual  
28 methanogenic pathways using isotope analysis of δ<sup>13</sup>C. Additionally, we determined the  
29 diversity of the methanogenic communities (analysis of T-RFLP targeting the mcr-A gene in  
30 the sediment samples), while abundances of archaea, methanogens and methanotrophs were  
31 determined by qPCR. The CH<sub>4</sub> production was detected in six samples (out of 11 examined)  
32 and ranged from 0.12 to 644.72 nmol gDW<sup>-1</sup> d<sup>-1</sup>. Methanotrophy was found in all examined  
33 sediment samples and ranged from 654 to 10,875 nmol gDW<sup>-1</sup> d<sup>-1</sup>. Abundance of  
34 methanogens and methanotrophs (Mcr-A and pmo-A gene copy numbers) was not  
35 significantly different and quite stable around 10<sup>6</sup> to 10<sup>7</sup> copies gDW<sup>-1</sup>. The group specific  
36 qPCR showed high fluctuations, while the highest counts were reported for  
37 *Methanomicrobiales* and *Methanosarcinales* (10<sup>5</sup> to 10<sup>8</sup> copies per gram dry sediment),  
38 followed by *Methanobacteriales* (10<sup>3</sup> to 10<sup>5</sup> copies per gram dry sediment). A significant  
39 proportion of unidentified methanogens was found in almost every locality. Isotope analysis  
40 of δ<sup>13</sup>C showed that (CH<sub>4</sub>) is produced mainly by hydrogenotrophic methanogens. We see no  
41 trend in the studied parameters along the Elbe River. The molecular data showed no spatial  
42 characteristics, while we found hotspots of the measured CH<sub>4</sub> processes (CH<sub>4</sub> production and  
43 oxidation) due to other local driving factors (e.g. carbon content). Thus, our results indicate  
44 that the observed variability of the CH<sub>4</sub> production and oxidation rates is only indirectly  
45 linked to the presence or quantities of different microbial guilds.

46

47 **Key words:** methane, river sediments, methanogenic potential, methanotrophic potential,  
48 qPCR, TRFL-P, stable carbon isotope, delta 13C, mcr-A, pmo-A,

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## 50        **1. Introduction**

51 Methane (CH<sub>4</sub>) is a significant component of the aquatic carbon cycling and is involved in  
52 many biogeochemical and physical processes. Since biological methane production is mainly  
53 linked to wet anoxic soils and sediments, streams and rivers are one of many sources of  
54 atmospheric methane contributing 15 – 40 % to the total CH<sub>4</sub> efflux of wetlands and lakes  
55 (Stanley et al. 2016). Sediments are very important sites of riverine metabolism including  
56 their role in methanogenesis (Dahm et al. 1987). Generally, the mineralization of the organic  
57 matter under anaerobic conditions is carried out by several microbial organisms and results -  
58 in the absence of other electron acceptors like nitrate, iron, manganese etc. - in the release of  
59 CH<sub>4</sub> and CO<sub>2</sub> (Zeikus 1983, Schink 1997). Two major feeding habits of methanogens can be  
60 differentiated: acetoclastic (acetate conversion to CH<sub>4</sub> and CO<sub>2</sub>) and hydrogenotrophic (H<sub>2</sub>  
61 and CO<sub>2</sub> to CH<sub>4</sub> and water). These two pathways can be discriminated using isotopic  
62 techniques due to diverse strength of isotopic fractionation during different methanogenic  
63 pathways, which lead to different isotopic composition of resulted CH<sub>4</sub> (Conrad 2005).

64 The CH<sub>4</sub>, which is formed in the sediments is subsequently released via diffusion, ebullition  
65 or through plants to the surface water or the atmosphere, where it is transported via advection  
66 or dispersion, respectively. Simultaneously, the CH<sub>4</sub> is subject of significant oxidation by CH<sub>4</sub>  
67 oxidizing bacteria during its transport in aquatic ecosystems. Moreover, all the processes  
68 involved in the aquatic CH<sub>4</sub> cycle are subject to large temporal and particularly spatial  
69 heterogeneity (Stanley et al. 2016). Understanding the variability of methane-related  
70 processes is key factor leading to more precise estimates of lotic ecosystems relevance in the  
71 global methane budget, which is recently based on scarce data (Bastviken et al. 2011).

72 Previous studies conducted in large rivers show, that rivers are mostly oversaturated in  
73 dissolved CH<sub>4</sub> with respect to the atmosphere equilibrium (i.e. rivers are a net source of CH<sub>4</sub>  
74 to the atmosphere). Frequently observed inverse relationship between discharge and CH<sub>4</sub>  
75 concentration is most probably given either by dilution (Koné et al. 2010, Anthony et al.  
76 2012) or by higher temperature during low water periods. Increased temperature further  
77 enhances microbial activity and thus decreases oxygen levels (Borges et al. 2018). Notably,  
78 CH<sub>4</sub> emissions from rivers may reflect the properties of the surrounding catchments, such as  
79 topography, soil type and texture, land use, hydrological connectivity with wetlands and other  
80 anthropogenic activities as input of wastewaters (Jones and Mulholland 1998, Silvennoinen  
81 2008, Yang et al. 2012, Borges et al. 2015). Generally, the studies considering CH<sub>4</sub> in large  
82 rivers were focused mainly on its concentration in surface water and its eventual flux to the  
83 atmosphere, but the data concerning the sediment related processes are missing (Teodoru et

84 al. 2015, Barbosa et al. 2016). Hence only few data related to CH<sub>4</sub> processes in sediments of  
85 large rivers exists and almost no data comes from complex longitudinal studies, despite of fact  
86 that river sediments have great potential as source of CH<sub>4</sub> due to high methanogenic biomass  
87 (Buriánková et al. 2012).

88 Many studies examining CH<sub>4</sub> production in stream and rivers confirm that methanogens are  
89 ubiquitous members of the microbial community within river hyporheic sediments (e.g.  
90 Sanders et al. 2007, Trimmer et al. 2012, Chaudhary et al. 2017). Currently there are seven  
91 orders of methanogenic archaea described in the literature: *Methanomicrobiales*,  
92 *Methanosarcinales*, *Methanocellales*, *Methanobacteriales*, *Methanococcales*, *Methanopyrales*  
93 and *Methanomassiliicoccales* (Borrel et al. 2011, Borrel et al. 2013, Borrel et al. 2014, Lang  
94 et al. 2015). *Methanomicrobiales* and the *Methanosarcinales* followed by *Methanobacteriales*  
95 dominate the methanogenic communities in freshwater sediments of lakes and rivers (Chan et  
96 al. 2005, Chaudhary et al. 2013). Moreover, *Methanocellales* are common in rice field soils or  
97 peats and have rarely been found in lake sediment (Scavino et al. 2003, Galand et al. 2005,  
98 Conrad et al. 2010). Our previous studies conducted in another European river (Sitka, Czech  
99 republic) revealed three major methanogenic groups using molecular techniques (denaturing  
100 gradient gel electrophoresis, terminal restriction fragment length polymorphism [T-RFLP],  
101 quantitative polymerase chain reaction [qPCR] and cloning): *Methanosarcinales*,  
102 *Methanomicrobiales* and *Methanobacteriales* (Buriánková et al. 2013, Brablcova et al. 2014,  
103 Chaudhary et al. 2014, Chaudhary et al. 2017). Hence we focused our attempts to clarify the  
104 role of these groups using T-RFLP and qPCR in the present study.

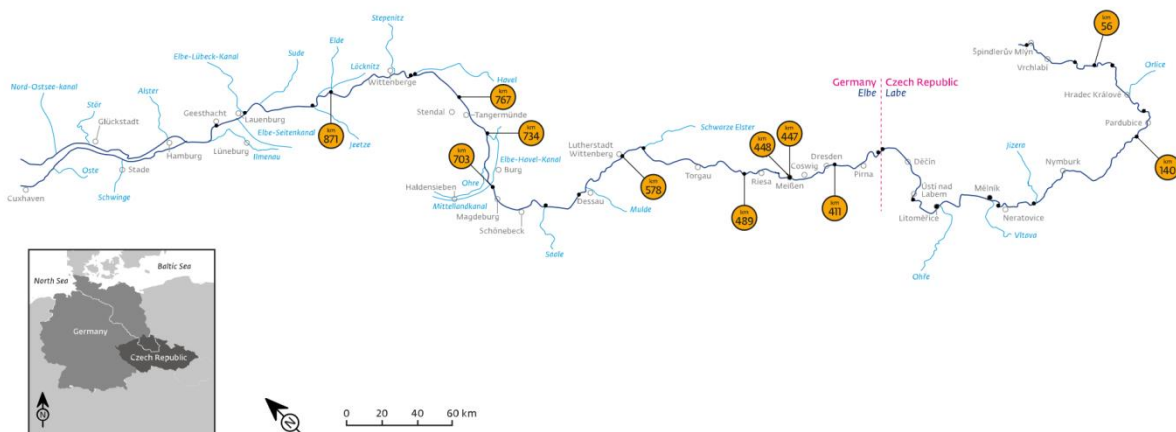
105 In principle, one can raise four hypotheses to describe the turnover of organic matter in river  
106 sediments along the longitudinal profile of a river: either (1) decrease or (2) increase of the  
107 CH<sub>4</sub> related processes along the river flow; further (3) no correlation with environmental  
108 factors but hotspots of the microbial activities due to other local factors (e.g. carbon content  
109 etc.), and (4) no obvious impact resulting in comparable process rates along the riverbed. To  
110 validate which of these hypothesis may be applied for river systems, our aim was to describe  
111 the following processes and elucidate how our results could support the above-mentioned  
112 hypothesis: (i) methanogenic and methanotrophic potential of the sediments, (ii) an isotopic  
113 signal of CH<sub>4</sub> including determination of methanogenic pathways to the total CH<sub>4</sub> production,  
114 (iii) the community composition (TRFLP) and quantification (qPCR) of archaea,  
115 methanogens and methanotrophs in the sediment samples. Samples for this study were taken  
116 during a large sampling campaign along the Elbe River carried out in October 2013, from  
117 Špindlerův Mlýn (km 8) to Geesthacht (km 948) (more detailed in Matoušů et al. 2018).



118 **Material and methods**

119 **1.1. Study site**

120 The Elbe River rises at an elevation of 1,386 m above sea level in the Krkonoše (Giant  
121 Mountains) in the northeast of the Czech Republic, flowing through the central part of the  
122 Czech Republic and through central and northern Germany before discharging into the North  
123 Sea at Cuxhaven, 110 km northwest from Hamburg. Its total length is 1,094 km and its  
124 catchment area is 148,268 km<sup>2</sup>. Sediment samples for this study were taken at 11 different  
125 sites along the river flow in October 2013. Localization of each sampling sites including  
126 sediment characteristics are specified in Fig. 1 and Table 1. (Note that we limited the  
127 sampling to the freshwater regions of the Elbe River; samples below the weir in Geesthacht  
128 are at least partially influenced by the North Sea.)



129  
130 **Fig. 1:** The sampling sites on the longitudinal profile of the Elbe River. The numbering is  
131 determined by the distance from the river source in km (see “River km” in Tab. 1). Original  
132 map: [https://commons.wikimedia.org/wiki/File:Lauf\\_der\\_Elbe.png](https://commons.wikimedia.org/wiki/File:Lauf_der_Elbe.png); modified.

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135 **1.2. Sampling of sediment**

136 Triplicates samples of the upper sediment (down to a depth of 10 cm) were collected by hand  
137 shovel near the shore. A bulk sediment was used for a granulometric analysis, while sediment  
138 intended for incubation experiments were sieved through a 1-mm sieve immediately after the  
139 sampling to remove coarse detritus, stones or invertebrates. For the molecular analysis, frost-

140 resistant vials containing 5 g of fresh sediment were put into a liquid nitrogen storage box,  
141 sediment samples for nutrient analyses, organic carbon content, methanogenic potential,  
142 isotope composition and methanotrophic activity were transferred into 50 mL Falcon tubes  
143 and stored in a cool box until further processing.

144 Sediments for the granulometric analysis were sieved through a system of ten sieves with  
145 decreasing mesh sizes. All separate parts of the sediment were weighted and grain median  
146 size was analysed using the Gradistat software (version 8.0) (Blott and Pye 2001). The dry  
147 weight of the sample was determined gravimetrically. The carbon and nitrogen contents of the  
148 sediments were quantified on a CHNS-element analyzer by the Analytical Chemical  
149 Laboratory of the University of Marburg.

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### 151 **1.3. Incubation experiments**

152 For the investigation of methanogenic pathways and methane production potential about 40 g  
153 (wet weight) of the sediment were transferred in triplicate into 60 ml sterile serum bottles,  
154 flushed with N<sub>2</sub>, closed with butyl rubber stoppers and incubated at 25 °C in the dark. At the  
155 start of the incubation 5 ml of distilled autoclaved water were added for later sampling of the  
156 liquid phase. The gas headspace of half of the bottles was supplemented with 3% CH<sub>3</sub>F (v/v)  
157 to specifically inhibit acetotrophic methanogenesis (Janssen and Frenzel 1997). The gas  
158 samples were taken repeatedly (twice a week) during the course of incubation (4–6 weeks)  
159 and analysed for concentrations of CH<sub>4</sub>, carbon dioxide (CO<sub>2</sub>) and δ<sup>13</sup>C of CH<sub>4</sub> and CO<sub>2</sub>. At  
160 the end of the incubation, the bottles were sacrificed to determine concentration and δ<sup>13</sup>C of  
161 acetate.

162 Methane oxidation potential of sediments was determined in triplicate for each sample. Sterile  
163 bottles (250 ml) were filled with 20 g of sediment (wet weight) and closed by a cap with  
164 PTFE silicone septa. The headspace (ambient air) was enriched with CH<sub>4</sub> to give a final  
165 concentration of 10,000 ppm and incubated at 25 °C in the dark. The concentration of CH<sub>4</sub> in  
166 the headspace of each bottle was measured at T<sub>0</sub> h and then nine times during 170 h. The CH<sub>4</sub>  
167 production and oxidation potentials were calculated from the linear slope of CH<sub>4</sub>  
168 concentration change over time.

169 In the sediment incubation experiments, CH<sub>4</sub> was analyzed by gas chromatography (GC)  
170 using a flame ionization detector (Shimadzu, Kyoto, Japan). CO<sub>2</sub> was analyzed in the same  
171 instrument after conversion to CH<sub>4</sub> with a methanizer (Ni-catalyst at 350°C, Chrompack,  
172 Middelburg, Netherlands).

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#### 1.4. Isotopic analyses

Isotope measurements of  $^{13}\text{C}/^{12}\text{C}$  in gas samples were performed on a gas chromatograph combustion isotope ratio mass spectrometer (GC-C-IRMS) system (Thermo Fisher Scientific, Bremen, Germany). The precision of repeated analysis was  $\pm 0.2\%$  when 1.3 nmol of  $\text{CH}_4$  was injected. The principle operation was described by Brand (1996) with details given in several recent publications (Blaser et al. 2013; Penger et al. 2012; Penger et al. 2014). An isotopic analysis and quantification of acetate were performed on a high pressure liquid chromatography (HPLC) system (Spectra System P1000, Thermo Fisher Scientific, San Jose, CA, USA; Mistral, Spark, Emmen, the Netherlands) equipped with an ion-exclusion column (Aminex HPX-87-H, BioRad, München, Germany) and coupled to Finnigan LC IsoLink (Thermo Fisher Scientific, Bremen, Germany) as described by Krummen et al. (2004). Isotope ratios were detected on an IRMS (Finnigan MAT Deltaplus Advantage). The HPLC-C-IRMS system had a detection limit of about 5  $\mu\text{M}$  and a precision of  $\pm 0.3\%$ . Details on acetate determination and calculations of isotope fractionation factors and the contribution of hydrogenotrophic methanogenesis have been described in Blaser et al. (2013) or Penger et al. (2014) and it is summarized in the supplementary material (text and Fig. S1).

#### 1.5. Molecular analyses

DNA was extracted from the fresh sediment before the start of the incubation using the PowerSoil DNA Isolation Kit (MO BIO, USA), according to the manufacturer's instructions. The extracted DNA was used to characterize the *mcr-A* gene by T-RFLP (Terminal-restriction length polymorphism) according to Chin et al. (Chin et al., 1999; Liu et al., 1997) using the primers *mcr-A* f (TAY GAY CAR ATH TGG YT) and *mcr-A* r (ACR TTC ATN GCR TAR TT) published by Springer et al. (Springer et al., 1995) with a FAM (6-carboxyfluorescein)-label at the forward primer. The *mcr-A* gene amplicons were digested with Sau96I (Fermentas), and the products were size-separated in an ABI 3130 DNA sequencer (Applied Biosystems, Darmstadt, Germany). The normalization and standardization of the T-RFLP profiles was performed according the method from Dunbar et al (2001). To assign the resulting fragments we used published literature values (Chin et al. 2004; Conrad et al. 2008; Kemnitz et al. 2004; Lueders et al. 2001; Ramakrishnan et al. 2001; Mach et al. 2015) as well as a clone library, which was constructed in our lab in order to thoroughly characterize the

207 methanogenic community at different locations and depth of Sitka stream (Chaudhary et al.  
208 2017).

209

### 210 **1.6. Quantitative polymerase chain reaction (qPCR) in sediment samples**

211 In order to quantify the microbial community we used a set of different primers targeting the  
212 total archaea (16S *rRNA* genes), methanogenic archaea (*mcrA* gene), three major  
213 methanogenic orders *Methanobacteriales* (MBT-set), *Methanomicrobiales* (MMB-set), or  
214 *Methanosarcinales* (MSL-set), and methanotrophs (*pmoA* gene) (Ovreas *et al.*, 1997, Luton  
215 *et al.*, 2002, Yu *et al.*, 2005) (Table S1). qPCR was performed using the BioRad CFX  
216 Connect™ qPCR Detection System (BioRad, USA). The 25µL real-time PCR mixture was  
217 prepared using the Brilliant II SYBR master mix (Agilent Technologies, USA), 12.5 µL of 2x  
218 reaction solution, 0.25 µL of each primer (final concentration 0.25 µM), 5 µL of template  
219 DNA, and 7 µL of PCR-grade water. The two-step amplification protocol was applied as  
220 follows: initial denaturation for 5 min at 94 °C followed by 45 cycles of 30 s at 94 °C and  
221 combined annealing and extension for 30 s at X°C (X values are given in Table S1). The  
222 fluorescent signal was measured at the end of each annealing/extension step. DNA samples  
223 were analyzed in triplicate at each point. Standard curves were generated for the  
224 methanogenic strains, by amplifying the target genes with PCR. The PCR products were  
225 cloned into the pGEM-T Easy vector (Promega, Madison, WI). The plasmids were extracted,  
226 serially diluted, and used as templates in qPCR.

227

### 228 **1.7. Statistical analysis**

229 Data analyses were performed by using the STATISTICA 12 software (StatSoft 2013).  
230 Shapiro-Wilk test was used to test the normal distribution of a data. The Spearman's  
231 correlation analysis of data were used to find the relationships among environmental  
232 parameters as independent variables (e.g. carbon content, grain median size) and  
233 experimentally measured parameters as dependent variables (e.g. CH<sub>4</sub> production and  
234 oxidation potential). All statistical tests used a significance level of 5 %.

235

236 **2. Results**

237 **2.1. Methane production and oxidation by sediment**

238 The CH<sub>4</sub> production in top sediments (0-10 cm) was recorded only for six sites (out of 11  
239 examined): Valy (km 140), Meissen – river (km 447), Meissen – harbour (km 448),  
240 Muehlberg (km 489), Hohenwarthe (km 703) and Dömitz (km 871) (Fig. 2). Methanogenic  
241 potential of these sediments ranged from 0.12 to 644.72 nmol gDW<sup>-1</sup> d<sup>-1</sup> with the highest CH<sub>4</sub>  
242 production in Meissen - harbour (mean 551.68 ± 46.68 nmol gDW<sup>-1</sup> d<sup>-1</sup>). The methanogenic  
243 potential was positively correlated with the carbon content of the sediment (r = 0.64, p <  
244 0.05).

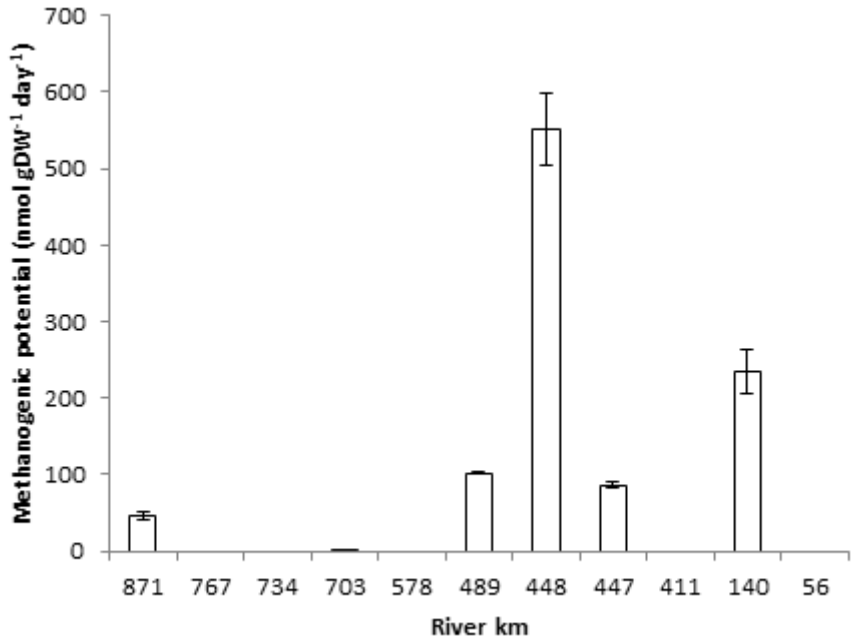
245 Isotopic analyses of CO<sub>2</sub> and CH<sub>4</sub> were used to calculate the contribution of different  
246 methanogenic pathways (Fig. 3). These analyses were performed for CH<sub>4</sub> productive  
247 sediments except the site Hohenwarthe (km 703), where the formation of CH<sub>4</sub> was  
248 insufficient for isotopic analyses during the incubation (compare Fig. 2). The  
249 hydrogenotrophic pathway of CH<sub>4</sub> formation was dominant during the whole incubation for  
250 all five sites accounting for 52 to 78 % of total CH<sub>4</sub> release. Detailed sediment characteristics  
251 are provided in Table 1.

252 **Table 1.** Basic study sites description with sediment characterization (mean values  $\pm$  SE; n.m. = not measured).

River km	Sampling site	Grain median size (mm)	Water content (%)	Sediment carbon content (%)	Sediment nitrogen content (%)	$\delta^{13}\text{C}$ of sediment carbon (‰)	$\delta^{13}\text{C}$ of acetate (‰) <sup>a</sup>	Acetate conc. (mM) <sup>a</sup>
56	Verdek	19.3	18.9 $\pm$ 1.2	0.19 $\pm$ 0.01	0.01 $\pm$ 0.0	79.8 $\pm$ 60.2 <sup>b</sup>	n.m.	n.m.
140	Valy	0.20	31.2 $\pm$ 1.4	1.70 $\pm$ 0.12	0.13 $\pm$ 0.0	-25.6 $\pm$ 0.4	-14.5 $\pm$ 0.1	1.3 $\pm$ 0.7
411	Wachwitz	17.1	19.5 $\pm$ 0.6	0.25 $\pm$ 0.01	0.01 $\pm$ 0.0	-16.7 $\pm$ 2.7	n.m.	n.m.
447	Meissen	9.51	17.3 $\pm$ 0.6	0.42 $\pm$ 0.03	0.03 $\pm$ 0.0	-23.2 $\pm$ 0.9	-26.3 $\pm$ 0.4	3.9 $\pm$ 0.6
448	Meissen - harbor	0.32	68.9 $\pm$ 2.6	5.70 $\pm$ 0.34	0.46 $\pm$ 0.1	-27.2 $\pm$ 0.4	-17.8 $\pm$ 1.1	0.4 $\pm$ 0.1
489	Muehlberg	11.4	16.3 $\pm$ 0.4	0.12 $\pm$ 0.01	0.01 $\pm$ 0.0	-0.4 $\pm$ 5.7	-18.4 $\pm$ 3.8	3.8 $\pm$ 2.2
578	Lutherstadt Wittenberg	0.41	18.5 $\pm$ 0.9	0.06 $\pm$ 0.01	0.00 $\pm$ 0.0	-23.6 $\pm$ 1.0	n.m.	n.m.
703	Hohenwarthe	0.49	16.2 $\pm$ 0.4	0.23 $\pm$ 0.01	0.01 $\pm$ 0.0	-20.8 $\pm$ 2.1	n.m.	n.m.
734	Bittkau	0.56	16.1 $\pm$ 0.6	0.17 $\pm$ 0.01	0.01 $\pm$ 0.0	27.8 $\pm$ 26.8 <sup>b</sup>	n.m.	n.m.
767	Arneburg	7.49	14.2 $\pm$ 0.6	0.03 $\pm$ 0.01	0.00 $\pm$ 0.0	-10.2 $\pm$ 8.3	n.m.	n.m.
871	Domitz	0.91	14.4 $\pm$ 0.8	0.08 $\pm$ 0.03	0.00 $\pm$ 0.0	-11.6 $\pm$ 0.0	-27.2 $\pm$ 1.8	4.0 $\pm$ 1.1

253 <sup>a</sup> An acetate concentration and  $\delta^{13}\text{C}$  of acetate were measured after the incubation in samples inhibited by  $\text{CH}_3\text{F}$

254 <sup>b</sup> Observed positive values of  $\delta^{13}\text{C}$  of sediment carbon are connected with very depleted carbon pool in incubated samples, which leads to significant isotopic  
 255 enrichment of carbon with  $^{13}\text{C}$ .



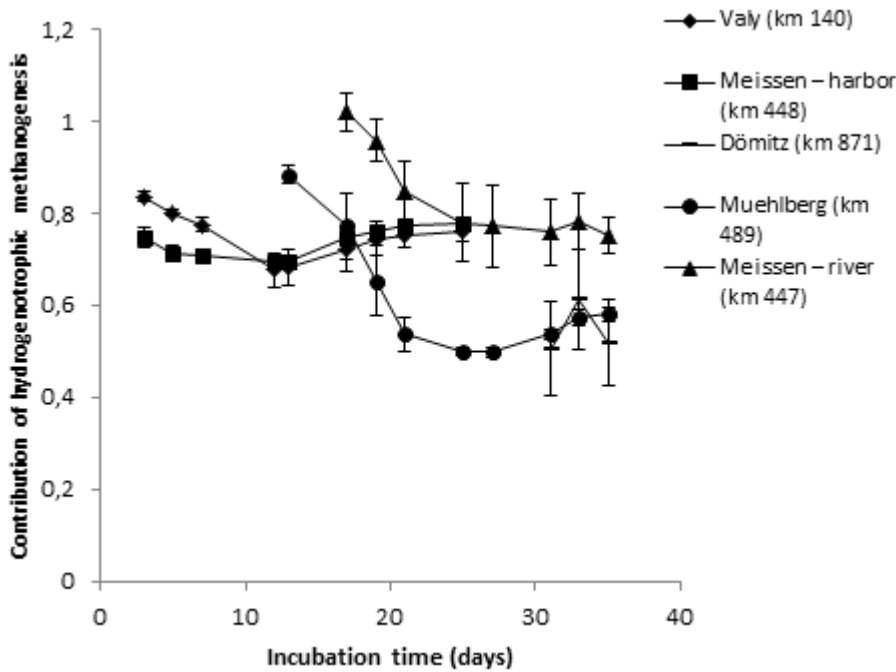
256

257 **Fig. 2:** Methanogenic potential of sediments (mean values ± SE).

258

259

260



261

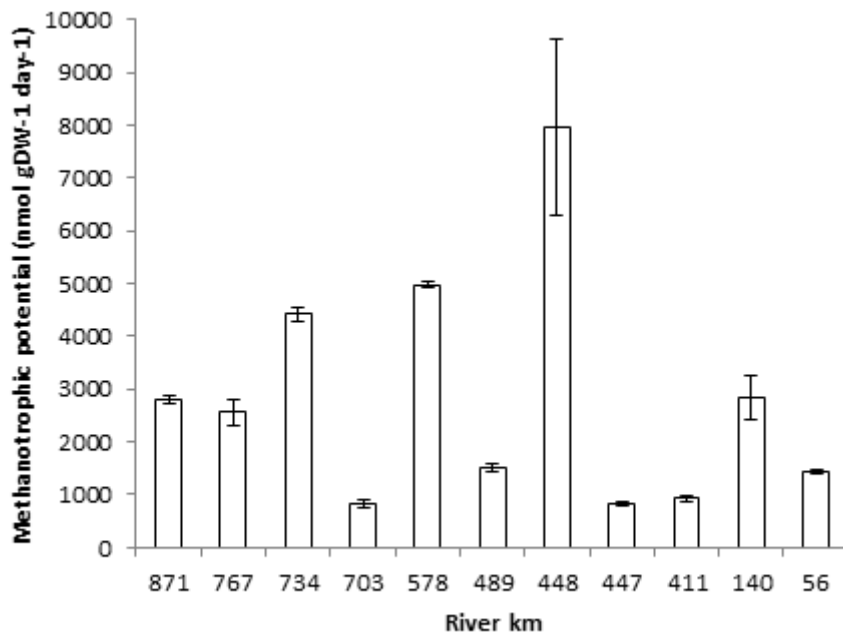
262 **Fig. 3:** Contribution of hydrogenotrophic methanogenesis (H<sub>2</sub>/CO<sub>2</sub>) to total CH<sub>4</sub> production

263 of examined sediments during incubation experiments (mean values ± SE).

264

265

266 The aerobic methanotrophic potential of sediments was recorded for all sites and ranged from  
 267 654 to 10,875 nmol gDW<sup>-1</sup> d<sup>-1</sup> (Fig. 4). The calculated methanotrophic potential of sediments  
 268 was always higher than the methanogenic potential at each sampling site. The highest values  
 269 of methanotrophic potentials were observed at the km 448 (Meissen-harbour). The CH<sub>4</sub>  
 270 production and oxidation in surface sediments varied spatially along the Elbe River  
 271 longitudinal profile without any clear trend.  
 272



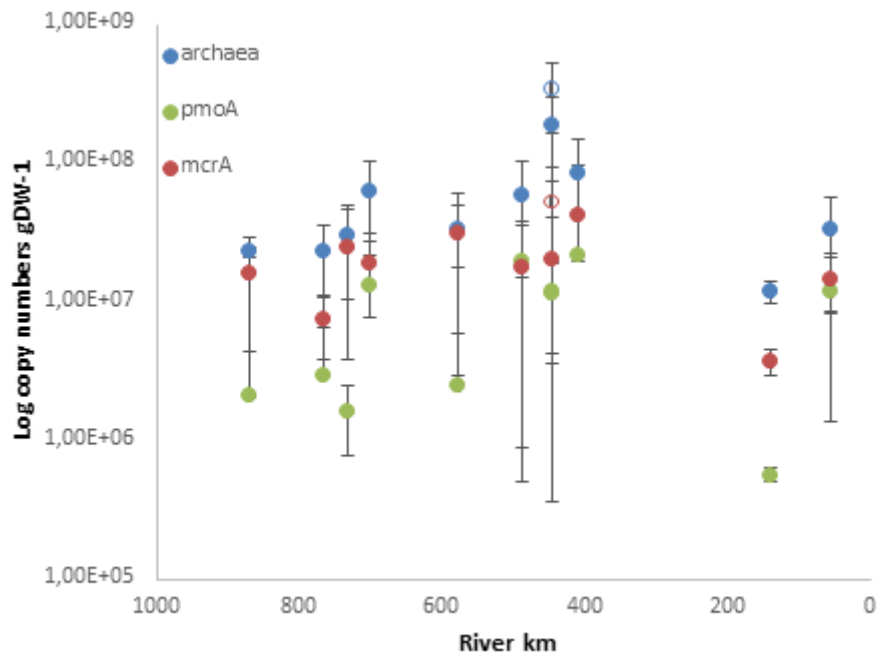
273  
 274 **Fig. 4:** Methane oxidation potential of sediments (mean values ± SE).

275  
 276 **2.2. Population dynamics of archaea (arc), methanogens (mcrA) and methanotrophs**  
 277 **(pmoA) in sediments**

278 **2.2.1. Quantitative polymerase chain reaction (qPCR)**

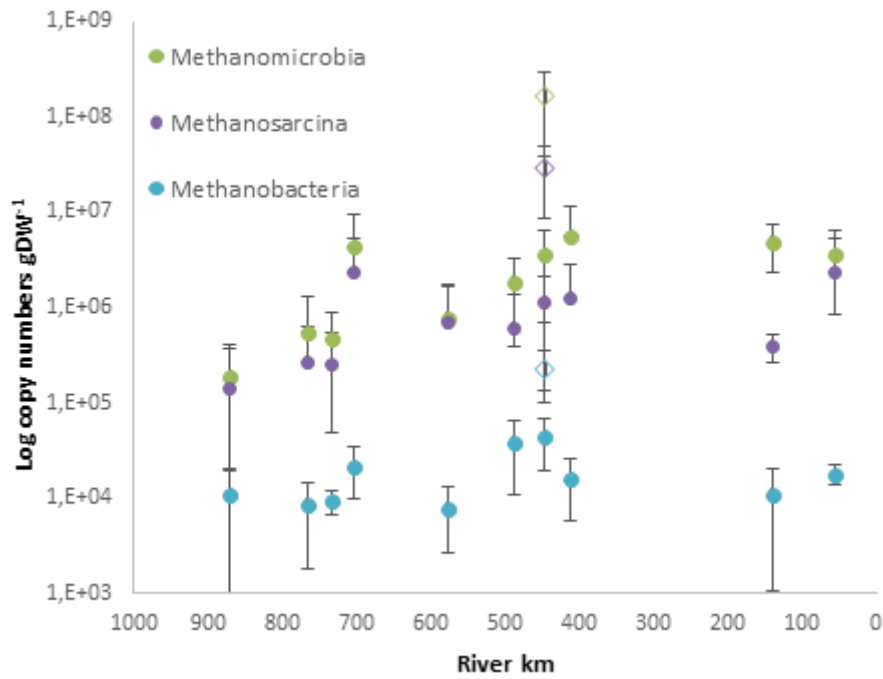
279 The abundances of total archaea (16S rRNA gene), methanotrophs (*pmo-A*, coding methane  
 280 monooxygenases) and methanogens (*mcr-A*, coding for a subunit of the methyl-coenzyme M  
 281 (CoM) reductase) were determined by qPCR in the fresh samples (Fig. 5). The copy number  
 282 of archaeal 16S rRNA genes were relatively stable in the range of 10<sup>7</sup> to 10<sup>8</sup> copies per gram  
 283 dry sediment. The lowest levels were found at the km 140 (Valy), whereas one order of  
 284 magnitude higher values could be reported from the km 448 (Meissen – harbour) and km 447  
 285 Meissen-river sediments. Copy numbers for *mcr-A* and *pmo-A* were 10<sup>6</sup> to 10<sup>7</sup>. While *mcr-A*  
 286 copies were highest at the km 448 (Meissen - harbor), *pmoA* had its maximum at the km 411  
 287 (Wachwitz).





288  
 289 **Fig. 5:** Abundance (log copy numbers gDW<sup>-1</sup>) of total archaea, methanotrophs (pmoA) and  
 290 methanogens (mcrA) in examined sediments (mean values ± SE). Open symbols represent site  
 291 Meissen – harbour (km 448).

292  
 293 The results of the group specific qPCR revealed a similar order of magnitude but showed  
 294 higher fluctuations. The highest counts were reported for *Methanomicrobiales* and  
 295 *Methanosarcinales* (10<sup>5</sup> to 10<sup>8</sup> copies per gram dry sediment). Again the Meissen-harbour  
 296 sample (km 448) showed the highest copy numbers while otherwise there was a slight  
 297 decrease of copy numbers from spring to mouth for both methanogenic orders.  
 298 *Methanobacteriales* showed more variability along the river and ranged from 10<sup>3</sup> to 10<sup>5</sup> copies  
 299 per gram dry sediment (Fig. 6).

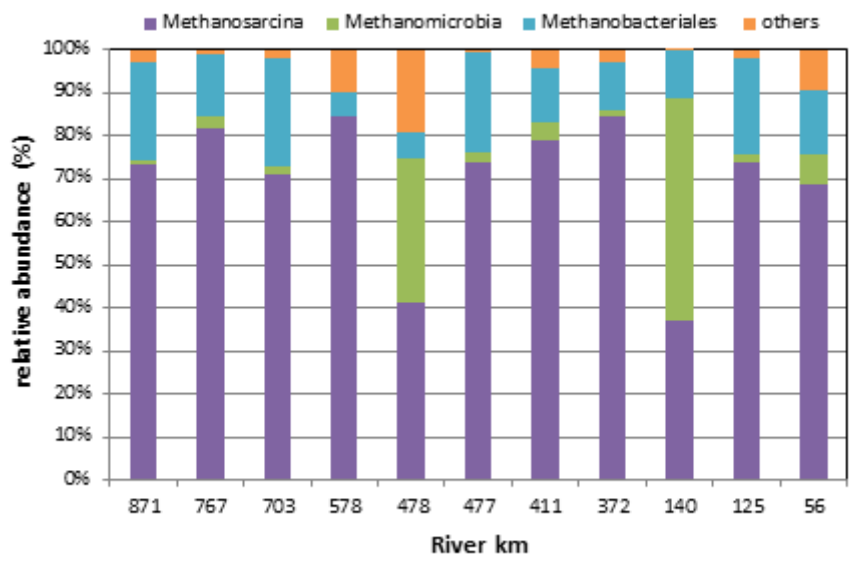


300  
 301 **Fig. 6:** The abundance (log copy numbers gDW<sup>-1</sup>) of individual orders of methanogens in the  
 302 examined samples determined by the group specific qPCR (mean values ± SE). Open symbols  
 303 represent site Meissen – harbour (km 448).

304

305 2.2.2. Terminal restriction length polymorphism of *mcr-A* gene

306 The composition of the methanogenic communities along the Elbe River was determined by  
 307 an analysis of T-RFLP targeting the *mcr-A* gene in the sediment samples (Fig. 7). All  
 308 localities were dominated by the T-RF's attributed to *Methanosarcinales* (27 to 84 %)   
 309 followed by T-RF's assigned to *Methanobacteriales* (5 to 60 %). Methanogens belonging to  
 310 *Methanomicrobiales* were found at a very low level. A number of very long T-RF's of (506-  
 311 10bp) could not be identified to known T-RF's. Details on TRFLP results are given in the  
 312 supplementary material (Fig. S2).



313

314 **Fig. 7:** The relative abundance of individual orders of methanogens in the examined  
 315 sediments determined by analysis of T-RFLP.

316

### 317 **3. Discussion**

#### 318 **3.1. Methanogenic and methanotrophic potential of the sediments**

319 Despite incubation under wet anoxic conditions, methanogenic activity was detected for  
320 roughly half of the samples (six out of eleven sampling sites). This might be caused by lower  
321 level of organic substrates in inactive sediments (carbon content below 1%) or by the  
322 availability of alternative electron acceptors (dissolved  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$ ,  $\text{Fe}^{3+}$ ) in well-oxygenated  
323 river surface sediments (Huttunen et al. 2006; Duc et al. 2010). For instance, hardly detectable  
324  $\text{CH}_4$  production was also observed in Amazonian white water lakes, probably due to relatively  
325 high iron and low organic carbon (below 1.5%) content (Conrad et al. 2014).

326 The  $\text{CH}_4$  formation in the analysed sediments did not show any clear trend along the Elbe  
327 River profile. However, sites with high methanogenic potential (Meissen–harbour and Valy)  
328 were characterised by fine sediment fraction and high organic carbon content compare to  
329 other sampled sites (see Table 1) signifying importance of these local factors in methane  
330 production. Many authors (e.g. Sanders et al. 2007, Maeck et al. 2013, Sollberger et al. 2014,  
331 Bednařík et al. 2017) indicated that sites, where fine and organic matter rich sediment is  
332 accumulated, are particularly active sites of  $\text{CH}_4$  production. Moreover, sites with high  
333 methanogenic potential (Meissen–harbour and Valy) in our study coincided with high  $\text{CH}_4$   
334 concentration in the surface water (for details see Matoušů et al. 2018). This implies that the  
335  $\text{CH}_4$  input into the water column may originate in some hot-spots of  $\text{CH}_4$  production rather  
336 than a continuous supply from the sediment. Despite of nearby existence of large city  
337 (Dresden) and its possible anthropogenic pollution, there was not observed unusually  
338 increased methanogenic activity in the sediment sample from the river channel in Meissen  
339 (km 447) or Wachwitz (km 411). Therefore, increased methanogenic potential in this study is  
340 rather linked to local factors allowing accumulation of sediment than to potential pollution  
341 from city (Meissen or Dresden). Similarly, there were not observed significantly different  
342 physico-chemical parameters in these sites, which is presented in our previous study (Matoušů  
343 et al. 2018). Nevertheless, effect of anthropogenic pollution in rivers on methanogenic activity  
344 and river  $\text{CH}_4$  concentration exists and it was previously reported for instance by Dzyuban  
345 2011 in polluted tributaries of the Rybinsk Reservoir or by Alshboul et al. (2016) in effluents  
346 and receiving streams downstream of the municipal wastewater treatment plants in Germany.  
347 The methanogenic potential presented in this study for the Elbe River is in the range of the  
348 values reported for other streams and rivers (0-1,990  $\text{nmol gDW}^{-1} \text{d}^{-1}$ ; Table 2). Previously  
349 reported methanogenic potential from a lower part of the Elbe River by Gebert et al. (2006)  
350 reaches a similar range of values as measured in this study. This comparison suggests that

351 obtained results reported here probably correspond to general natural capacity of this  
352 environment and values are not skewed. It also shows that similar representative results can  
353 be reached with different methodological approaches (incubation time, sediment amount).  
354 However, methanogenic potential is significantly lower (0.01 and 3.99  $\mu\text{mol gDW}^{-1} \text{d}^{-1}$ )  
355 compared to  $\text{CH}_4$  production in lakes and rice paddy soils (Yao et al. 1999; Conrad et al.  
356 2010; Duc et al. 2010).

357 Results of this study demonstrate that  $\text{CH}_4$  in sediments of the Elbe River is produced  
358 predominantly from  $\text{CO}_2$  reduction. Although acetoclastically produced  $\text{CH}_4$  should  
359 theoretically prevail (Conrad 1999), dominance of hydrogenotrophic methanogenesis is not  
360 unusual for freshwater ecosystems (Krüger et al. 2002; Galand et al. 2010; Conrad et al. 2010;  
361 Conrad et al. 2014). A higher contribution of hydrogenotrophic methanogenesis is probably  
362 connected to only a partial oxidation of organic matter as described in more detail in Conrad  
363 et al. (2009). Although samples in this study were taken from the sediment surface layer (0-10  
364 cm), generally well oxygenated (i.e. with complete degradation of organic matter),  
365 microzones with low oxygen level are likely to occur (Boulton et al. 1998). These anoxic  
366 microzones may provide places for anaerobic processes like methanogenesis and for  
367 incomplete degradation of organic matter in the surface sediments (Deborde et al. 2010). A  
368 more balanced contribution of methanogenic pathways to  $\text{CH}_4$  production was found in the  
369 Sitka stream sediments, while hydrogenotrophically produced  $\text{CH}_4$  reached 36 – 51 % (Mach  
370 et al. 2015). Similarly high contributions of hydrogenotrophic methanogenesis (57 – 90 %)  
371 were detected in the Morava River sediments upstream of the weirs, where fine and organic  
372 rich sediments were accumulated (Bednařík et al. 2017).

373 In contrast, methanotrophic activity (measured under substrate addition) occurred in all  
374 samples, while the  $\text{CH}_4$  oxidation potential was comparable to other reports from various  
375 freshwater ecosystems (Bender and Conrad 1994; Sanders et al. 2007; Shretsha et al. 2010).  
376 However, it should be noted, that the  $\text{CH}_4$  production and oxidation rates were not measured  
377 under in situ conditions and with addition of substrate in case of  $\text{CH}_4$  oxidation  
378 measurements. Thus they represent potential rates suitable for comparison of studied sites, but  
379 they do not allow mass balance calculation. Moreover, comparison of the  $\text{CH}_4$  oxidation rates  
380 between different studies is very limited, because obtained results can be highly affected by  
381 the diverse incubation setting (e.g. incubation temperature, initial  $\text{CH}_4$  concentration) due to  
382 strong substrate and temperature dependence of oxidation rates (Shelley et al. 2015).

383

384

385 **Table 2:** Overview of literature values regarding the isotopic analysis of CH<sub>4</sub> and its production in river sediments

Site	$\delta^{13}\text{C}$ of CH <sub>4</sub> (‰)	f <sub>mc</sub> (%)*	CH <sub>4</sub> production		Reference
			potential	$\delta^{13}\text{C}_{\text{org}}$ (‰)	
			(nmol gDW <sup>-1</sup> d <sup>-1</sup> )		
inflows of Lake Biwa, Japan	-64 to -47	n.s.	n.s.	n.s.	Murase et al. 2003
White Oak River, North Carolina	-70.8 to -65.2	18 to 42	n.s.	n.s.	Avery and Martens 1999
three streams in eastern Amazonia	-75.1 to -52.7	n.s.	n.s.	-29.7 to -22.8	Moura et al. 2008
five rivers in USA	-56.6 to -36	n.s.	n.s.	n.s.	Sansone et al. 1999
Sitka Stream	-98.6 to -48.2	26 to 51	0 to 960	-26.7 to -25.8	Mach et al. 2015
Elbe River	-71.1 to -54.1	52 to 78	0 to 645	-27.2 to -0.4	Bednařík et al. (this study)
Morava River	-63.9 to -52.5	37 to 89	0 to 1,999	-28.5 to -26.2	Bednařík et al. 2017
River Itchen, U.K.	-58	33	528 to 1,920	n.s.	Shelley et al. 2015
Nine rivers in Germany	n.s.	n.s.	120 to 720	n.s.	Gebert et al. 2006
River Frome, England	n.s.	n.s.	48 to 384 <sup>†</sup>	n.s.	Sanders et al. 2007

386 \* f<sub>mc</sub> = part of hydrogenotrophically produced methane

387 <sup>†</sup> nmol CH<sub>4</sub> g<sup>-1</sup> (wet sediment) h<sup>-1</sup>

388 n.s. = not specified

389

390

391

392

### 3.2. Molecular analyses of sediments

The molecular data suggests that even though the overall numbers on the group level are quite stable, the community composition is much more variable. In general *Methanomicrobiales* and *Methanosarcinales* are the most dominant methanogens detected followed by *Methanobacteriales*. This was supported by the TRFLP results, which also showed the dominance of *Methanosarcina* and *Methanomicrobia*. However, the qPCR results can not directly be compared to T-RFLP since T-RFLP is based on the highly degenerated *mcrA* primers and it gives only relative abundances, while the order specific q-PCR is supposed to provide reasonable estimates of absolute numbers for the respective methanogenic order according to the standards used. The methanogenic community based on T-RFLP of *mcrA* has been so far primarily described for rice field soils (Lueders et al. 2001, Ramakrishnan et al. 2001, Chin et al. 2004, Kemnitz et al. 2004, Conrad et al. 2008). In rice field soil the TRFLP patterns are more diverse and contain additional methanogenic orders (Lueders et al. 2001, Ramakrishnan et al. 2001, Chin et al. 2004, Kemnitz et al. 2004, Conrad et al. 2008). A recent study in river sediments also found *Methanosarcina* as dominant TRF (Chaudhary et al. 2017.)

In general, our results are in good agreement with reported methanogenic community profiles from other freshwater habitats: *Methanosarcinales* and *Methanomicrobials* have been described as dominant methanogenic members using various archaea/methanogen-specific primers, e.g. from freshwater river and estuarine sediment (Munson et al. 1997, Purdy et al. 2002, Buriankova et al. 2013, Brablcova et al. 2014), as well as from peat bog sites (Galand et al. 2005), freshwater lake sediments (Falz et al. 1999, Koizumi et al. 2004), Florida Everglades wetland soils (Castro et al. 2004), hydrocarbon-contaminated aquifer (Kleikemper et al. 2005) and deep-sea hydrothermal sediments (Dhillon et al. 2005).

When we compare our molecular methods-based results with the activity measurements, it is obvious that they seemingly do not fit together: while we found ten times higher methanotrophic potential compared to the methanogenic potential, the microbial abundance are quite congruent. However, methanotrophs rely on a constant flux of two gaseous substrates: oxygen and methane and hence appear in high quantities at the oxic-anoxic interface where the oxygen and CH<sub>4</sub> gradients overlap. We analysed a sediment mixture of the top 10 cm where the methanotrophs became dispersed, which may explain why they displayed a high activity when stimulated with a specific substrates. In contrast, methanogens are active in all anoxic parts of the sediment and generally more dispersed than methanotrophs.

427 The second discrepancy is that we found a high contribution of hydrogenotrophic  
428 methanogenesis based on our isotope analysis, while our molecular studies revealed high  
429 numbers of potentially acetoclastic methanogens (*Methanosarcinales*) and a lower number of  
430 the strict hydrogenotrophic methanogens (*Methanobacteriales* and *Methanomicrobiales*). This  
431 may have two reasons: On one hand, *Methanosarcinales* can live acetoclastic as well as  
432 hydrogenotrophic, and hence as the more substrate versatile microbes may use different  
433 substrates according to the environmental conditions. On the other hand, it is generally  
434 observed that the microbial abundance and community patterns only rarely correlate with  
435 their activity (Mach et al. 2015, Chaudhary et al. 2017). This can be confirmed by our results,  
436 demonstrating that the activities (methanogenic and methanotrophic potential rates)  
437 correspond only to a certain extent to the molecular data. *Methanosarcina* have been  
438 described as relatively oxygen tolerant, containing a series of genes encoding oxygen  
439 detoxification (Zhang et al. 2006; Angel et al. 2011).

440

#### 441 **4. Conclusions**

442 The methanogenic potential of the sediments (using the natural available substrate) showed  
443 CH<sub>4</sub> production potential comparable to previously published river systems. However, only  
444 approximately half of the samples could be activated (most probably due to substrate  
445 limitation) and these samples showed a strong variance (over one to two orders of magnitude).  
446 However, all sediment samples showed a methanotrophic potential (under substrate addition),  
447 while it differed by one order of magnitude between studied sites.

448 Molecular analyses of the underlying microbial community revealed constant quantities of  
449 several marker genes (16S archaea, mcrA, pmoA) over the river continuum. This suggest that  
450 the observed variability of the microbial activities (i.e. CH<sub>4</sub> production and oxidation) as well  
451 as the resulting CH<sub>4</sub> concentrations in the water column are only indirectly linked to the  
452 presence of different microbial guilds, but rather affected by their activity (which has not  
453 directly been tested in this study). Further insight into the methanogenic community (TRFLP  
454 and the group specific qPCR) revealed large variations in the methanogenic populations of  
455 different sediment samples.

456 Coming back to our original hypothesis mentioned in the introduction, we see no trend in the  
457 studied parameters along the Elbe River. However, we found hotspots of the measured CH<sub>4</sub>  
458 processes (Fig. 2 and 4) due to other local driving factors (e.g. carbon content, etc.).  
459 Therefore, based on results presented in this study, spatial heterogeneity of sediment  
460 characteristics (grain median size, carbon content) seems to be more relevant for prediction of



461 methanogenic or methanotrophic activity than only abundance of methanogens or  
462 methanotrophs.

463

464

465

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# Sediment methane dynamics along the Elbe River

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## Supplementary material

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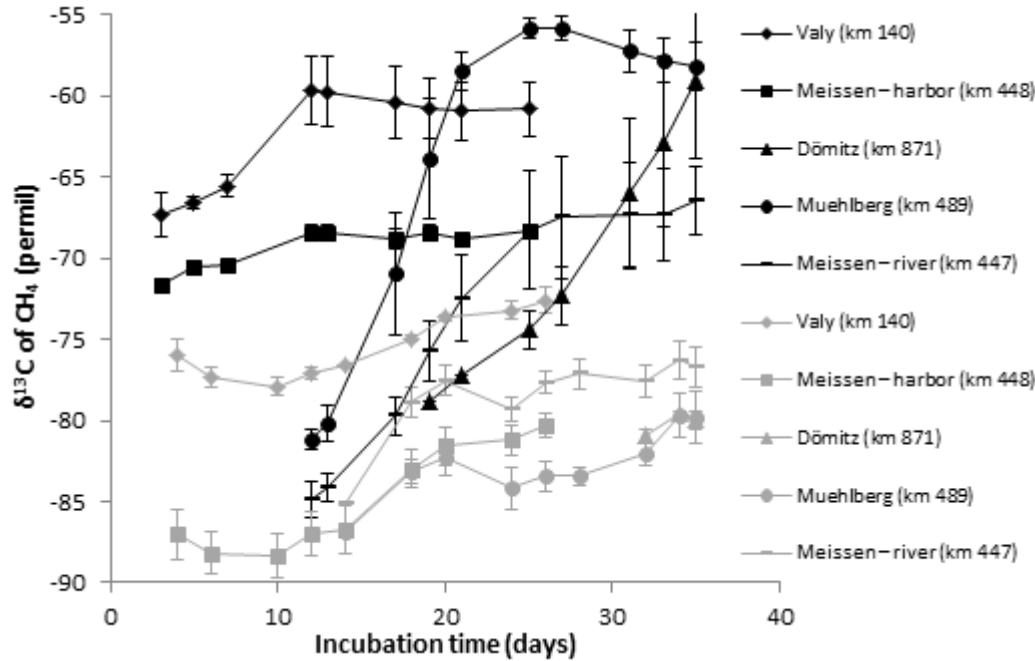
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22

23 Fig S1: Delta <sup>13</sup>C values of methane for sediment incubations of different methanogenic sites.  
 24 In black incubations under N<sub>2</sub>; in grey incubations under N<sub>2</sub> + 3% CH<sub>3</sub>F.

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26 In the well studied systems (e.g. rice paddies and lake sediments) methane emission can be  
 27 linked to two dominating processes: acetoclastic (eq. 1) and hydrogenotrophic (eq. 2)  
 28 methanogenesis:



31 To distinguish the two dominant methanogenic pathways the natural abundance of stable  
 32 carbon isotopes can be used if the  $\delta^{13}\text{C}$  of methane and of its precursors and the methanogenic  
 33 fractionation factors are known (Conrad, 2005). The acetoclastic methanogenesis expresses a  
 34 smaller kinetic isotopic ( $\alpha = 1.009 - 1.027$ ) effect (Gelwicks et al., 1994; Govert and Conrad,  
 35 2009; Penning et al., 2006) than the hydrogenotrophic methane formation ( $\alpha = 1.045 - 1.073$ )  
 36 (Valentine et al., 2004). The inhibition of acetoclastic methanogenesis by methylfluoride  
 37 (CH<sub>3</sub>F) allows quantifying the contribution of both pathways (Conrad et al., 2011; Janssen  
 38 and Frenzel, 1997).

39 While the acetoclastic pathway is dominating in e.g. rice paddy soils (up to 67% of methane  
40 release (Conrad, 1999)) freshwater sediments and gut environments are dominated by  
41 hydrogen driven methanogenesis (Conrad, 1999).

## 42 *Calculations*

43 The carbon isotopic signature was given in the delta notation relative to the Vienna Pee Dee  
44 Belemnite (V-PDB) standard. The fractionation factor  $\alpha$  for a reaction  $A \rightarrow B$  are defined  
45 after (Hayes, 1993):

$$46 \quad \alpha_{A, B} = (\delta^{13}\text{C}_A + 10^3) / (\delta^{13}\text{C}_B + 10^3) \quad (3)$$

47 Isotopic calculations of fractionation factors and estimation of the approximate partition of  
48 hydrogenotrophic methanogenesis of the total methanogenesis were calculated according to  
49 (Conrad, 2005):

50 The apparent fractionation factor ( $\alpha_{\text{app}}$ ) for conversion of  $\text{CO}_2$  to  $\text{CH}_4$  is given by:

$$51 \quad \alpha_{\text{app}} = (\delta\text{CO}_2 + 10^3) / (\delta\text{CH}_4 + 10^3) \quad (4)$$

52 where  $\delta\text{CO}_2$  and  $\delta\text{CH}_4$  are directly measured isotopic signatures of the carbon in  $\text{CO}_2$  and  
53  $\text{CH}_4$ , respectively.

54 Fractionation factor for hydrogenotrophic methanogenesis ( $\alpha_{\text{mc}}$ ) is given by:

$$55 \quad \alpha_{\text{mc}} = (\delta\text{CO}_2 + 10^3) / (\delta_{\text{mc}} + 10^3) \quad (5)$$

56 where  $\delta_{\text{mc}}$  is carbon isotopic signature of methane solely produced from carbon dioxide  
57 (directly measured from assays inhibited by methylfluoride). Partition of hydrogenotrophic  
58 methanogenesis is calculated by the following mass balance equation (6):

$$59 \quad f_{\text{mc}} = (\delta_{\text{CH}_4} - \delta_{\text{ma}}) / (\delta_{\text{mc}} - \delta_{\text{ma}}) \quad (6)$$

60 where  $f_{mc}$  is the partition of hydrogenotrophic methanogenesis and  $\delta_{ma}$  is carbon isotopic  
61 signature of methane solely produced from acetate. It is calculated from the following  
62 equation:

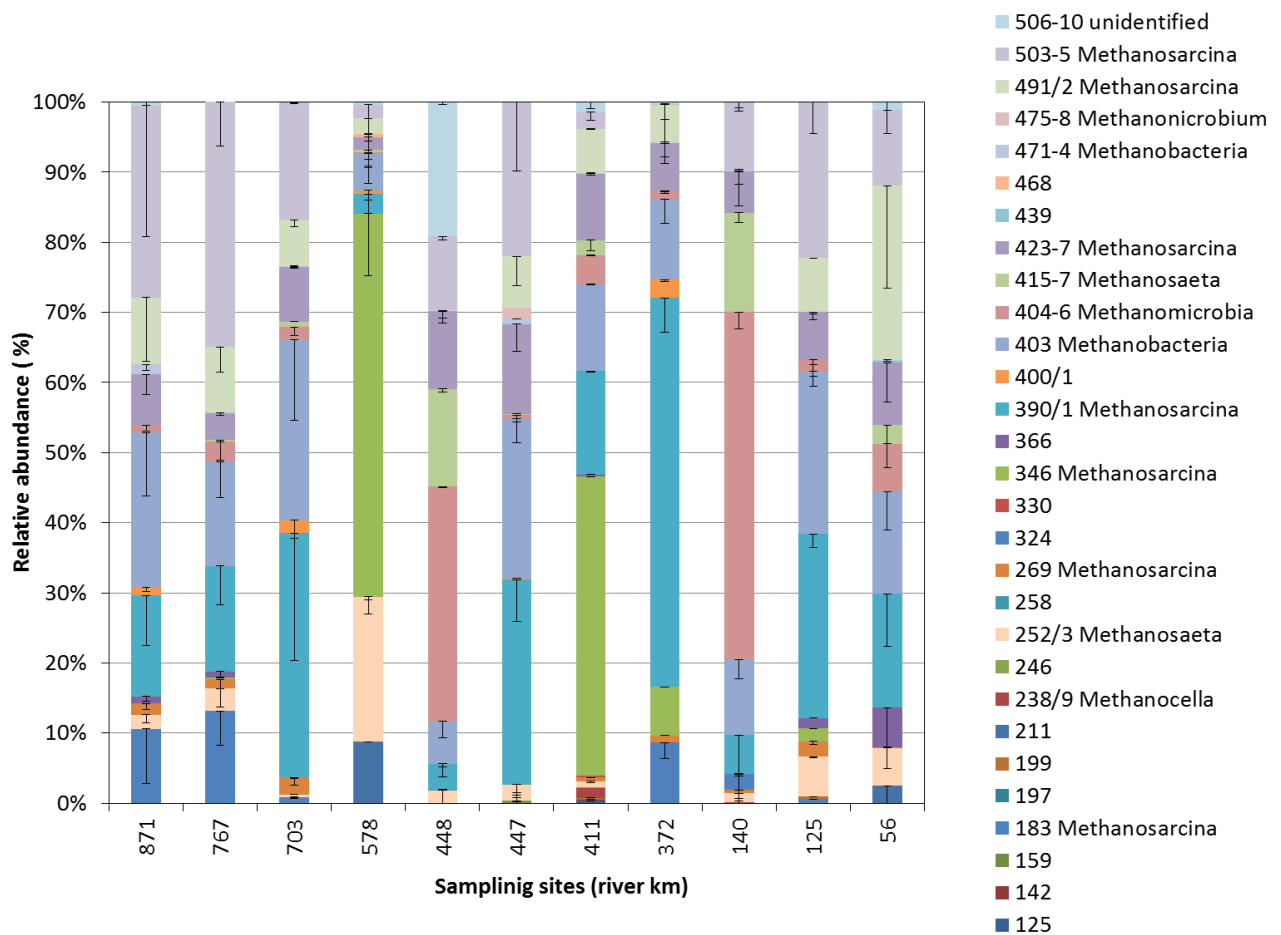
$$63 \quad \delta_{ma} = (1/\alpha_{ma})(\delta_{ac} + 10^3 - \alpha_{ma} * 10^3) \quad (7)$$

64 where  $\alpha_{ma}$  is fractionation factors for acetoclastic methanogenesis and  $\delta_{ac}$  is the measured  
65 isotopic signal of acetate. Several published  $\alpha_{ma}$  have been used to estimate the contribution of  
66 hydrogenotrophic methanogenesis e.g. (Gelwicks et al., 1994; Goevert and Conrad, 2009;  
67 Penning et al., 2006). We used  $\alpha_{ma}$  1.009 for our calculations which is a somewhat moderate  
68 assumption used by several authors e.g. (Aschenbach et al., 2013)

69

Name	Target Group	Sequence(5' - 3')	Annealing Temperature (°C)	Amplicon size (bp)	Reference
<b>PARCH340-F</b> <b>PARCH519-R</b>	Archaea	CCC TAC GGG GYG CAS CAG TTA CCG CGG CKG CTG	58.3	152	Øvrea's et al. 1997
<b>pmoA 189-F</b> <b>pmoA682-R</b>	Methanotrophs	GGN GAC TGG GAC TTC TGG GAA SGC NGA GAA GAA SGC	56	531	Holmes et al. 1995
<b>MCRA-F</b> <b>MCRAR- R</b>	Methanogens	GGT GGT GTM GGD TTC ACM CAR TA TTC ATT GCR TAG TTW GGR TAG TT	55	488	Luton et al. 2002
<b>MBT857-F</b> <b>MBT1196-R</b>	Methanobacteriales	CGW AGG GAA GCT GTT AAG T TAC CGT CGT CCA CTC CTT	53.4	342	Yu et al. 2005
<b>MMB282-F</b> <b>MMB832-R</b>	Methanomicrobiales	ATC GRT ACG GGT TGT GGG CAC CTA ACG CRC ATH GTT TAC	50.7	506	Yu et al. 2005
<b>MSL812-F</b> <b>MSL1159-R</b>	Methanosarcinales	GTA AAC GAT RYT CGC TAG GT GGT CCC CAC AGW GTA CC	52.7	354	Yu et al. 2005





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73 Fig S2: Relative abundance of methanogenic groups as determined by T-RFLP of *mcrA* in the  
 74 Elbe River sediments

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## **7.2. Effect of weir impoundments on methane dynamics in a river**

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## Effect of weir impoundments on methane dynamics in a river



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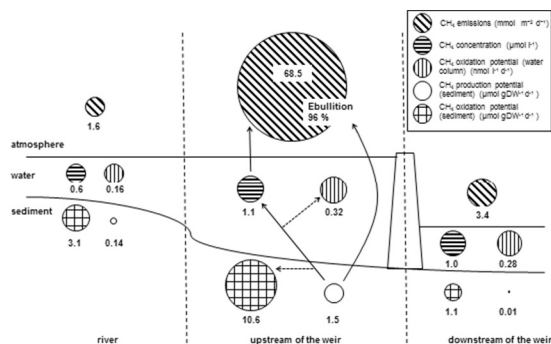
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### HIGHLIGHTS

- CH<sub>4</sub> emissions were higher upstream of the weirs compared to river reaches.
- Sediments upstream of the weirs resemble many previous observations for lake systems.
- Small impoundments significantly affect the CH<sub>4</sub> cycle in a river.

### GRAPHICAL ABSTRACT



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### ABSTRACT

We measured CH<sub>4</sub> concentration, CH<sub>4</sub> oxidation in the water column and total CH<sub>4</sub> emissions to the atmosphere (diffusion and ebullition) in three weir impoundments and river reaches between them, in order to understand their role in river methane (CH<sub>4</sub>) dynamics. Sediment samples were also collected to determine CH<sub>4</sub> consumption and production potentials together with the contribution of individual methanogenic pathways. The CH<sub>4</sub> surface water concentration increased 7.5 times in the 16 km long river stretch. Microbial CH<sub>4</sub> oxidation in the water column reached values ranging from 51 to 403 nmol l<sup>-1</sup> d<sup>-1</sup> and substantially contributed to the CH<sub>4</sub> removal from surface water, together with CH<sub>4</sub> emissions. The total CH<sub>4</sub> emissions to the atmosphere varied between 0.8 and 207.1 mmol CH<sub>4</sub> m<sup>-2</sup> d<sup>-1</sup> with the highest values observed upstream of the weirs (mean 68.5 ± 29.9 mmol CH<sub>4</sub> m<sup>-2</sup> d<sup>-1</sup>). Most of the CH<sub>4</sub> was transported through the air-water interface by ebullition upstream of the weirs, while the ebullition accounted for 95.8 ± 2.0% of the total CH<sub>4</sub> emissions. Both CH<sub>4</sub> production and oxidation potential of sediments were higher upstream of the weirs compared to downstream of the weirs. The contribution of hydrogenotrophic methanogenesis to total CH<sub>4</sub> sediment production was 36.7–89.4% and prevailed upstream of the weirs. Our findings indicate that weirs might influence river CH<sub>4</sub> dynamics, especially by increased CH<sub>4</sub> production and consumption by sediments, followed by increasing CH<sub>4</sub> emissions to the atmosphere.

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## 1. Introduction

Methane (CH<sub>4</sub>) emissions from inland freshwater ecosystems (lakes, reservoirs and rivers) are believed to be important contributions to global CH<sub>4</sub> flux (Ciais et al., 2013; Bastviken et al., 2011). It has been estimated that CH<sub>4</sub> effluxes from lakes or wetlands are equivalent to ca. 20–50% of global CH<sub>4</sub> emission, but fluvial systems are one of the least studied freshwater types (Stanley et al., 2016). The controlling factors of CH<sub>4</sub> fluxes from rivers and the characteristics of their spatial and temporal heterogeneity are still poorly understood (Saarnio et al., 2009; Ortiz-Llorente and Alvarez-Cobelas, 2012).

One major factor affecting the CH<sub>4</sub> dynamic in rivers is the presence of dams. In general, impoundments can cause significant changes to river habitats, morphology, sediment transport and physicochemical properties of the water (Ogbeibu and Oribhabor, 2002; Rickard et al., 2003; Gao et al., 2013; Kattel et al., 2016) and is also known to be important in greenhouse gas dynamics (Louis et al., 2000; Sobek et al., 2012). Impoundments in central Europe are environments with a high organic carbon burial rate; with smaller impoundments having greater deposition and accumulation rates per unit area (Downing et al., 2008). There is a high contribution of small impoundments to the total area of impoundments, making it obvious that small aquatic bodies can play an important role in the global carbon cycles (Downing et al., 2006; Holgerson and Raymond, 2016).

The microbial processes related to CH<sub>4</sub> production occur to a large extent in the hyporheic zone of streams and rivers. Here anaerobic carbon cycling is the prevailing process and CH<sub>4</sub> is one of the major components of interstitial dissolved organic carbon (Dahm et al., 1991; Baker et al., 1999; Fischer et al., 2005).

Generally, H<sub>2</sub>/CO<sub>2</sub> and acetate have been recognized as the two dominant substrates for methanogenic Archaea in freshwater ecosystems. When carbohydrates are anaerobically degraded to CH<sub>4</sub> and CO<sub>2</sub>, acetoclastic methanogenesis (using acetate) would theoretically contribute 67% of CH<sub>4</sub> production following the stoichiometry of the degradation, while the contribution of hydrogenotrophic methanogenesis (using H<sub>2</sub>/CO<sub>2</sub> as substrate) would contribute the remaining 33%. This holds true for rice field soils (Conrad et al., 2002; Fey et al., 2004; Scavino et al., 2013), but recent studies from numerous freshwater ecosystems show various contributions of the two dominating methanogenic pathways. Low contributions of hydrogenotrophic methanogenesis (using H<sub>2</sub>/CO<sub>2</sub>) have been reported for rivers – 18–45% (Avery and Martens, 1999; Mach et al., 2015). Much higher contributions of hydrogenotrophic methanogenesis have been frequently reported for lakes – 50–90% (Murase and Sugimoto, 2001; Conrad et al., 2011; Conrad et al., 2014) and peatlands – 46–89% (Galand et al., 2005; Galand et al., 2010). Nevertheless, very few data are available concerning spatial variations of methanogenic pathways in freshwater sediments, particularly those from rivers.

Damming reduces the flow velocity of the water, which in turn decreases the oxygen availability. Hence CH<sub>4</sub> produced in the sediment is less likely to be oxidised in the sediment and the water column. This has two consequences: (1) the increased production and reduced oxidation of CH<sub>4</sub> results in a supersaturation and increased ebullition events and (2) the supersaturated water increases the CH<sub>4</sub> concentration in downstream water sections.

Sites with increased sedimentation and CH<sub>4</sub> concentration are thus considered to be ‘hot spots’ of CH<sub>4</sub> ebullition (Sobek et al., 2012; Maeck et al., 2013). Bubble release is a frequent transportation pathway of CH<sub>4</sub> from lakes and reservoirs (Bastviken et al., 2004; DelSontro et al., 2010), while it is considered to be less important in natural streams and rivers, where most of the emission measurements have focused on transport through the air–water interface by diffusion (Striegl et al., 2012; Yang et al., 2012; Silvennoinen et al., 2008). There is now good evidence however, that bubble-mediated fluxes are an important CH<sub>4</sub> emission mechanism in running waters (Baulch et al., 2011; Crawford et al., 2014; Sawakuchi et al., 2014).

The aim of this study was to quantify the effect of three weirs and weir impoundments on individual components of the CH<sub>4</sub> dynamic in a 16 km river reach. Our hypotheses were that impoundments cause changes in sediment composition that leads to increased rate of CH<sub>4</sub> production and reduced rate of CH<sub>4</sub> consumption; these changes lead to increased CH<sub>4</sub> emissions to the atmosphere. We examined 1) changes in river CH<sub>4</sub> concentrations, 2) emissions to the atmosphere (diffusion and ebullition), 3) methanotrophic activity in water column and 4) CH<sub>4</sub> production and consumption by sediments including the determination of methanogenic pathways using stable carbon isotope values.

## 2. Material and methods

### 2.1. Study site

The study river was the Morava (Czech Republic), where it flows through Olomouc city (from 49°36.8′ N, 17°15.2′ E to 49°29.7′ N, 17°16.7′ E). The Morava is a second-order tributary of the Danube with a mean annual discharge of 26.4 m<sup>3</sup> s<sup>-1</sup> at our study site. The Morava river catchment is characterised by the development of the relief features on the marginal West-European platform, young Carpathian folded mountain ranges and of the Pannonian basin. The present georelief has been characterised by alternating period of quiet development and by periods of abrupt changes. About 54% of the catchment area is agricultural land (45% arable), 34% is covered by forest, 1.5% urbanized areas and 1.4% is covered by water. The main sources of nutrients are municipalities and agricultural activities. Coniferous forests prevail in the upper part of the catchment. The river upstream from the study area meanders through the floodplain forests of the Protected Landscape Area Litovelské Pomoraví, however our study reach of Morava River is straightened, its cross section is modified and its river banks are stabilized.

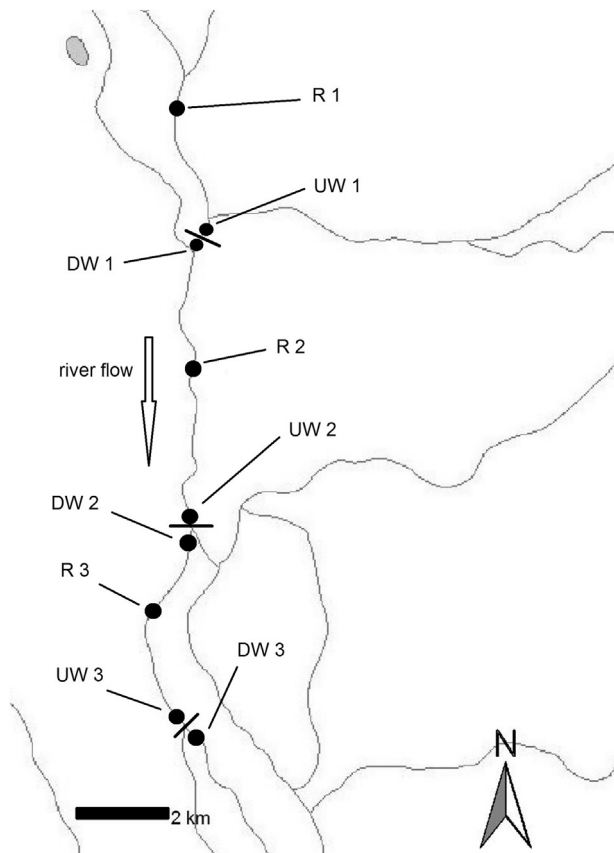
Three weirs are situated in this 16 km stretch of the river (Table 1), constructed for stabilization of the vertical alignment and of the riverbed, energy production using small hydropower plants, and retention of surface water for water supply. Nine sampling points were chosen along the river (Fig. 1). Three sites were located in river parts unaffected by impoundments, between the weirs (R 1–3); three sites were directly in impoundments upstream of the weirs (UW 1–3) and three sites were situated just downstream of the weirs (DW 1–3). All samples were collected twice in the summer months of 2014, one week, in mid-July and one in mid-August. Measurements of all parameters were performed simultaneously for each sampling site. Data from both sampling periods are presented together.

### 2.2. Sediment samples and incubation experiments

Triplicate samples of surface sediment layer (0–10 cm) from each site were collected in July 2014 by scuba diving. Sediments were then sieved through a 1-mm sieve and stored at 4 °C until subsequent analyses and laboratory experiments. Sediments for the granulometric analysis were sieved through a system of ten sieves of decreasing mesh sizes. All separate parts of the sediment were weighted and grain median size was analyzed using the software Gradistat (version 8.0) (Blott and Pye, 2001). The dry weight of the sample was determined gravimetrically. The carbon content of the sediments was quantified on a CHNS-

**Table 1**  
Basic parameters of weirs.

Weir	Backwater length (m)	Volume of reservoir (m <sup>3</sup> )	Weir length at overflow edge (m)	Maximum water depth (m)	Height of weir (m)
W1	2600	139,000	40.8	3.2	2.7
W2	2000	160,000	40.0	4.1	2.4
W3	5290	450,000	50.6	2.7	3.7



**Fig. 1.** Schematic map identifying the nine sampling sites along the studied section of Morava river (UW = upstream of the weir, DW = downstream of the weir, R = river section).

element analyzer by the Analytical Chemical Laboratory of the University of Marburg.

Investigation of methanogenic pathways and methane production potential used approximately 30 g (wet weight) of the sediment in duplicate, transferred into 60-ml sterile serum bottles, flushed with  $N_2$ , closed with butyl rubber stoppers and incubated at 25 °C in the dark. Five ml of distilled autoclaved water was added into each bottle at the start of the incubation for later sampling of the liquid phase. The gas headspace of half of the bottles was supplemented with 3%  $CH_3F$  to specifically inhibit acetotrophic methanogenesis (Janssen and Frenzel, 1997). Gas samples (200  $\mu$ l) were taken repeatedly (twice a week) during the course of incubation (4–6 weeks) and analyzed for concentrations of  $CH_4$ ,  $CO_2$  and  $\delta^{13}C$  of  $CH_4$  and  $CO_2$ . At the end of the incubation, the bottles were sacrificed to sample the liquid phase, stored frozen (–20 °C) for later analyses of concentration and  $\delta^{13}C$  of acetate. The rate of  $CH_4$  production was calculated from the slope of the linear regression given by the graph of  $CH_4$  concentration increase over time.

Methane oxidation potential of the sediment was determined in triplicate for each sample. Sterile bottles (250 ml) were filled with 20 g of sediment (wet weight) and closed by a cap with PTFE silicone septa. The headspace (ambient air) was enriched with  $CH_4$  to give a final concentration of 10,000 ppm and incubated at 25 °C in dark. The concentration of  $CH_4$  in the headspace of each bottle was measured at 0 h and then nine times during 185 h. Potential  $CH_4$  oxidation rates were obtained from the slope of the  $CH_4$  decrease over time.

In the sediment incubation experiments,  $CH_4$  was analyzed by gas chromatography (GC) using a flame ionization detector (Shimadzu, Kyoto, Japan), while  $CO_2$  was analyzed after conversion to  $CH_4$  with a methanizer (Ni-catalyst at 350 °C, Chrompack, Middelburg, Netherlands). Isotope measurements of  $^{13}C/^{12}C$  in gas samples were performed on a gas chromatograph combustion isotope ratio

mass spectrometer (GC–C–IRMS) system (Thermo Fisher Scientific, Bremen, Germany). The principal operation was described by Brand (Brand, 1996). Other details are given in Penger et al., 2012; Blaser et al., 2013; Penger et al., 2014.

Isotopic analysis and quantification of acetate were performed on a high pressure liquid chromatography (HPLC) system (Spectra System P1000, Thermo Fisher Scientific, San Jose, CA, USA; Mistral, Spark, Emmen, the Netherlands) equipped with an ion-exclusion column (Aminex HPX-87-H, BioRad, München, Germany) and coupled to Finnigan LC IsoLink (Thermo Fisher Scientific, Bremen, Germany) as described (Krummen et al., 2004). Isotope ratios were detected on an IRMS (Finnigan MAT Deltaplus Advantage).

The carbon isotopic signature was given in the delta notation relative to the Vienna Pee Dee Belemnite (V-PDB) standard. The fractionation factor  $\alpha$  for a reaction  $A \rightarrow B$  is defined after (Hayes, 1993) as:

$$\alpha_{AB} = (\delta^{13}C_A + 10^3) / (\delta^{13}C_B + 10^3) \quad (1)$$

Isotopic calculations of fractionation factors and estimation of the approximate partition of hydrogenotrophic methanogenesis to total methanogenesis were calculated according to Conrad (2005):

The apparent fractionation factor ( $\alpha_{app}$ ) for conversion of  $CO_2$  to  $CH_4$  is given by:

$$\alpha_{app} = (\delta_{CO_2} + 10^3) / (\delta_{CH_4} + 10^3) \quad (2)$$

where  $\delta_{CO_2}$  and  $\delta_{CH_4}$  are directly measured isotopic signatures of the carbon in  $CO_2$  and  $CH_4$ , respectively.

Fractionation factor for hydrogenotrophic methanogenesis ( $\alpha_{mc}$ ) is given by:

$$\alpha_{mc} = (\delta_{CO_2} + 10^3) / (\delta_{mc} + 10^3) \quad (3)$$

where  $\delta_{mc}$  is carbon isotopic signature of methane solely produced from carbon dioxide (directly measured from assays inhibited by methylfluoride). Partition of hydrogenotrophic methanogenesis is calculated by the following mass balance equation:

$$f_{mc} = (\delta_{CH_4} - \delta_{m\alpha}) / (\delta_{mc} - \delta_{m\alpha}) \quad (4)$$

where  $f_{mc}$  is the partition of hydrogenotrophic methanogenesis and  $\delta_{ma}$  is carbon isotopic signature of methane solely produced from acetate, the latter calculated from the following equation:

$$\delta_{m\alpha} = (1/\alpha_{ma})(\delta_{ac} + 10^3 - \alpha_{ma}10^3) \quad (5)$$

where  $\alpha_{ma}$  is fractionation factor for acetoclastic methanogenesis ( $\alpha_{ma} = 1.009$ ; Govert and Conrad, 2009) and  $\delta_{ac}$  is the measured isotopic signal of acetate.

### 2.3. Physical parameters of water

All physical parameters were measured at the depth from which the water samples were collected (10 cm below the water level), except flow velocity, which was calculated from several measuring depths. Concentration of dissolved oxygen together with temperature was measured using an oximeter Eutech DO 450. Conductivity was measured using a Hanna DiST 3 conductivity meter HI 98303. Flow velocity was measured using a portable flowmeter (Marsh-McBirney model 2000 Flo-Mate).

### 2.4. Water samples

Surface water samples ( $n = 4–8$  per sampling site) for analysis of dissolved  $CH_4$  were collected at a depth of 10 cm below the water

level, in vials (40 ml) with screw-tops, covered by a polypropylene cap with PTFE silicone septa. They were immediately treated by injecting concentrated sulfuric acid (200 µl) to stop the microbial activity (final concentration ca. 90 mM). All samples were transported to the laboratory in a cool box.

Concentrations of dissolved CH<sub>4</sub> in the surface water were measured using headspace equilibration. Dissolved CH<sub>4</sub> was extracted from the water by replacing 15 ml of water with N<sub>2</sub> and then vigorously shaking the vials for 30 s. to release the supersaturated gas from the water to facilitate equilibration between the water and gas phases. All samples were equilibrated at laboratory temperature. CH<sub>4</sub> was analyzed from the headspace by injecting 0.2 ml of gas sub-sample with a gas-tight syringe with valve into a 6890 N (Agilent, USA) gas chromatograph, equipped with a flame ionization detector, with a 0.53 mm × 30 m GS Alumina column. Gas concentration in water was calculated using Henry's law.

Dissolved organic carbon (DOC) was analyzed in samples filtered through glass-fibre filters of 0.4-µm nominal pore size (GF-5, Macherey-Nagel, Düren, Germany) with a Shimadzu TOC-5000A analyzer (Shimadzu, Japan).

The CH<sub>4</sub> oxidation rate was determined as outlined in [Busmann et al. \[2015\]](#). Briefly, water samples were processed in triplicates with two killed controls. Immediately after collecting the samples we injected 100 µl (10 µCi) of <sup>3</sup>H-CH<sub>4</sub> (American Radiolabeled Chemicals, Inc.). The samples were vigorously shaken for 60 s and incubated in the dark at near in situ temperature. Control samples were killed by injecting 200 µl of concentrated sulphuric acid before the tracer was added. Activities in "live" samples were stopped the same way, but after approximately 24 h. Samples were stored in the dark at 4 °C, prior to being analyzed within 1 week. In the laboratory the samples were opened, and the total radioactivity of the sample - including the labelled CH<sub>4</sub> and labelled produced H<sub>2</sub>O - measured immediately by mixing a 1 ml aliquot of each sample with 5 ml of scintillation cocktail (Ultima Gold™ LLT) and analyzed with a liquid scintillation counter (Tri-Carb® 2910 TR, Perkin Elmer; or Tri-Carb® 2900 TR, Packard). Subsequently the samples were sparged with air for half an hour to expel all remaining CH<sub>4</sub>. Afterwards the microbially-produced radioactive water was analyzed in the same way, by mixing a 1 ml aliquot of each sample with 5 ml of scintillation cocktail and analyzed on liquid scintillation counter as described above. The calculation of the microbial mediated CH<sub>4</sub> oxidation is based on the transformation of added radioactively marked tracer (<sup>3</sup>H-CH<sub>4</sub>) into the oxidation products (<sup>3</sup>H-H<sub>2</sub>O) during a timed incubation (detailed in [Busmann et al., 2015](#)).

## 2.5. Emissions to the atmosphere

Methane emissions (diffusion + ebullition) across the air-water interface were determined by a floating chamber method. The open-bottom floating polyethylene chambers (volume 3.1 l covering an area of 0.024 m<sup>2</sup>) were deployed on the water surface (see [Rulík et al., 2013](#); [Bednařík et al., 2015](#)). The chambers (n = 5 per sampling site) were anchored and allowed to float on the water's surface for a period of 3 hours. Pre-incubations were performed to assess linearity of gas concentrations in headspace of the chambers and establish the incubation time required for reliable flux measurements. Samples of headspace gas were collected through the rubber stopper inserted at the chamber's top using a 100 ml gas-tight syringe. The syringe was pumped two times to mix the chamber content before withdrawing 50 ml of gas from the chamber. Samples of ambient air were collected in each study point for determining the initial background concentrations. Emissions were calculated as the difference between initial background and final concentration in the chamber headspace, and expressed according to Eq. (6):

$$E = [(c_I - c_R) \times V \times 24 / (t \times 1000)] / p \quad (6)$$

where  $E$  is gas emission in mmol m<sup>-2</sup> day<sup>-1</sup>;  $c_I$  is concentration of the methane in the chamber headspace in µmol l<sup>-1</sup>;  $c_R$  is concentration of the methane in background air in µmol l<sup>-1</sup>;  $V$  is volume of the chamber in l;  $t$  is time of incubation in hr.;  $p$  is the area of chamber expressed in m<sup>2</sup>.

We determined CH<sub>4</sub> diffusion fluxes to the atmosphere using calculations derived from recent studies ([Striegl et al., 2012](#); [McGinnis et al., 2014](#); [Borges et al., 2015](#); [Bodmer et al., 2016](#)).

The air-water CH<sub>4</sub> diffusion flux ( $F$ ) was computed according to

$$F = k(c_w - c_a) \quad (7)$$

where  $k$  is the gas transfer velocity for water-air gas exchange (m d<sup>-1</sup>),  $c_w - c_a$  is the gas concentration gradient between the river and the atmosphere (mol m<sup>-3</sup>).

For the river flux calculations  $k$  is defined as ([Fortescue and Pearson, 1967](#))

$$k = 1.46(DV/h)^{-0.5} \quad (8)$$

where  $D$  is the molecular diffusion coefficient of CH<sub>4</sub> in the water ([Broecker and Peng, 1974](#)),  $V$  is the water velocity (m s<sup>-1</sup>) and  $h$  is the water depth (m).

In addition to that, we used  $E$  (Eq. (6)) to calculate diffusive gas transfer velocity ( $k$ ) by inverting the equation for Fick's law of gas diffusion, as follows:

$$k = E / (c_w - c_a) \quad (9)$$

where  $k$  is the gas transfer velocity (m d<sup>-1</sup>),  $E$  is taken from Eq. (6) and  $c_w - c_a$  is the gas concentration gradient between the river and the atmosphere (mol m<sup>-3</sup>).

For better comparison of  $k$  with different studies, we standardized  $k$  of CH<sub>4</sub> to  $k_{600}$  (equivalent to  $k$  of CO<sub>2</sub> at 20 °C) computed according to:

$$k_{600} = k(600/S_c)^{-0.5} \quad (10)$$

where  $k$  is the calculated from Eq. (8),  $S_c$  is the Schmidt number for CH<sub>4</sub> in situ temperature ([Wanninkhof, 1992](#)).

Ebullition measurements were carried out, using triplicate submerged gas traps located at three sites upstream of the weirs (UW 1–3) and one river stretch (R 3) on Morava River. The other sites did not allow the installation of the bubble traps due to shallow water depth. The traps consisted of inverted funnels (0.65 m internal diameter, covering an area 0.332 m<sup>2</sup>) sealed with butyl-rubber septum. The bottom of the funnel was situated 1.5 m below the water level. The funnels remained on site for 24 h. Accumulated gas volumes during the sampling period were collected manually through a butyl-rubber septum using a 100 ml gas-tight syringe at the end of the experiment. The total volume of gas collected was estimated from an external scale on each funnel. Methane concentration in bubbles was multiplied by the volume of accumulated gas over the sampling period to determine CH<sub>4</sub> ebullition fluxes. Care was taken to avoid disturbing the benthic substrate during funnel traps operation. All gas samples were stored in 12 ml evacuated soda glass vials (Labco Limited UK) until analysis.

Degassing at weirs was estimated on the basis of methane concentration difference and water discharge according to the calculation given by [Maeck et al. \(2013\)](#) (Eq. (8)).

$$E_{degassing} = [(c_{uw} - c_{dw}) \times Q] / P_{oe} \quad (11)$$

where  $E_{degassing}$  is the emissions at the weir (mol m<sup>-2</sup> d<sup>-1</sup>),  $c_{uw}$  is methane concentration upstream of the weir (mol m<sup>-3</sup>),  $c_{dw}$  is methane concentration downstream of the weir (mol m<sup>-3</sup>),  $Q$  is a water discharge (m<sup>3</sup> d<sup>-1</sup>) and  $P_{oe}$  is an area of overflow edge (m<sup>2</sup>).



## 2.6. Statistics

Data analyses were performed using the software STATISTICA 12 (StatSoft, 2013). Differences between groups of data were examined using a Kruskal–Wallis test. Multiple regression analysis and Spearman's correlation analysis of data were used to find the relationships among variables. All statistical tests used a significance level of 5%.

## 3. Results

### 3.1. Environmental parameters

Water temperature (mean  $20 \pm 0.1$  °C), conductivity (mean  $319.2 \pm 3.2$   $\mu\text{S cm}^{-1}$ ) and dissolved organic carbon (mean  $2.7 \pm 0.2$   $\text{mg l}^{-1}$ ) were relatively stable along the river profile without significant differences between each type of observed sites (Table 2). Mean values of dissolved oxygen concentrations and flow velocity ranged from 5.0 to 12.1  $\text{mg l}^{-1}$  and from 0.06 to 0.6  $\text{m s}^{-1}$ , respectively, with values significantly lower for sections upstream of the weirs compared to river sections downstream of the weirs and unaffected river sections (K-W test;  $p < 0.01$ ) (Table 2). Grain median size was the smallest in sediments upstream of the weirs (mean  $1.8 \pm 0.9$  mm), whereas the biggest grain median size was observed in the river sections with mean  $15.2 \pm 3.7$  mm. Total sediment carbon content was significantly highest upstream of the weirs (mean  $5.5 \pm 0.8\%$ ) compared to other studied sites (K-W test;  $p < 0.001$ ).

### 3.2. Methane oxidation and production by sediments

Consumption of  $\text{CH}_4$  by sediments during the sediment incubation experiments was observed in all samples, with mean values in the range of 0.6–11.8  $\mu\text{mol gDW}^{-1} \text{d}^{-1}$  (Fig. 2a). The  $\text{CH}_4$  oxidation potential of sediments was an order of magnitude higher upstream of the weirs (mean  $10.6 \pm 1.7$   $\mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) compared to sites downstream of the weirs (mean  $1.1 \pm 0.2$   $\mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) and three times higher than the river sections (mean  $3.1 \pm 1.5$   $\mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) (K-W test;  $p < 0.001$ ).

Formation of  $\text{CH}_4$  by sediments wasn't observed in all samples, unlike  $\text{CH}_4$  consumption, and mean values ranged from 0 to 2  $\mu\text{mol gDW}^{-1} \text{d}^{-1}$ . The  $\text{CH}_4$  production potential was the highest for sites upstream of the weirs (mean  $1.5 \pm 0.4$   $\mu\text{mol gDW}^{-1} \text{d}^{-1}$ ), followed by the river sections (mean  $0.14 \pm 0.07$   $\mu\text{mol gDW}^{-1} \text{d}^{-1}$ ), however without significant difference (Fig. 2b). The lowest values of  $\text{CH}_4$  production were observed downstream of the weirs (mean  $0.009 \pm 0.003$   $\mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) (K-W test;  $p < 0.05$ ). Both  $\text{CH}_4$  formation and consumption were strongly positively correlated with sediment carbon content ( $r = 0.63$ ,  $p < 0.05$ ;  $r = 0.80$ ,  $p < 0.05$ , respectively).

The  $\delta^{13}\text{C}$  of organic matter in sediments was  $-27.1 \pm 0.3\%$  ( $n = 27$ ) (Table 3). The  $\delta^{13}\text{C}$  of acetate was  $^{13}\text{C}$  enriched compared to the organic matter with mean  $-22 \pm 2.4\%$  ( $n = 9$ ) in inhibited samples, while there was no observed acetate accumulation in uninhibited samples. The  $\delta^{13}\text{C}$  of  $\text{CH}_4$  measured in the end of the uninhibited incubation

was on average  $6.4 \pm 1.8\%$  more depleted in  $^{13}\text{C}$  than the  $\delta^{13}\text{C}$  of  $\text{CH}_4$  from bubbles trapped in situ. Detailed results of  $^{13}\text{C}$  measurements of sediment samples with sufficient  $\text{CH}_4$  production are given in Table 3.

Contribution of hydrogenotrophic methanogenesis ( $f_{\text{mc}}$ ) to  $\text{CH}_4$  production was calculated for samples with sufficient  $\text{CH}_4$  formation during the incubation experiment (UW 1–3, DW 1, R 1, R 3). In general, the time courses of the contribution of  $\text{H}_2/\text{CO}_2$ -dependent methanogenesis to total produced  $\text{CH}_4$  were relatively constant for individual samples and tended to be higher for sediments upstream of the weirs compared to the sediments from river sections or downstream of the weirs (Fig. 3). Hydrogenotrophic methanogenesis contributed 36.7 to 89.4% of the total  $\text{CH}_4$  production and was dominant ( $>50\%$ ) in sediments upstream of the weirs, while acetoclastic methanogenesis was probably prevailing in remaining sediments.

Despite the different contribution of methanogenic pathways between the sites, the  $\delta^{13}\text{C}$  of  $\text{CH}_4$  released in the end of incubation and in situ (bubbles), the apparent fractionation factor ( $\alpha_{\text{app}}$ ; using Eq. (2)) and fractionation factor for hydrogenotrophic methanogenesis ( $\alpha_{\text{mc}}$ ; using Eq. (3)) were approximately the same for all locations (see Table 3).

### 3.3. Methane in surface water

Methane concentration in the surface water tended to increase along the river stretch (Fig. 4a) with significantly higher concentrations in surface water upstream of the weirs (mean  $1145.3 \pm 127.5$   $\text{nmol l}^{-1}$ ) compared to river sections (mean  $617.2 \pm 139.5$   $\text{nmol l}^{-1}$ ), while methane concentrations in sites downstream of the weirs were not significantly different (mean  $1060.6 \pm 171.8$   $\text{nmol l}^{-1}$ ) (K-W test;  $p < 0.05$ ). Measured  $\text{CH}_4$  surface water concentration at R1 (the most upstream location) was 7.5 times lower than at location DW 3 (the most downstream location). Multiple regression analysis of the data did not reveal any relationship between  $\text{CH}_4$  concentration in surface water and measured environmental parameters in Table 2 ( $F = 2.32$ ,  $p = 0.26$ ).

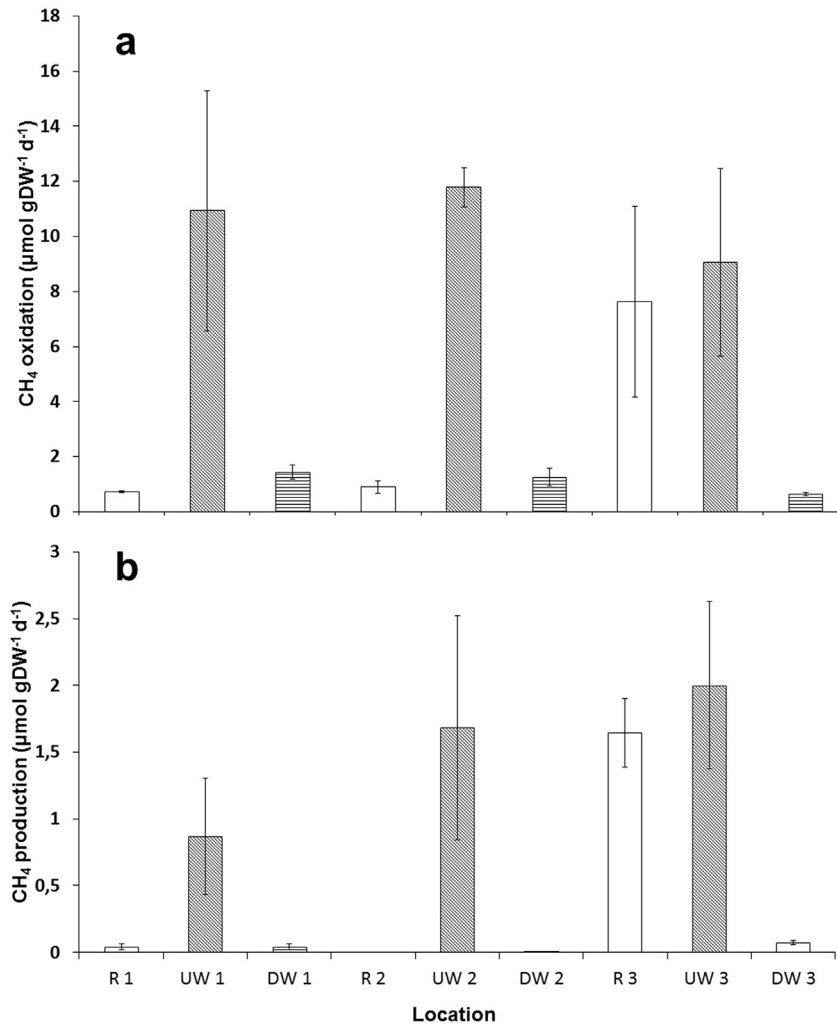
Mean aerobic  $\text{CH}_4$  oxidation potential in the water column of the Morava River varied from 51 to 403  $\text{nmol l}^{-1} \text{d}^{-1}$  (Fig. 4b). Consequently, mean potential  $\text{CH}_4$  turnover time in surface water (black squares, Fig. 4b) was constant across the studied locations with values between 3.9 and 5.6 days, except one location (DW 2), where the oxidation was slower (potential turnover time 15 days). Hence, methanotrophy in the water column was an important process in the  $\text{CH}_4$  budget of most locations.

### 3.4. Methane emissions to the atmosphere

The  $\text{CH}_4$  emissions were determined by direct measurements with floating chambers and bubble traps. However, considering the high contribution of bubbles to total  $\text{CH}_4$  emissions upstream of the weirs, we also calculate the  $\text{CH}_4$  diffusion through the air–water interface using the gas transfer velocity ( $k$ ) and  $\text{CH}_4$  concentration difference between

**Table 2**  
Selected physicochemical properties of studied sites (mean values  $\pm$  SE) (R = river; UW = upstream of the weir; DW = downstream of the weir).

Site	Flow velocity ( $\text{m s}^{-1}$ )	Dissolved oxygen ( $\text{mg l}^{-1}$ )	Dissolved oxygen saturation (%)	Temperature (°C)	Conductivity ( $\mu\text{S cm}^{-1}$ )	Dissolved organic carbon ( $\text{mg l}^{-1}$ )	Grain median size (mm)	Sediment carbon content (%)
R 1	$0.35 \pm 0.03$	$9.7 \pm 0.3$	$109 \pm 5$	$20.6 \pm 0.7$	$305 \pm 15$	$2.5 \pm 0.1$	19.41	$0.2 \pm 0.0$
UW 1	$0.06 \pm 0.01$	$8.4 \pm 0.2$	$95 \pm 3$	$20.5 \pm 0.3$	$319 \pm 7$	$2.5 \pm 0.2$	2.12	$4.9 \pm 1.1$
DW 1	$0.56 \pm 0.03$	$12.1 \pm 1.8$	$135 \pm 20$	$20.1 \pm 0.6$	$320 \pm 10$	$2.7 \pm 0.5$	8.19	$0.2 \pm 0.0$
R 2	$0.35 \pm 0.03$	$9.8 \pm 0.6$	$108 \pm 5$	$20.1 \pm 0.7$	$308 \pm 11$	$2.7 \pm 0.3$	7.85	$0.2 \pm 0.0$
UW 2	$0.10 \pm 0.01$	$7.9 \pm 0.1$	$88 \pm 1$	$20 \pm 0.1$	$316 \pm 1$	$2.8 \pm 0.2$	0.13	$6.2 \pm 0.1$
DW 2	$0.33 \pm 0.02$	$7.7 \pm 0.4$	$86 \pm 5$	$19.9 \pm 0.1$	$316 \pm 0$	$3.1 \pm 0.1$	1.76	$0.3 \pm 0.0$
R 3	$0.18 \pm 0.00$	$8.7 \pm 0.1$	$94 \pm 1$	$19.3 \pm 0.2$	$320 \pm 5$	$2.7 \pm 0.2$	18.37	$4.0 \pm 0.7$
UW 3	$0.07 \pm 0.01$	$5 \pm 0.2$	$56 \pm 2$	$19.9 \pm 0.4$	$340 \pm 3$	$2.7 \pm 0.1$	3.16	$5.3 \pm 1.3$
DW 3	$0.24 \pm 0.02$	$7.6 \pm 1$	$82 \pm 11$	$20.2 \pm 1.6$	$330 \pm 8$	$2.4 \pm 0.1$	10.02	$0.2 \pm 0.0$



**Fig. 2.** Methane oxidation (A) and production potentials (B) of sediments (mean values  $\pm$  SE) calculated from the incubation experiments (UW = upstream of the weir, DW = downstream of the weir, R = river section).

surface water and the atmosphere (Eq. (7)). All values concerning the determination of the CH<sub>4</sub> emissions are given in Table 4.

The CH<sub>4</sub> emissions determined by floating chamber method and by calculation using Eq. (7) were not significantly different in river reaches and downstream of the weirs except the site DW 2. However CH<sub>4</sub> emissions upstream of the weirs obtained by floating chamber method were considerably higher than calculated CH<sub>4</sub> diffusion, because the higher contribution of ebullition to the total CH<sub>4</sub> emissions. Hence we calculated the total CH<sub>4</sub> emission from CH<sub>4</sub> diffusive fluxes (based on *k* and

concentration difference) and from directly measured ebullition by bubble traps.

All sampling sites, especially weirs, were strong sources of CH<sub>4</sub>. Total CH<sub>4</sub> fluxes to the atmosphere (diffusion + ebullition) were in the range of 0.8 to 207.1 mmol CH<sub>4</sub> m<sup>-2</sup> d<sup>-1</sup> (Table 4). The CH<sub>4</sub> emissions were significantly higher upstream of the weirs (mean 68.5  $\pm$  29.9 mmol CH<sub>4</sub> m<sup>-2</sup> d<sup>-1</sup>) followed by sites downstream of the weir (mean 3.4  $\pm$  0.6 mmol CH<sub>4</sub> m<sup>-2</sup> d<sup>-1</sup>) and lowest in river sections (mean 1.6  $\pm$  0.3 mmol CH<sub>4</sub> m<sup>-2</sup> d<sup>-1</sup>) (K-W test; *p* < 0.001).

**Table 3**

Results based on the <sup>13</sup>C values of parameters measured in the CH<sub>4</sub> producing sediment samples (mean values  $\pm$  SE). (UW = upstream of the weir, DW = downstream of the weir, R = river section, n.m. = not measured).

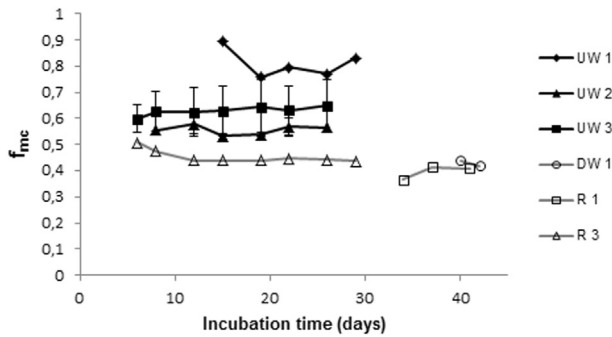
Site	$\delta^{13}\text{C}$ of sediment carbon (‰)	Sediment incubation experiments				In situ samples (ebullition)	
		$\delta^{13}\text{C}$ of acetate (‰) <sup>a</sup>	$\alpha_{\text{app}}$ <sup>b</sup>	$\alpha_{\text{mc}}$ <sup>c</sup>	$\delta^{13}\text{C}$ of CH <sub>4</sub> (‰) <sup>d</sup>	$\alpha_{\text{app}}$	$\delta^{13}\text{C}$ of CH <sub>4</sub> (‰)
UW 1	-28.5 $\pm$ 0.06	-20.9 $\pm$ 0.0	1.040 $\pm$ 0.00	1.068 $\pm$ 0.01	-63.9 $\pm$ 0.0	1.043 $\pm$ 0.01	-54.9 $\pm$ 6.0
UW 2	-28.3 $\pm$ 0.07	-17.7 $\pm$ 3.1	1.051 $\pm$ 0.01	1.079 $\pm$ 0.01	-60.4 $\pm$ 0.9	1.048 $\pm$ 0.00	-59.4 $\pm$ 0.7
UW 3	-28.2 $\pm$ 0.05	-14.3 $\pm$ 5.0	1.054 $\pm$ 0.00	1.078 $\pm$ 0.01	-60.2 $\pm$ 0.1	1.047 $\pm$ 0.00	-52.5 $\pm$ 2.9
DW 1	-27.2 $\pm$ 0.17	-29.8 $\pm$ 0.0	1.049 $\pm$ 0.00	1.081 $\pm$ 0.00	-63.8 $\pm$ 0.0	n.m.	n.m.
R 1	-26.2 $\pm$ 0.05	-29.0 $\pm$ 3.2	1.054 $\pm$ 0.00	1.074 $\pm$ 0.00	-59.8 $\pm$ 0.0	n.m.	n.m.
R 3	-28.3 $\pm$ 0.10	-24.9 $\pm$ 0.7	1.049 $\pm$ 0.00	1.081 $\pm$ 0.01	-61.6 $\pm$ 0.0	1.038 $\pm$ 0.01	-53.8 $\pm$ 10.2

<sup>a</sup> Measured in the samples inhibited by CH<sub>3</sub>F.

<sup>b</sup>  $\alpha_{\text{app}}$  calculated according to Eq. (2).

<sup>c</sup>  $\alpha_{\text{mc}}$  calculated according to Eq. (3).

<sup>d</sup> End of the incubation of the uninhibited samples.



**Fig. 3.** Contribution of  $H_2/CO_2$ -dependent methanogenesis ( $f_{mec}$ ) to total  $CH_4$  production of sediments during incubation experiments.  $f_{mec}$  was calculated using Eq. (4) with measured values for  $\delta_{CH_4}$  (uninhibited incubations) and  $\delta_{mec}$  ( $CH_3F$  inhibited incubations) and calculated values for  $\delta_{ma}$  using Eq. (5) and  $\alpha_{ma} = 1.009$ .

Degassing at weirs (calculated using Eq. (11)) reached mean values of  $3.2 \pm 2.3$ ,  $4.5 \pm 1.4$  and  $0 \pm 0$  mol  $CH_4 m^{-2} d^{-1}$  for site UW 1, UW 2 and UW 3, respectively. Bubble release was recorded for all locations with placed bubble traps (UW 1–3, R 3). Estimated average contribution of ebullition to total  $CH_4$  emissions was 91, 98, 98 and 10% for site UW 1, UW 2, UW 3 and R 3, respectively, implying, that most of the  $CH_4$  upstream of the weirs was transported through air-water interface by ebullition. Individual values of  $CH_4$  content in trapped bubbles ranged from 16.2 to 61.1% and the  $\delta^{13}C$  of  $CH_4$  in bubbles ranged from  $-64$  to

$-31$ ‰ (average  $-55 \pm 9$ ‰). We did not quantify the remaining composition of the bubbles.

#### 4. Discussion

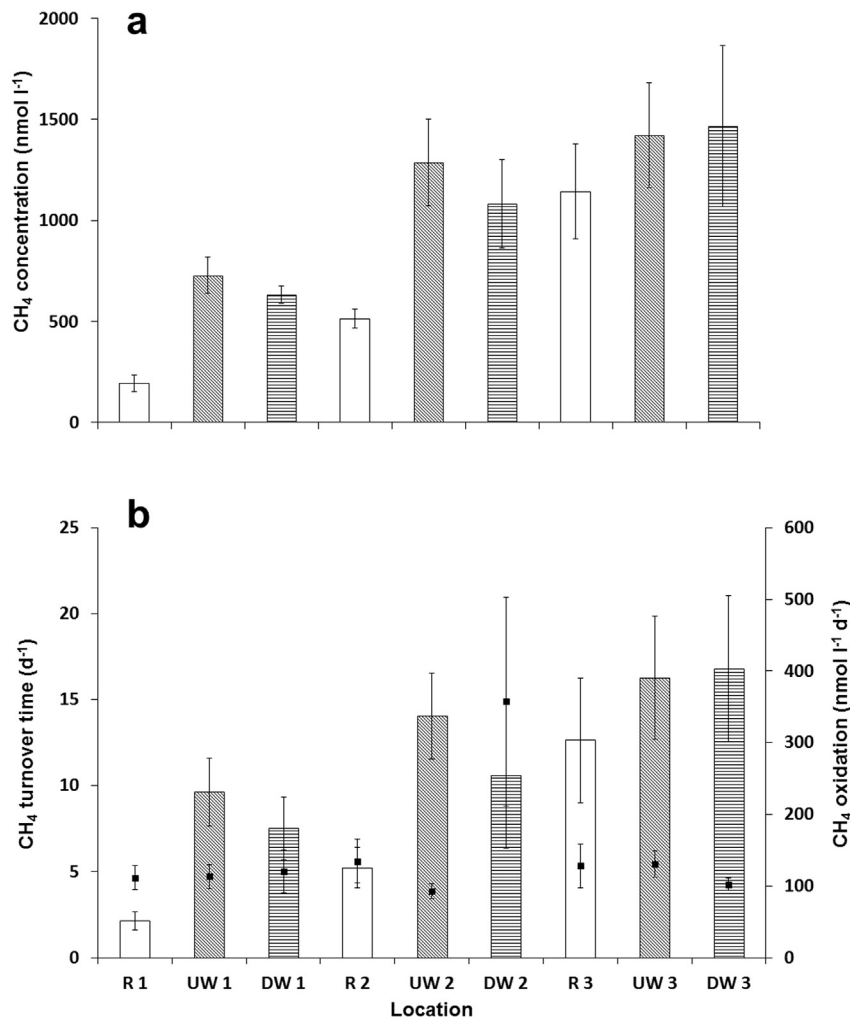
Our results show that dams affect several processes in the  $CH_4$  dynamics of a river continuum. We summarized our findings in Fig. 5 and will discuss the individual processes. First we discuss the sediment samples, then the water samples, and finally the  $CH_4$  emissions to the atmosphere.

However, we did not intend to calculate the  $CH_4$  mass balance, because the  $CH_4$  production and oxidation rates were not measured under in situ conditions and with addition of substrate in case of  $CH_4$  oxidation measurements. Thus they represent potential rates suitable for comparison of studied sites, but they do not allow mass balance calculation.

##### 4.1. Effect of dams on the $CH_4$ dynamics in river sediments

We found that dams across a river resulted in several effects on the sediment processes including changes of the sediment characteristics, enhanced microbial activities in the sediment, ebullition of trace gases, and different contributions of hydrogenotrophic methanogens to the released  $CH_4$ .

In detail, we observed significant changes in the sediment characteristics between individual river reaches in this study: The highest



**Fig. 4.**  $CH_4$  concentrations in surface water (A) (mean value of  $n = 4-8 \pm SE$ ) and aerobic  $CH_4$  oxidation potential in surface water (B) expressed as potential  $CH_4$  turnover time (black squares) and  $CH_4$  oxidation rate (bars) (mean values of  $n = 6-12 \pm SE$ ) along the studied river reach (UW = upstream of the weir, DW = downstream of the weir, R = river section).

**Table 4**The CH<sub>4</sub> fluxes from the surface water to the atmosphere and calculated gas transfer velocity *k* (mean values ± SE).

Site	Directly measured CH <sub>4</sub> emission <sup>a</sup> (mmol m <sup>-2</sup> d <sup>-1</sup> )	Calculated CH <sub>4</sub> diffusive fluxes <sup>b</sup> (mmol m <sup>-2</sup> d <sup>-1</sup> )	Total CH <sub>4</sub> emission <sup>c</sup> (mmol m <sup>-2</sup> d <sup>-1</sup> )	kCH <sub>4</sub> <sup>d</sup> (cm h <sup>-1</sup> )	k600 <sup>e</sup> (cm h <sup>-1</sup> )	kCH <sub>4</sub> <sup>f</sup> (cm h <sup>-1</sup> )
R 1	0.75 ± 0.18	0.83 ± 0.18	0.83 ± 0.27	19.0 ± 2.29	19.3 ± 2.32	18.1 ± 4.41
UW 1	63.4 ± 30.9	0.54 ± 0.15	8.15 ± 2.26	3.14 ± 0.06	3.18 ± 0.06	297.2 ± 225.3
DW 1	2.28 ± 0.32	2.78 ± 0.02	2.78 ± 0.02	18.4 ± 0.27	18.7 ± 0.28	15.2 ± 5.78
R 2	2.26 ± 0.59	2.01 ± 0.38	2.05 ± 0.29	16.4 ± 0.79	16.6 ± 0.80	17.6 ± 6.76
UW 2	176.5 ± 32.7	1.15 ± 0.55	66.1 ± 1.00	3.57 ± 0.53	3.61 ± 0.54	747.2 ± 489.6
DW 2	10.31 ± 1.47	3.67 ± 1.30	3.65 ± 1.25	14.1 ± 0.20	14.3 ± 0.20	46.0 ± 17.9
R 3	1.62 ± 0.60	1.92 ± 0.63	2.13 ± 0.71	7.05 ± 0.10	7.14 ± 0.10	7.69 ± 5.09
UW 3	392.1 ± 108.6	1.27 ± 0.55	131.9 ± 75.2	3.83 ± 0.19	3.89 ± 0.19	1820 ± 1398
DW 3	2.87 ± 0.55	3.81 ± 1.84	3.77 ± 1.77	10.8 ± 0.17	11.0 ± 0.18	13.1 ± 10.4

<sup>a</sup> Diffusion together with ebullition directly measured by chamber method.<sup>b</sup> Calculated according to Eq. (7).<sup>c</sup> The total CH<sub>4</sub> emissions estimated from calculated diffusive fluxes (Eq. (7)) and ebullition measured with bubble traps.<sup>d</sup> Calculated according to Eq. (8).<sup>e</sup> Calculated according to Eq. (10).<sup>f</sup> Calculated according to Eq. (9).

sediment carbon content that was observed upstream of the weirs ( $5.5 \pm 0.8\%$ ), was at the lower end of values reported from lake sediments (1–34%; Huttunen et al. (2006); Duc et al. (2010)). Carbon content was much lower ( $0.8 \pm 0.6\%$ ) in sediments downstream of the weirs and in river reaches, which corresponds to values from other rivers (1.1% Trimmer et al. (2009), 0.2–0.8% Sollberger et al. (2014)).

Organically rich and fine sediments upstream of the weirs resulted in considerably higher values of potential CH<sub>4</sub> consumption and production compared to locations downstream of the weirs and river sections. Slow moving river sections with higher sedimentation rates have been previously shown to result in enhanced methanogenesis and ebullition (Sanders et al., 2007; Maeck et al., 2013).

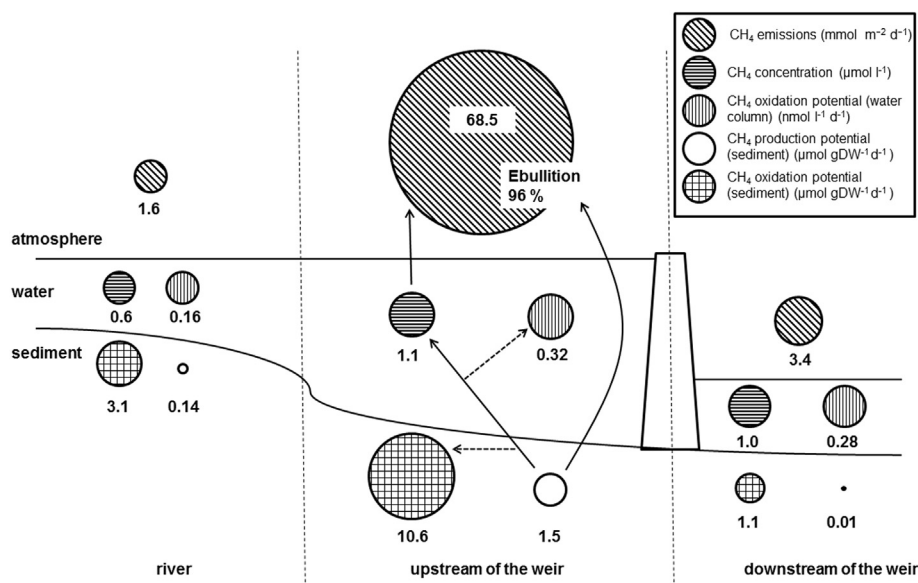
The high methanogenic potential of sediments recorded upstream of the weirs is comparable to organically rich lake sediments ( $0.2$ – $8.4 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ; Conrad et al. (2009); Duc et al. (2010)), while the CH<sub>4</sub> formation in the other locations studied was lower, comparable to other rivers (Gebert et al., 2006; Sanders et al., 2007; Mach et al., 2015). The only exception was observed in location R 3, where high organic carbon content in accumulated sediment and incubation potentials were almost the same as upstream of the weirs, which could be given by local conditions of this river reach. Location R 3 could be

partially affected by backwater region of third weir as indicated by reduced flow velocity and higher organic carbon content in sediment, although median grain size of sediment is high and more comparable to river reaches.

Anoxic conditions and high CH<sub>4</sub> concentrations together with increased contribution of ebullition are positively related to the proportion of fine sediments (Baulch et al., 2011). It is consistent with our results, where we observed significant amounts of CH<sub>4</sub> transported in bubbles from sites with the highest CH<sub>4</sub> production potential and the finest median grain size. The same positive relationship exists between organic matter content in sediments and CH<sub>4</sub> formation (Yang, 1998; Gebert et al., 2006).

Measurements of stable isotopic composition of CH<sub>4</sub> produced during inhibited and uninhibited incubation experiments showed prevailing hydrogenotrophic methanogenesis upstream of the weirs, whereas CH<sub>4</sub> was rather produced from acetate in river reaches. Again the methanogenic pathways upstream of the weirs corresponds more to lake sediments, while the river reaches in this study represent values previously observed in other rivers (see Introduction).

Conrad (1999) and Conrad et al. (2009) provided conceivable explanations for the prevalence of hydrogenotrophic methanogenesis in



**Fig. 5.** Schematic illustration of measured parameters related with CH<sub>4</sub> budget in impounded river. Size of circles for each parameter corresponds to its mean value. Arrows suggest possible fate of CH<sub>4</sub> produced in sediments. Solid arrows show CH<sub>4</sub> diffusion to the surface water and to the atmosphere (or release directly to the atmosphere in bubbles). Dashed arrows indicates potential sink for CH<sub>4</sub> during its transport. Discrepancy between CH<sub>4</sub> production potential and CH<sub>4</sub> oxidation potential in sediments is given by substrate addition (CH<sub>4</sub>) during incubation experiments (see 2.1.). Hence it serves mainly for comparison of microbial activity between locations and not for calculation of net CH<sub>4</sub> emissions from the sediments.

freshwater sediments. It may be due to: non-steady-state conditions (different pace of acetate and H<sub>2</sub> consumption), additional sink of acetate (syntrophic acetate-consuming microbial consortia) or incomplete degradation of organic matter. Generally, the additional sink of acetate in natural habitats is caused by competition of methanogens with ferric iron reducers or sulfate reducing bacteria, which inhibit methanogenesis until depletion of their electron acceptors (Zinder, 1993). The next potential sink is acetate turnover by syntrophic acetate oxidation, which is characterised by a small isotopic effect. Hence, it may easily be overlooked and thus be more widespread than presently assumed (Conrad and Klose, 2011). Nevertheless, syntrophic acetate oxidation takes place mainly in sediments, where the acetoclastic *Methanosaeta* is absent, which is not a usual situation in freshwater sediments (Karakashev et al., 2006). Moreover, we can probably exclude the contribution of homoacetogenesis as additional source of acetate or sink of H<sub>2</sub> and CO<sub>2</sub>, because we observed  $\delta^{13}\text{C}$  of acetate values enriched (~5‰) compared to  $\delta^{13}\text{C}$  of organic matter, while the  $\delta^{13}\text{C}$  of acetate is strongly depleted compared to  $\delta^{13}\text{C}$  of organic matter, when the homoacetogens contribute to acetate production together with fermentation of organic matter. These factors together can theoretically lead to an increased contribution of hydrogenotrophic methanogenesis in our examined sediments; however only partial mineralization of organic matter, which may result in CH<sub>4</sub> production from H<sub>2</sub> and CO<sub>2</sub>, seems to be the most plausible explanation for the dominance of CH<sub>4</sub> production by CO<sub>2</sub> reduction in these sediments as well as in lake sediments (Conrad et al., 2009). This finding together with the high CH<sub>4</sub> production potentials and other sediment characteristics (Table 2) shows that CH<sub>4</sub> related processes upstream of the weirs are more comparable with lake sediments than river sediments.

#### 4.2. Effect of dams on the CH<sub>4</sub> dynamics of river water

One would expect that the impacts of dams on the microbial processes in the sediment described above might also shape other differences, which we find in the water column. Neither the CH<sub>4</sub> concentration nor the methanotrophic activity in the water column, however were dramatically impacted by the weirs.

Our CH<sub>4</sub> surface concentrations observed in this study (0.1–2.2  $\mu\text{mol l}^{-1}$ ) were in the range of values reported from other streams and rivers: 0.01–1.2  $\mu\text{mol l}^{-1}$  (Lilley et al., 1995), 0.01–2.2  $\mu\text{mol l}^{-1}$  (Wilcock and Sorrel, 2008), and 0.04–6.8  $\mu\text{mol l}^{-1}$  (Yvon-Durocher et al., 2011).

The increase of CH<sub>4</sub> concentrations downstream is a usual trend in rivers, when CH<sub>4</sub> sources contribute gradually to the overall CH<sub>4</sub> concentration in surface water (Wang et al., 2009; Anthony et al., 2012; Striegl et al., 2012). Downstream increase of CH<sub>4</sub> concentrations observed in this study is relatively high for such a short river reach, pointing to important CH<sub>4</sub> sources. An even greater increase of surface CH<sub>4</sub> concentration along an 800 m long pool resulting from ponding behind a spillway has been recorded by Lilley et al. (1995), where the concentration changed from 7.7 to 244 nM.

CH<sub>4</sub> concentrations were strongly affected by microbial CH<sub>4</sub> oxidation in the water column without difference between types of locations (i.e. upstream of the weirs, downstream of the weirs, river sections). The potential CH<sub>4</sub> turnover time was about six days, which corresponds to high methanotrophic activity in a temperate river (De Angelis and Scraton, 1993). Undetectable microbial CH<sub>4</sub> oxidation was observed by Anthony et al. (2012) in the Willamette River and by Busmann (2013) in the Lena River. However, there may be considerable differences between rivers or between separate sites of the same river. For instance Lilley et al. (1995) recognized microbial CH<sub>4</sub> oxidation responsible for 25% of methane sink in the Columbia River, but it was negligible in the Wenatchee River. Dzyuban (2011) observed significantly higher rates of bacterial CH<sub>4</sub> oxidation in his examined group of rivers, which were subject to stronger anthropogenic impact, compared with streams without input of organic pollutants.

Considering the high methanotrophic activity in surface water of the Morava River, the occurrence of ebullition is probably more important because CH<sub>4</sub> release in bubbles directly to the atmosphere is not exposed to CH<sub>4</sub> oxidizing bacteria in the water column of shallow impoundments (McGinnis et al., 2006).

#### 4.3. Effects of dams on CH<sub>4</sub> emission

Our values were measured during the summer and over a short term. Thus, they may correspond to an upper maximum of seasonal fluxes and they are not useful for annual estimates of CH<sub>4</sub> emissions (Ortiz-Llorente and Alvarez-Cobelas, 2012; Mach et al., 2016). Although the lower end of this range derived from river reaches (R 1–R 3) is consistent with published values from rivers (Ortiz-Llorente and Alvarez-Cobelas, 2012), the upper limits derived from locations upstream of the weirs (UW 1–UW 3) are considerably larger (mean  $68.5 \pm 29.9 \text{ mmol CH}_4 \text{ m}^{-2} \text{ d}^{-1}$ ). These values are more comparable to surface CH<sub>4</sub> evasion estimates reported from reservoirs (0.2–131  $\text{mmol CH}_4 \text{ m}^{-2} \text{ d}^{-1}$ , DelSontro et al. (2011) or lakes (0.3–312  $\text{mmol CH}_4 \text{ m}^{-2} \text{ d}^{-1}$ , Gonzalez-Valencia et al. (2013).

Moreover, high CH<sub>4</sub> concentrations upstream of the weirs probably influenced CH<sub>4</sub> emissions downstream of the weirs, where emissions were significantly higher than from unaffected river sites. This effect of dams has also been observed by Guerin et al. (2006), where CH<sub>4</sub> concentrations were significantly higher in river sections just below the dam with a decreasing trend downstream of the dams and CH<sub>4</sub> fluxes from downstream rivers were considerably higher than fluxes from the reservoir surface. Although methane loss mediated by degassing at weirs can reach high efficiencies, up to 80% of CH<sub>4</sub> contained in surface water, river downstream of the weirs still remained affected by increased CH<sub>4</sub> concentration (Abril et al., 2005; Richard et al., 2005). The CH<sub>4</sub> concentration decrease between surface water upstream of the weir and downstream of the weir observed in our study was on average 5.5% ranging from 0 to 32%. Consequently, downstream emissions together with degassing at weirs can represent a dominant part of the total emissions (Li and Zhang, 2014) and therefore, it is important to consider all these components when studying the effect of impoundment on CH<sub>4</sub> emissions. Likewise it is clear, that small impoundments significantly increase the importance of rivers as a substantial source of CH<sub>4</sub> to atmosphere.

It has been shown that estimation of CH<sub>4</sub> ebullition and diffusion through the air–water interface is more accurate in case of long term measurements for days or weeks (Joyce and Jewell, 2003; Maeck et al., 2014), because afternoon fluxes could be approximately twice as high as fluxes near sunset and sunrise (Bastviken et al., 2010; Yang et al., 2012). During warmer periods ebullition becomes more important because higher water temperatures, higher CH<sub>4</sub> concentrations and lower gas solubility (Crill, 1996). Ebullition accounted for 74.5% of total CH<sub>4</sub> release to the atmosphere in this study, while wide range of ebullition contribution to total emissions have been reported for rivers (10–80%; Wilcock and Sorrel, 2008; Baulch et al., 2011; Sawakuchi et al., 2014) and lakes (28–98%; Casper et al., 2000; Bastviken et al., 2004; Bastviken et al., 2010).

The CH<sub>4</sub> content in bubbles, which ranged from 16.2 to 61.1%, was comparable to existing literature values from various freshwater ecosystems: streams 26% (Baulch et al., 2011), lakes 44–88% (Casper et al., 2000), and reservoirs 0.001–69% (Deshmukh et al., 2014). Remaining bubble content is usually formed by N<sub>2</sub> and CO<sub>2</sub> (Casper et al., 2000). Ebullition probably contributed to considerable difference in total CH<sub>4</sub> emissions observed between sites upstream of the weirs and river reaches just below the weirs (see above). The CH<sub>4</sub> released in bubbles does not dissolve to surface water in large extent in a shallow water column (McGinnis et al., 2006). Therefore, river sites with similar surface CH<sub>4</sub> concentrations could release significantly different amounts of CH<sub>4</sub> to the atmosphere.

## 5. Conclusions

The CH<sub>4</sub> dynamics of a river continuum are dramatically impacted by the introduction of weir (compare Fig. 5). In general we see increased microbial activities in the weir sediments affecting methanogenic as well as methanotrophic potentials, resulting in increased emissions at the weir (via diffusion and especially ebullition) and downstream river reaches. Many parameters found for weirs resemble observations for lake systems (reduced water velocity, fine sediment fraction, higher sediment carbon content, high CH<sub>4</sub> production potentials, ebullition of CH<sub>4</sub>, high contribution of hydrogenotrophic methanogenesis, high methanotrophic potentials in the sediment as well as in the water). In contrast CH<sub>4</sub> production in the sediments of the other river sections was an order of magnitude lower and characterised by a higher contribution of acetoclastic methanogenesis. The CH<sub>4</sub> emissions within the 16 km river stretch varied over four orders of magnitude, with a peak for the weirs and lower emissions from the other river sections.

Altogether our results confirm that the CH<sub>4</sub> dynamics in a river system show a high local variability and that multiple measurements are needed to characterize the sources and fates of the CH<sub>4</sub>. Overall, CH<sub>4</sub> emissions from average weir impoundment can reach up to 42 times higher values than from the river section of an equal area. Therefore, sampling carried out regardless of occurrence of small impoundments can considerably underestimate CH<sub>4</sub> emissions to atmosphere from lotic ecosystems and the importance of these barriers for river CH<sub>4</sub> dynamic.

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### **7.3. Methane formation and consumption by sediments in the cross-channel profile of the small river impoundment**

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## *Methane dynamics in aquatic sediments*

### **Methane formation and consumption by sediments in a cross-channel profile of a small river impoundment**

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#### **ABSTRACT**

Rivers are a natural source of methane (CH<sub>4</sub>) into the atmosphere and may contribute significantly to total CH<sub>4</sub> emissions. Even though the details of sources of CH<sub>4</sub> in rivers are not fully understood, weirs have been recognized as a hotspot of CH<sub>4</sub> emissions. In this study, we investigated CH<sub>4</sub> production and consumption in air-exposed river sediments along a cross-channel transect located upstream of a weir. Stable carbon isotopes were used for determination of individual methanogenic pathways. In order to understand the relationship between physicochemical and biological processes, additional parameters such as organic matter, grain median size, and carbon and nitrogen content were characterized as well. Generally, samples from the surface sediment layer (0-10 cm) had higher CH<sub>4</sub> production than sediments from the deeper layer (10-20 cm) during the incubation experiments. Sediments near the bank zones and in the mid-channel were characterized by the highest organic carbon content (6.9 %) as well the highest methanogenic activity (2.5 mmol g<sup>-1</sup> DW d<sup>-1</sup>). The CH<sub>4</sub> production was predominated by H<sub>2</sub>/CO<sub>2</sub> dependent methanogenesis in the surface sediment layer (0-10 cm),

while the proportion of acetoclastic and hydrogenotrophic methanogenesis in the deeper sediment layer (10-20 cm) was balanced. The CH<sub>4</sub> oxidation potential of sediments showed the same spatial pattern as observed for the CH<sub>4</sub> production. Our results showed high spatial variability of sediment CH<sub>4</sub> production and oxidation in the cross-channel profile upstream of the weir, whereas the highest CH<sub>4</sub> dynamics were observed in the littoral zones. This variability was closely linked with the carbon and nitrogen content in the sediment samples.

## INTRODUCTION

Inland freshwater habitats including streams and rivers have been recognized to be an important source of methane (CH<sub>4</sub>) into the atmosphere (Bastviken *et al.*, 2011; IPCC, 2013; Deemer *et al.*, 2016). Recent studies show that impounded river zones are CH<sub>4</sub> emission “hotspots”, significantly enhancing our estimations of CH<sub>4</sub> emissions from rivers (Maeck *et al.*, 2013, Wilkinson *et al.*, 2015). River reaches immediately upstream of impoundments are characterized by significantly changed physicochemical parameters of the water, creating transitions between lentic and lotic water ecosystems (Gao *et al.*, 2013). For example, Ogbeibu and Oribhabor (2002) found significantly lower water transparency, current velocity and concentration of dissolved oxygen in reservoirs compared to rivers. The CH<sub>4</sub> emitted from these sites is mostly derived from sediment CH<sub>4</sub> production resulting from an increased sedimentation rate and poor vertical mixing, causing dissolved oxygen decline and anaerobic sedimentary activity upstream of the impoundments (Barth *et al.*, 2003; Maeck *et al.*, 2013). However, spatial variability of CH<sub>4</sub> production and oxidation in these environments is poorly understood, as is the contribution of individual methanogenic pathways to the total CH<sub>4</sub> production.

In principal, CH<sub>4</sub> is produced during anaerobic degradation of organic matter in freshwater sediments (Zinder, 1993). Anaerobic degradation of carbohydrates results in two dominant intermediates (H<sub>2</sub>/CO<sub>2</sub> and acetate), which are further processed by two different metabolic pathways of CH<sub>4</sub> production - hydrogenotrophic methanogenesis (using H<sub>2</sub>/CO<sub>2</sub>) and acetoclastic methanogenesis (using acetate). Quantification of the relative contribution of both sources can be made by using stable carbon isotopic signals, due to different <sup>13</sup>C/<sup>12</sup>C fractionation during conversion of CO<sub>2</sub> and acetate methyl to CH<sub>4</sub> (Conrad, 2005). Contribution of these methanogenic pathways to total CH<sub>4</sub> production differs in various freshwater habitats. A fraction of hydrogenotrophic methanogenesis is usually prevailing in lake sediments (Conrad

*et al.*, 2011), while acetoclastic methanogenesis dominates in rice paddy soils (Scavino *et al.*, 2013) and peatlands (Galand *et al.*, 2010).

Studies dealing with the spatial variability of CH<sub>4</sub> production in the sediments of lakes, reservoirs and rivers emphasize littoral zones as main sites of methanogenic activity (Bastviken *et al.*, 2008; Musenze *et al.*, 2014; Yang *et al.*, 2014). Murase *et al.* (2005) found that littoral sediments of lakes can reach substantially higher CH<sub>4</sub> production than profundal sediments, and further served as a source of dissolved CH<sub>4</sub> in lakes together with tributary rivers. Increased CH<sub>4</sub> release from littoral sediments is likely given by (1) greater availability of labile organic matter from the aquatic vegetation, (2) wave turbulence and bottom shear stress, which enhance sediment flux rates, and (3) higher temperatures in summer months, which in turn support higher CH<sub>4</sub> production rates (Bussmann, 2005; Hofmann *et al.*, 2010).

Similarly, there is an evident spatial distribution of surface water CH<sub>4</sub> concentration in large rivers. Richey *et al.* (1988) and Anthony *et al.* (2012) have shown CH<sub>4</sub> cross-channel gradients with increased CH<sub>4</sub> concentrations observed nearby the banks compared to mid-channel, while Sawakuchi *et al.* (2014) observed a different trend in the Amazon and Pará rivers, with high mid-channel CH<sub>4</sub> concentrations and fluxes. However, the cross-channel variability of CH<sub>4</sub> is not usually included in river studies because the mixing of the entire water column in streams and rivers is assumed. Studies dealing with the spatial distribution of CH<sub>4</sub> in the sediments of rivers report higher CH<sub>4</sub> concentration in pore water of nearshore and riparian habitats, while the hyporheic sediments in mid-channel have usually lower CH<sub>4</sub> concentrations (Jones *et al.*, 1995; Crawford *et al.*, 2014). This pattern probably results from the different rate of water exchange between surface water and sediments, leading to oxygen depletion in the uppermost sediment layer of nearshore habitats (Malard *et al.*, 2002), while the sediments on the central river bottom are oxygenated to a large depth due to rapid vertical hydrological exchange with the flowing water column (Fischer *et al.*, 2005). Consequently, oxygen depletion together with high sedimentation rate and supply of allochthonous labile organic matter from the riparian vegetation creates suitable conditions for high methanogenic activity in nearshore sediments (Jones *et al.*, 1995; Jones and Mulholland, 1998; Stanley *et al.*, 2016). However, it should be noted that microbial activity in total (i.e. including aerobic and anaerobic bacterial metabolism) is highest in the central channel, only due to connectivity with the surface water, which supplies the organic matter into deeper sediment layers (Fischer *et al.*, 2005).

Our previous study revealed a significant effect of weir impoundments on methane river dynamics (production, oxidation, emission, methanogenic pathways) compared to usual river reaches (Bednařík *et al.*, 2017), but a study describing the cross-channel variability of CH<sub>4</sub> in

the sediments upstream of weirs is not known to us. Hence, the overall aim of this study was to get more detailed information about the spatial variability of CH<sub>4</sub>-related processes in a river impoundment. For this purpose, we examined 1) CH<sub>4</sub> production and oxidation rates of the sediments; 2) relationships between environmental variables; and 3) contribution of the methanogenic pathways to total CH<sub>4</sub> production using stable carbon isotopes, all in a cross-channel profile and two different sediment depths of a small river impoundment in Central Europe.

## **METHODS**

### **Study site and sampling**

The study area was located upstream of a weir situated in the Morava River, Czech Republic (Fig. 1; 49°35'12''N, 17°15'43''E). The Morava River is a seventh-order river (according to Strahler, 1957) with a mean annual water discharge of 26.4 m<sup>3</sup> s<sup>-1</sup> in the area of our study site. The impoundment upstream of the weir is 40 m width at its widest point, and has a maximum depth of 3.2 m. The backwater length is approximately 2.6 km. Riparian vegetation is composed mainly by grasses, including reeds and willows. The river channel was without any aquatic macrophytes.

Sediment samples were collected along the cross-section profile of the impoundment (Fig. 2). Triplicates were taken by a piston corer from eight different distances from the bank line to the mid-channel (0 m, 2 m, 4 m, 6 m, 8 m, 10 m, 12 m, 14 m) and from two sediment depths (0-10 cm and 10-20 cm) for each distance. Samples were collected during artificial reduction of the water level in July 2014 caused by weir manipulation, and allowing efficient and accurate sediment sampling when the sediments were shortly exposed to air.

### **Incubation experiments**

Sediments intended for incubation experiments were sieved through a 1-mm sieve to remove coarse detritus, stones or invertebrates, and stored at 4 °C until subsequent analyses and laboratory experiments were carried out. Samples for granulometric analysis were dried and then sieved through a system of ten sieves of decreasing mesh sizes. All separate fractions of the sediment grain sizes were weighed, and grain median size was analyzed using the software Gradistat (version 8.0) (Blott and Pye, 2001). The C, N, and H content of the sediments was quantified on a CHNS-element analyzer (vario MICRO cube, Hanau) by the Analytical Chemical Laboratory of the University of Marburg, Germany. The dry weight of the sample was determined gravimetrically.

For determination of CH<sub>4</sub> production potential and methanogenic pathways, approximately 30 g (wet weight) of the sediments were transferred into 60-ml sterile serum bottles in triplicates, flushed with N<sub>2</sub>, closed with butyl rubber stoppers and incubated at 25 °C in a dark room. At the start of the incubation (before flushing with N<sub>2</sub>), 5 ml of distilled autoclaved water was added into each bottle for sampling of the liquid phase. The liquid phase was sampled at the end of the incubation for analyses of concentration and δ<sup>13</sup>C of acetate. The gas headspace of half of the bottles was supplemented with 3% CH<sub>3</sub>F to specifically inhibit acetotrophic methanogenesis (Janssen and Frenzel, 1997). Gas samples (200 µl) were taken repeatedly (twice a week) during the course of incubation (4-6 weeks) and analyzed for concentrations of CH<sub>4</sub>, CO<sub>2</sub>, and δ<sup>13</sup>C of CH<sub>4</sub> and CO<sub>2</sub>. The CH<sub>4</sub> concentration was analyzed by gas chromatography (GC) using a flame ionization detector (Shimadzu, Kyoto, Japan) and CO<sub>2</sub> concentration was analyzed after conversion to CH<sub>4</sub> with a methanizer (Ni-catalyst 350 °C, Chrompack, Middelburg, the Netherlands).

Twenty grams (wet weight) of sediment samples for determination of the CH<sub>4</sub> oxidation potential were placed in sterile bottles (250 ml) in triplicates, closed by a cap with PTFE silicone septa with ambient air in the headspace and then supplemented with CH<sub>4</sub> to give a final concentration of 10 000 ppm. The incubation was performed at 25 °C in a dark room. The concentration of CH<sub>4</sub> in the headspace of each bottle was measured at 0 h and then ten times over 190 h. The CH<sub>4</sub> production and oxidation potentials were calculated from the slope of CH<sub>4</sub> concentration change over time.

### **Isotopic analyses and calculations**

Isotope measurements of <sup>13</sup>C/<sup>12</sup>C in gas samples were performed on a gas chromatograph combustion isotope ratio mass spectrometer (GC-C-IRMS) system (Thermo Fisher Scientific, Bremen, Germany). The principal operation has been described by Brand (Brand, 1996). Other details are given in Penger *et al.* (2012) and Blaser *et al.* (2013). Isotopic analysis and quantification of acetate were performed on a high pressure liquid chromatography (HPLC) system (Spectra System P1000 [Thermo Fisher Scientific, San Jose, CA]; Mistral [Spark, Emmen, Netherlands]) equipped with an ion-exclusion column (Aminex HPX-87-H, BioRad, München, Germany) and coupled to Finnigan LC IsoLink (Thermo Fisher Scientific, Bremen, Germany) as described by Krummen *et al.*, 2004. Isotope ratios were detected on an IRMS (Finnigan MAT Deltaplus Advantage).

Isotopic calculations of fractionation factors and estimation of the approximate partition of hydrogenotrophic methanogenesis to total methanogenesis were calculated according to a previously published procedure (Conrad, 2005; Blaser and Conrad, 2016).

In principal, partition of hydrogenotrophic methanogenesis was calculated by the following mass balance equation:

$$f_{mc} = (\delta_{CH_4} - \delta_{ma}) / (\delta_{mc} - \delta_{ma}) \quad (\text{eq. 1})$$

where  $f_{mc}$  is the fraction of hydrogenotrophic methanogenesis;  $\delta_{CH_4}$  is the directly measured isotopic signature of the carbon in  $CH_4$ ;  $\delta_{mc}$  is the carbon isotopic signature of  $CH_4$  solely produced from  $CO_2$  (directly measured from assays inhibited by methylfluoride) and  $\delta_{ma}$  is the carbon isotopic signature of  $CH_4$  solely produced from acetate, the latter calculated from the following equation:

$$\delta_{ma} = (1/\alpha_{ma})(\delta_{ac} + 10^3 - \alpha_{ma}10^3) \quad (\text{eq. 2})$$

where  $\alpha_{ma}$  is the fractionation factor for acetoclastic methanogenesis ( $\alpha_{ma} = 1.009$ ; Govert and Conrad, 2009) and  $\delta_{ac}$  is the measured isotopic signal of acetate.

### Statistical analysis

Data analyses were performed using the software STATISTICA 12 (StatSoft, 2013). The Mann-Whitney U test was used for examination of differences between surface and deeper sediment layers, as well as between individual distances from the bank. Spearman's correlation analysis was used to find the relationship between variables. The significance level of  $P < 0.05$  was applied for all statistical analyses.

## RESULTS

### Sediment characteristics

The median grain size of sediments ranged from 0.3 to 9.8 mm, and it was significantly smaller at the surface sediment layer (0-10 cm; mean  $2.4 \pm 0.95$  mm) compared to deeper sediments (10-20 cm; mean  $6.5 \pm 1.3$  mm) ( $P < 0.05$ ). The carbon content ranged from 0.1 to 7.0 % and it was significantly higher in the surface sediment layer (mean  $2.5 \pm 0.6$  %) compared to deeper sediments (mean  $1.1 \pm 0.5$  %) ( $P < 0.05$ ). Similarly, the nitrogen content ranged from 0.01 to 0.67 % and was significantly higher in the surface sediment layer (mean  $0.24 \pm 0.05$  %) compared to deeper sediments (mean  $0.11 \pm 0.04$  %) ( $P < 0.05$ ). The C/N ratio was almost the same in all samples and both sediment depths, with a mean of  $10 \pm 0.1$  (Tab. 1).

### Methane production and oxidation by sediments

Mean  $\text{CH}_4$  production potential ranged between 0 and  $2.4 \mu\text{mol gDW}^{-1} \text{d}^{-1}$  (Fig. 3A). The methane production of nearshore sediments (0-4 m; mean  $1.3 \pm 0.3 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) was significantly higher compared to the rest of the samples (6-14 m; mean  $0.2 \pm 0.1 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) ( $P < 0.05$ ). In total, the surface sediment layer (0-10 cm; mean  $0.9 \pm 0.2 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) had significantly higher  $\text{CH}_4$  production than the deeper layer (10-20 cm; mean  $0.3 \pm 0.2 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) ( $P < 0.05$ ). Mean  $\text{CH}_4$  oxidation potential of sediments ranged from 0.5 to  $13.3 \mu\text{mol gDW}^{-1} \text{d}^{-1}$  and showed the same spatial pattern as observed for the  $\text{CH}_4$  production potential (Fig. 3B). The nearshore sediments (0-4 m; mean  $7.4 \pm 1.1 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) had higher  $\text{CH}_4$  oxidation potential compared to rest of the sampled sediments (6-14 m; mean  $1.7 \pm 0.6 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) ( $P < 0.05$ ). In addition, the surface sediment layer (0-10 cm; mean  $5.6 \pm 1.1 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) reached significantly higher  $\text{CH}_4$  oxidation than deeper sediments (10-20 cm; mean  $2.0 \pm 0.7 \mu\text{mol gDW}^{-1} \text{d}^{-1}$ ) ( $P < 0.05$ ).

Samples with the highest  $\text{CH}_4$  production rates were characterized by immediate and linear  $\text{CH}_4$  concentration increase during the sediment incubation (Fig. 4A), while the less productive samples were characterized by a lag phase, which takes 10-13 days from the start of the incubation, followed by linear or exponential  $\text{CH}_4$  concentration increase (Fig. 4B). The  $\text{CH}_4$  concentration in the headspace of the vials incubated for determination of the  $\text{CH}_4$  oxidation potential started to decrease after 7 h and continued to decrease linearly over the remaining time of the incubation (Fig. 5). The sediment  $\text{CH}_4$  oxidation and production potentials were strongly positively correlated with the carbon content as well as with the nitrogen content in sediments, while it was negatively correlated with the median grain size (Tab. 2).

## Methanogenic pathways

The mean  $\delta^{13}\text{C}$  of organic matter in sediments was  $-27.9 \pm 0.4$  ‰ VPDB ( $n = 48$ ) (Tab. 1). The  $\delta^{13}\text{C}$  of acetate accumulated in inhibited samples (without acetoclastic methanogenesis) was on average very comparable with  $^{13}\text{C}$  of the organic matter with mean  $-26 \pm 1.2$  ‰ VPDB ( $n = 25$ ). Acetate did not accumulate in uninhibited samples. The stable carbon isotopic composition of  $\text{CH}_4$  ( $\delta^{13}\text{C}\text{-CH}_4$ ) produced at the end of the uninhibited incubation was on average  $-69.0 \pm 1.6$  ‰ VPDB.

The contribution of hydrogenotrophic methanogenesis ( $f_{\text{mc}}$ ) to  $\text{CH}_4$  production was stable during the whole incubation time and ranged from 41 to 75 % for individual samples (Fig. 6). The  $\text{CH}_4$  production was predominated by the  $\text{H}_2/\text{CO}_2$ -dependent methanogenic pathway in the surface sediment layer (0-10 cm) with mean  $f_{\text{mc}} = 56 \pm 0.02$  % at the end of the incubation (Figure 6A). However, the contribution of acetoclastic and hydrogenotrophic methanogenesis to the total  $\text{CH}_4$  production in the deeper sediment layer (10-20 cm) was balanced with mean  $f_{\text{mc}} = 51 \pm 0.05$  % and prevalence of  $\text{CH}_4$  production from acetate at the end of incubation of the three samples (2 m, 12 m, 14 m; Fig. 6B).

## DISCUSSION

### Spatial changes in $\text{CH}_4$ production and oxidation

In this study, we observed the highest methanogenic potential to be in the surface sediment layer (0-10 cm) at the distance of 0-4 m from the banks. A similar pattern regarding the active littoral sediments has been previously reported for lakes as well as rivers and it is mostly given by hydrological isolation and increased sedimentation rate in the nearshore habitats (Fischer *et al.*, 2005; Murase *et al.*, 2005). Moreover, littoral vegetation not only reduces the flow velocity, but can also serve as an additional source of organic matter (Sanders *et al.*, 2007; Stanley *et al.*, 2016). We expected more homogenous sediment parameters in cross-channel profiles due to the overall decrease of the flow velocity upstream of the weir. However, fine grain size and higher organic carbon content in the nearshore sediments observed in our study indicate a substantially increased sedimentation rate in this habitat. Nevertheless, after the nearshore sediments (0-4m), we observed high potential  $\text{CH}_4$  formation and oxidation rates in the mid-channel surface sediments (14 m). The best possible explanation is the accumulation of the fresh sediments in mid-channel due to channel morphology, as the mid-channel habitat is the deepest site in the cross-channel profile. In addition, the sediment characteristics (grain median size, carbon and nitrogen content) of the mid-channel samples were very similar to those from the nearshore habitat.



The methanogenic and methanotrophic potentials of the surface sediment layer (0-10 cm) were higher compared to the deeper sediment layer (10-20 cm). Generally, one would expect that better conditions for methanogens might occur in the deeper sediments, where the penetration of dissolved oxygen from the overlaying river water is lower and alternative electron acceptors such as dissolved  $\text{Fe}^{2+}$ ,  $\text{NO}_3^-$ ,  $\text{SO}_4^{2-}$  are depleted (Zehnder and Stumm, 1988). However, the results of our study suggested that availability of substrate for methanogens is the main factor driving the rate of  $\text{CH}_4$  production, since a strong positive correlation between  $\text{CH}_4$  production potential and organic carbon content exists ( $r = 0.94$ ). High  $\text{CH}_4$  production in the surface sediment layer is not unexpected and has previously been reported in lake sediments (Conrad *et al.*, 2009), river sediments (Mach *et al.*, 2015) and sediments of impounded river zones (Wilkinson *et al.*, 2015).

Another possible factor influencing the  $\text{CH}_4$  production rate is likely the C/N ratio, which is frequently mentioned in studies dealing with the methanogenic activity of sediments. It was found that the total nitrogen content best reflects easily degradable organic substrates available for the methanogens and it is highly correlated with maximum methanogenesis (Yao *et al.*, 1999; Gebert *et al.*, 2006). Duc *et al.* (2010) recognized that the highest potential  $\text{CH}_4$  formation rate is in the sediments with lower C/N ratios ( $<10$ ), while the sediments with higher C/N ratios ( $\sim 20$ ) are characterized by lower  $\text{CH}_4$  formation rates despite the high organic carbon content, which is likely associated with the lability of organic matter. In our study, we observed a similar C/N ratio ( $\sim 10$ ; Tab. 1) across all sediment samples in this study. It may indicate the same source of organic material within all samples, and also confirms that the high variability of methanogenic activity was not caused by the degradability of organic substrates in the examined sediments, but rather by the total amount and the availability of organic substrates for the methanogens.

Despite the higher  $\text{CH}_4$  oxidation potential compared to the  $\text{CH}_4$  production potential of all incubated sediment samples (Fig. 3), sediments represent a source of  $\text{CH}_4$  into the surface water and the atmosphere. As suggested by Bednařík *et al.*, 2017, increased  $\text{CH}_4$  concentrations were observed in surface water, together with high contribution of ebullition to the total  $\text{CH}_4$  emission upstream of the weirs. The discrepancy between  $\text{CH}_4$  production potential and  $\text{CH}_4$  oxidation potential in sediments is given by substrate addition ( $\text{CH}_4$ ) during incubation experiments. Hence, this serves mainly for a comparison of the microbial activity between individual samples and not for calculation of the net  $\text{CH}_4$  flux from the sediments. It would be necessary to measure in-situ  $\text{CH}_4$  benthic fluxes from the sediments to the surface water using

the benthic chamber method in order to determine the net contribution of the sediments to the surface water CH<sub>4</sub> (Sansone *et al.*, 1998; Bednařík *et al.*, 2015).

We assume that short exposure (several hours) of sediments to air during the reduction of water level upstream of the weir (see Methods, above) had no significant effect on the results presented in this study. Several studies have revealed that methanogenic archaea can survive in aerated and dry soils even in numbers similar to the original state (Mayer and Conrad, 1990; Fetzer *et al.*, 1993). Hernández *et al.* (2019) have recently shown that even after sediment desiccation and rewetting, stay rates and pathways of CH<sub>4</sub> production remain similar despite changes in the microbial community composition.

### **Methanogenic pathways**

Generally, CH<sub>4</sub> production consists of three distinct phases: 1) the first is the lag phase (also reduction phase), during which most of the inorganic electron acceptors in the sediments, such as nitrate, sulfate or ferric iron, are depleted and only CO<sub>2</sub> is produced; 2) the methanogenic phase, characterized by strong CH<sub>4</sub> formation which maximally depends on the sediment characteristics; 3) in the third phase (the steady state phase), CH<sub>4</sub> production decreases to the stable and long-term level (Yao *et al.*, 1999; Gebert *et al.*, 2006). We observed no lag phase for the most active sediment samples where the CH<sub>4</sub> production was rapid. An absence of lag phase at the start of the CH<sub>4</sub> production could be explained either by 1) overlap of the reduction and methanogenic phases (i.e. CH<sub>4</sub> production started at a relatively high redox potential before the full depletion of inorganic electron acceptors; Yao *et al.*, 1999); or 2) inorganic electron acceptors were depleted already before the start of the incubation, because the sediments were fully anoxic (Conrad *et al.*, 2009). However, concentration of the alternative electron acceptors was not measured in our study. Nevertheless, it has been previously shown that the total CH<sub>4</sub> production is strongly negatively correlated with the duration of the lag phase (Yao *et al.*, 1999), which can be completely confirmed by the results of our study.

Despite different rates of CH<sub>4</sub> production, the resulting contribution of individual methanogenic pathways was very similar for all examined samples of surface sediments (0-10 cm). The predominant pathway of CH<sub>4</sub> production in surface sediments was the consistent reduction of H<sub>2</sub> and CO<sub>2</sub> (hydrogenotrophic methanogenesis) throughout the examined samples, which is in agreement with our previous results from sites upstream of the weirs, while the river sections are characterized rather by the predominant acetoclastic methanogenesis (Avery and Martens, 1999; Mach *et al.*, 2015; Bednařík *et al.*, 2017). The stoichiometrically given portion of individual methanogenic pathways (66 % from acetate and 33 % from H<sub>2</sub>/CO<sub>2</sub>; Conrad, 1999)

usually fits well on rice field soils (Conrad *et al.*, 2002; Fey *et al.*, 2004), but it can considerably vary throughout the different freshwater ecosystems (Murase and Sugimoto, 2001; Galand *et al.*, 2010; Conrad *et al.*, 2011). Possible explanations for the deviations in this portion are described for instance in Conrad (1999) and Conrad *et al.* (2009). In the case of higher contribution of CH<sub>4</sub> production from acetate, this can be easily explained by homoacetogenesis (reduction of CO<sub>2</sub> with H<sub>2</sub> via the acetyl-CoA) (Mach *et al.*, 2015), while the higher contribution of the hydrogenotrophic pathway to total CH<sub>4</sub> production can be given by several processes. Basically, we can exclude the not-steady-state conditions because the acetate did not accumulate in the uninhibited samples (without the addition of CH<sub>3</sub>F). Syntrophic acetate oxidation is exceptional in freshwater sediments and unlikely to explain the major part of CH<sub>4</sub> production. The most probable explanation remains incomplete degradation of organic matter, i.e. an additional source of H<sub>2</sub>, which deflects the resulting contribution of methanogenic pathways, which has been previously observed for lake sediments (Conrad *et al.*, 2009; Conrad *et al.*, 2011).

It is worth noting that hydrogenotrophic methanogenesis was not prevalent in the deeper sediment layer (10-20 cm), where the contribution of the individual methanogenic pathways was equivalent and acetoclastic methanogenesis was prevalent at the end of the incubation experiments of three samples (2 m, 12 m, 14 m). The shift in the contribution of methanogenic pathways was probably not caused by the lability or availability of organic substrate, because of the similarity of these parameters between examined sediment layers (Tab.1). One would expect that the composition of the microbial community can be important for the determination of methanogenic pathways. However, Conrad *et al.* (2011) have shown that the composition of microbial methanogenic communities does not correspond with the resulting contribution of individual pathways of CH<sub>4</sub> production. In spite of the molecular analysis of the methanogenic marker-gene (*mcrA*), which revealed a significantly different methanogenic community for the top layer in contrast to deeper layers, the contribution of individual methanogenetic pathways was very similar throughout all examined samples in Mach *et al.* (2015). Similarly, Chaudhary *et al.* (2017) have found no relationship between the absolute numbers of the methanogenic community and the level of CH<sub>4</sub> production. However, studies dealing with the varying contribution of methanogenic pathways in the vertical profile of freshwater sediments are very scarce and deserve to be considered in greater detail in further studies.

## CONCLUSIONS

We found that the most productive sites in the impounded river zones are littoral sediments; as was previously reported for different freshwater habitats, including lakes and rivers. However, we also observed substantially high CH<sub>4</sub> production in mid-channel sediments, which is likely due to channel morphology causing the accumulation of sediment in this habitat. Hence, the methanogenic and methanotrophic activity of sediments was associated with sites with the finest median grain size of sediments and were best correlated with carbon and nitrogen content. Our results show that it is necessary to consider the sampling location for better representation of particular water habitats. Sediment samples taken only in the littoral zones of water habitats can significantly misrepresent the further extrapolation of obtained results. Considering the substantial sediment CH<sub>4</sub> production potential upstream of weirs, studies focusing on quantification of direct CH<sub>4</sub> fluxes from sediments to surface water and conducted in-situ are necessary.

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Accepted Article

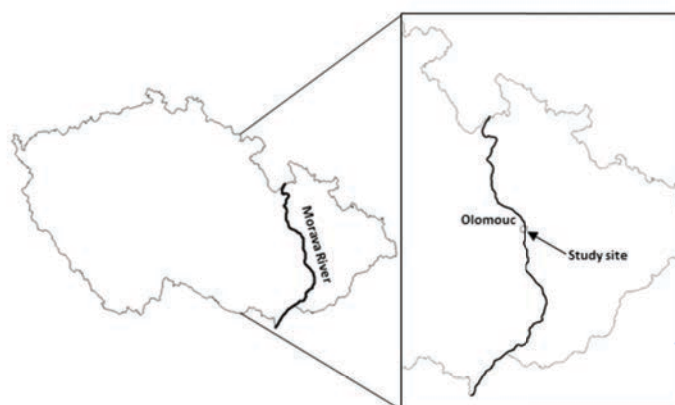


Tab. 1 Measured characteristics of the sediment samples at two different depths (mean values  $\pm$  SE, n = 3).

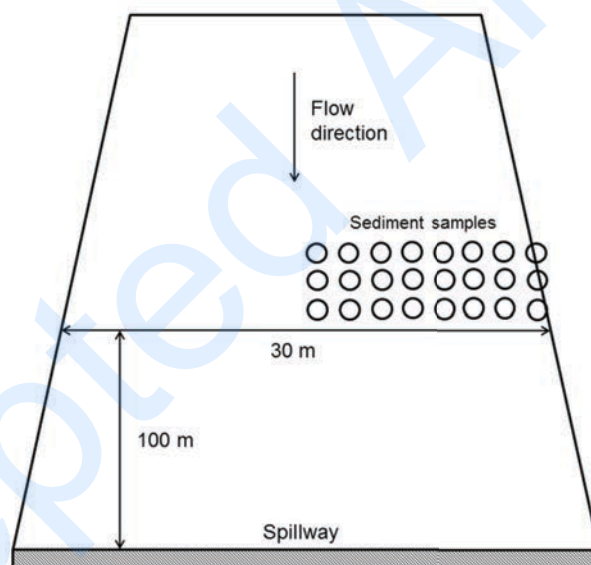
Distance from the bank (m)	Median grain size (mm)	Carbon content (%)	Nitrogen content (%)	C/N ratio	$\delta^{13}\text{C}$ of organic matter
<b>0-10 cm</b>					
<b>0</b>	0.47	6.7 $\pm$ 0.1	0.63 $\pm$ 0.01	10.6 $\pm$ 0.1	-28.3 $\pm$ 0.1
<b>2</b>	0.78	5.7 $\pm$ 0.1	0.53 $\pm$ 0.00	10.6 $\pm$ 0.1	-29.3 $\pm$ 1.4
<b>4</b>	7.70	2.4 $\pm$ 1.3	0.23 $\pm$ 0.12	10.2 $\pm$ 0.2	-28.1 $\pm$ 0.1
<b>6</b>	5.18	0.4 $\pm$ 0.1	0.04 $\pm$ 0.01	10.1 $\pm$ 0.2	-27.2 $\pm$ 0.1
<b>8</b>	0.79	0.2 $\pm$ 0.0	0.02 $\pm$ 0.00	10.1 $\pm$ 0.3	-27.1 $\pm$ 0.1
<b>10</b>	0.78	0.3 $\pm$ 0.1	0.03 $\pm$ 0.00	9.9 $\pm$ 0.1	-27.4 $\pm$ 0.1
<b>12</b>	3.08	0.3 $\pm$ 0.0	0.03 $\pm$ 0.00	9.8 $\pm$ 0.1	-27.2 $\pm$ 0.2
<b>14</b>	0.68	4.0 $\pm$ 1.6	0.39 $\pm$ 0.16	10.1 $\pm$ 0.1	-28.1 $\pm$ 0.1
<b>10-20 cm</b>					
<b>0</b>	0.26	6.9 $\pm$ 0.1	0.66 $\pm$ 0.00	10.4 $\pm$ 0.0	-27.9 $\pm$ 0.1
<b>2</b>	2.45	0.4 $\pm$ 0.1	0.04 $\pm$ 0.01	10.0 $\pm$ 0.1	-27.4 $\pm$ 0.2
<b>4</b>	9.75	0.4 $\pm$ 0.1	0.04 $\pm$ 0.01	10.3 $\pm$ 0.4	-27.3 $\pm$ 0.0
<b>6</b>	9.36	0.2 $\pm$ 0.1	0.02 $\pm$ 0.00	9.6 $\pm$ 0.2	-26.7 $\pm$ 0.1
<b>8</b>	4.46	0.2 $\pm$ 0.0	0.02 $\pm$ 0.00	10.0 $\pm$ 0.1	-26.8 $\pm$ 0.1
<b>10</b>	7.93	0.1 $\pm$ 0.0	0.01 $\pm$ 0.00	8.9 $\pm$ 0.2	-26.6 $\pm$ 0.3
<b>12</b>	9.21	0.3 $\pm$ 0.2	0.03 $\pm$ 0.01	9.8 $\pm$ 0.3	-27.2 $\pm$ 0.3
<b>14</b>	8.86	0.2 $\pm$ 0.0	0.02 $\pm$ 0.00	10.0 $\pm$ 0.2	-27.1 $\pm$ 0.1

Tab. 2 Values of the correlation coefficients (r) expressing the relationship between the examined variables. All correlations shown in the table are significant at  $P < 0.05$  (MPP =  $\text{CH}_4$  production potential; MOP =  $\text{CH}_4$  oxidation potential).

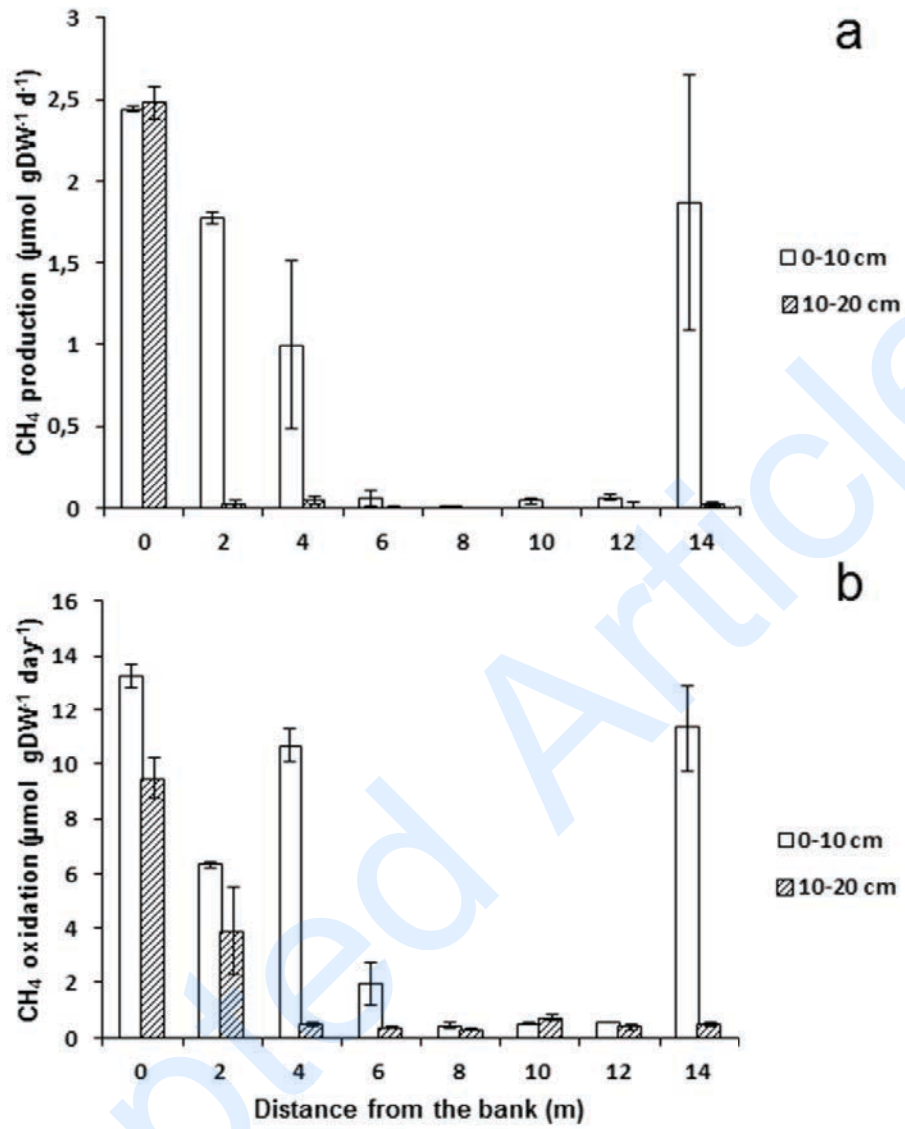
Variable	MPP	MOP	Carbon content	Nitrogen content	Median grain size
MPP	1				
MOP	0.67	1			
Carbon content	0.94	0.68	1		
Nitrogen content	0.94	0.70	0.99	1	
Median grain size	-0.62	-0.52	-0.62	-0.63	1



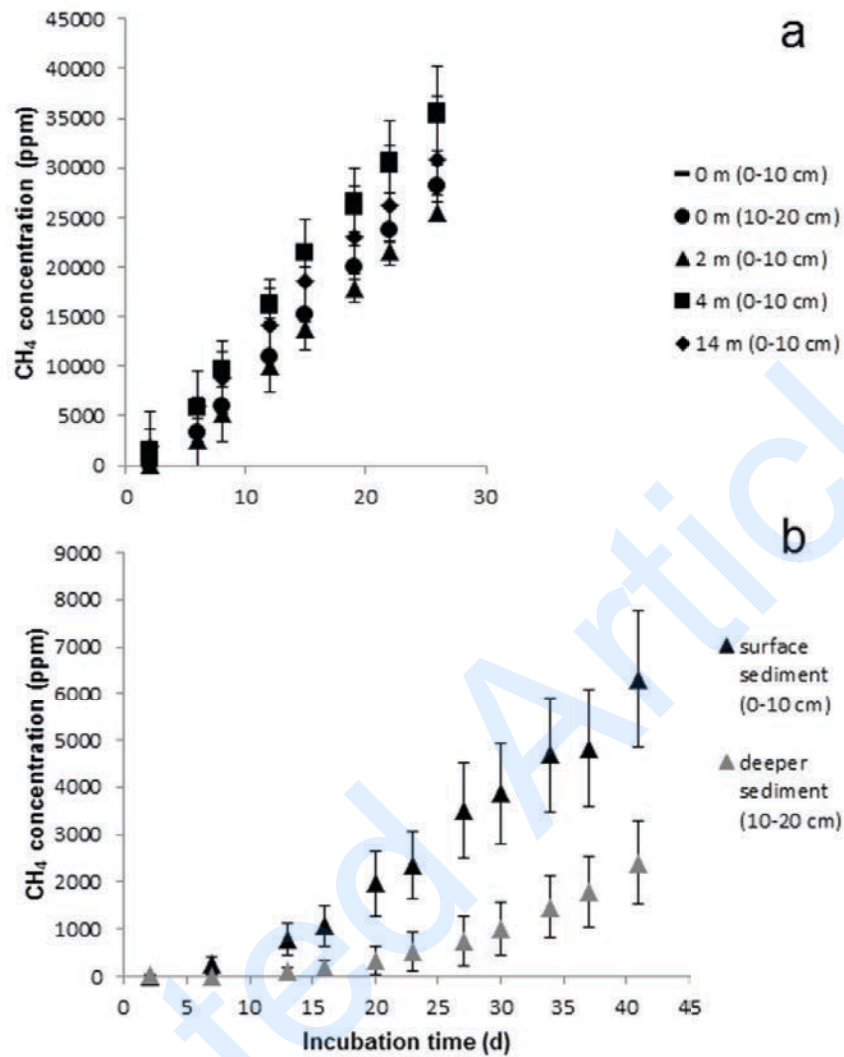
**Fig. 1.** Location of the Morava River in the Czech Republic and position of the study site in the river.



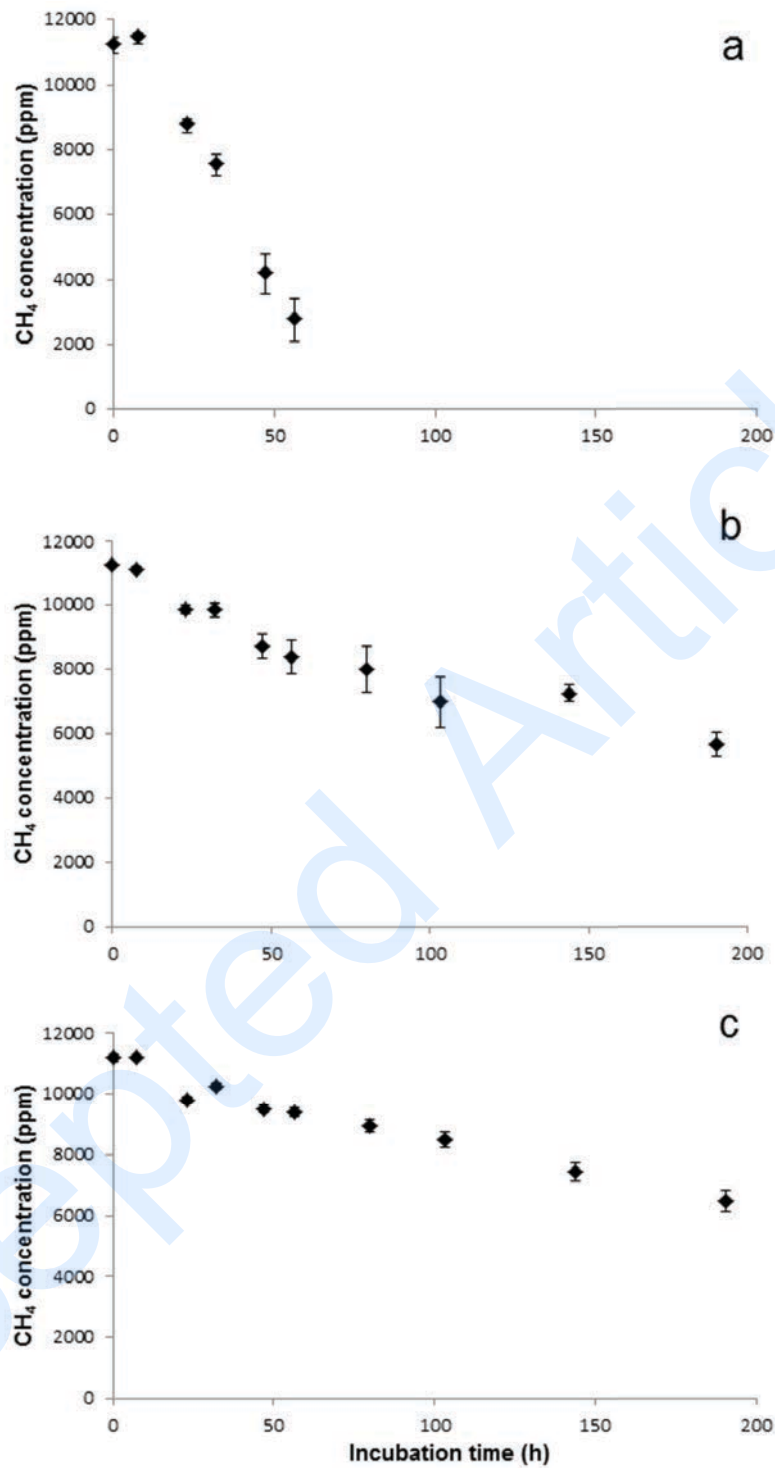
**Fig. 2.** The schematic of the sediment sampling in the cross-channel profile of a small river impoundment. Triplicates were taken from eight different distances from the bank line to the mid-channel (0 m, 2 m, 4 m, 6 m, 8 m, 10 m, 12 m, 14 m).



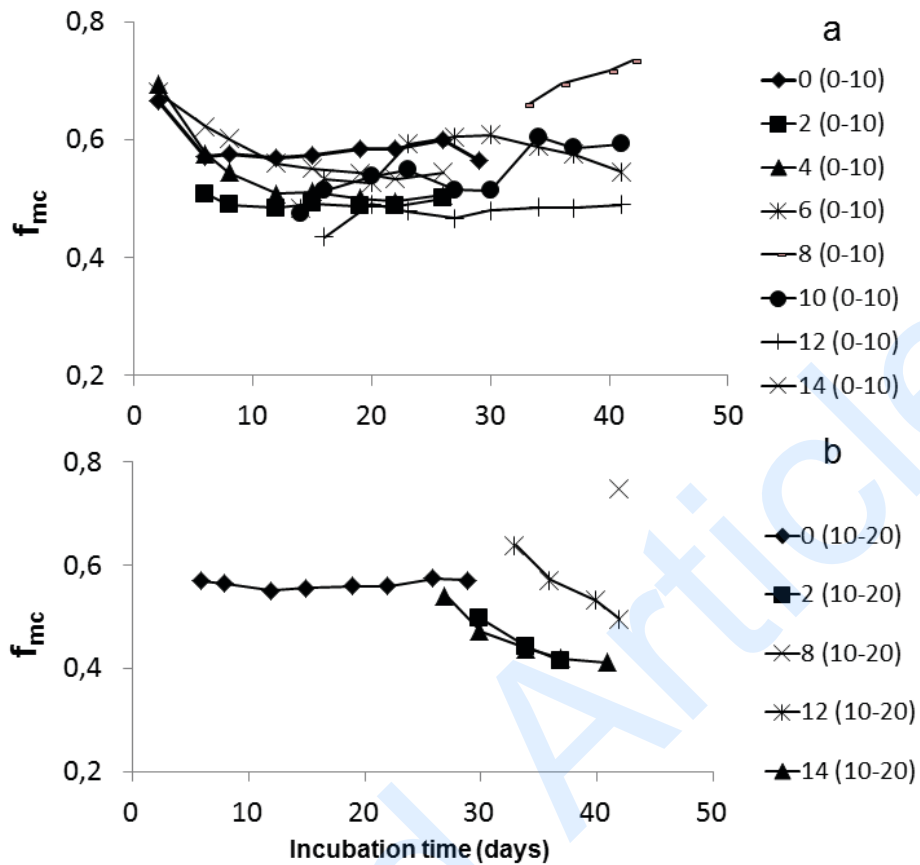
**Fig. 3.** The CH<sub>4</sub> production (a) and oxidation (b) potentials of sediments (mean values ± SE).



**Fig. 4.** Dynamics of CH<sub>4</sub> formation during the incubation of (a) the most productive sediment samples (mean values  $\pm$  SE), and (b) remaining (i.e. less productive samples characterized by a lag phase lasting more than one week after the start of the incubation) surface (6-12 m; black triangles) and deeper (2-14 m; gray triangles) sediment samples in anoxic conditions (mean values  $\pm$  SE).



**Fig. 5.** Methane concentration decrease during the incubation of (a) the most active sediment samples (0 m, 2 m, 4 m, 14 m), (b) remaining surface (0-10 cm) sediments (6-12 m), and (c) remaining deeper (10-20 cm) sediments (6-12 m) in oxic conditions (mean values  $\pm$  SE) .



**Fig. 6.** The contribution of hydrogenotrophic methanogenesis ( $f_{mc}$ ) to total  $CH_4$  production in the samples of (a) the surface sediments (0-10 cm) and (b) the deeper sediment layers (10-20 cm). The missing time points are given by insufficient  $CH_4$  production rate for stable carbon isotopes measurement of the incubated sample.

## **8. Curriculum vitae**

**Name:** Adam Bednařík  
**Born:** March 27, 1989 in Zlín

### **Education**

2013 to present PhD study – Ecology  
Faculty of Science, Palacký University, Olomouc  
2011 – 2013 Master study – Hydrobiology  
Faculty of Science, Palacký University, Olomouc  
2008 – 2011 Bachelor study – Protection and creation of the  
environment, Faculty of Science, Palacký University,  
Olomouc

### **Work experience**

2017 to present The Krkonoše Mountains National Park Administration  
(hydrobiologist)  
2014 – 2015 Department of Ecology and Environmental Sciences,  
Faculty of Science, Palacký University, Olomouc  
(pedagogical worker)

### **Cooperation on research projects**

2015 – 2016 Sources of methane in the river (Endowment Fund of  
Palacký University, Olomouc) - principal researcher of the  
project  
2016 – 2017 EuroRun – Assessing CO<sub>2</sub> Fluxes from European Running  
Waters - participant

### **Abroad stay**

01/2015 – 02/2015 Max Planck Institute for Terrestrial Microbiology,  
Marburg, Germany  
02/2014 – 05/2014 Max Planck Institute for Terrestrial Microbiology,  
Marburg, Germany  
01/2014 Universiti Brunei Darussalam – Kuala Belalong Field  
Studies Centre (KBFSC), Brunei

### **Membership**

2013 to present Czech Limnological Society

### **Publications in peer-reviewed journals with IF**

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# **Importance of stable carbon isotopes for studying dynamics of methanogenesis in rivers**

**Summary of the Ph.D. Thesis**

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### **Abstract**

Methane is one of the most important greenhouse gases. Despite of recent studies pointing out important contribution of running waters to natural methane emissions to the atmosphere, data concerning the methane sources in rivers are very scarce. This thesis deals with methane dynamic in river ecosystems with special emphasis on an effect of river impoundments on methane related processes. Beside the methane concentrations, oxidation, production and emission to the atmosphere, the changes in contribution of two main methanogenic pathways (hydrogenotrophic and acetoclastic) to the total methane production were determined using the stable carbon isotopes analysis. We found hotspots of the sediment methane production in a river continuum, which are connected with the local driving factors including mainly the existence of artificial barriers as weirs. Changes in rate of individual components of river methane dynamic were further examined in cascade of three weirs and river reaches between them. We found that river impoundments affect the sediment processes in several ways, including changes of the sediment characteristics (fine sediment fraction, higher sediment carbon content), enhanced microbial activities in the sediment (methane production and oxidation), ebullition of methane, and different contributions of hydrogenotrophic methanogens to the released methane. Thus, many parameters found for weir impoundments resemble observations for lake systems. Moreover, remarkable spatial variability in sediment methane production was demonstrated in cross-section profile of the one studied impoundment. Presented studies point to only the part of the samples could be activated for methane production despite of presence of methanogens (most probably due to substrate limitation). This suggest that the observed variability of the microbial activities as well as the resulting methane concentrations in the water column are only indirectly linked to the presence of different microbial guilds, but rather affected by their activity. Altogether our results confirm that the methane dynamics in a river system show a high local variability and that multiple measurements are needed to characterize the sources and fates of the methane. Obtained results might be further used for better estimates of importance of rivers in a global methane budget.

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## 1. Introduction

Methane (CH<sub>4</sub>) is one of the most potent greenhouse gases with a global warming potential ~28 times higher than carbon dioxide (CO<sub>2</sub>) over time horizon of 100 years and represents about 15 % of the anthropogenic greenhouse effect (IPCC 2013). From the mid-Holocene to about 300 years ago atmospheric methane concentration rose steadily by about 25 % (Brook et al. 2000). With human population increase and industrialization, methane concentration is now about 250 % higher than it was in the preindustrial age (Etheridge et al. 1998). Current atmospheric methane concentration is 1858 ppb (Dlugokencky 2018).

While the sinks of the methane in the environment are quite clear (reaction with hydroxyl radicals in the troposphere or oxidation by methanotrophic bacteria), the sources of methane are much more diverse. Among main atmospheric methane sources are included wetlands, ruminants, termites, oceans, freshwater sediments, landfills, biomass burning and fossil methane released during fossil fuel extraction (Wuebbles and Hayhoe, 2002). The most of the methane is produced microbiologically, while contribution of freshwater habitats (wetlands, rice fields) creates ~33 % of the annual atmospheric methane flux (Conrad 2009). Moreover, recent studies estimate that annual methane emission from fluvial ecosystems is equivalent to 20-50 % of lake or wetland effluxes (Stanley et al. 2016). However, compared to other natural ecosystems, very scarce data were available for estimation of their global significance and thus importance of lotic ecosystems contribution to global methane emissions from natural environments were overlooked in the past (Saarnio et al. 2009, Bastviken et al. 2011).

Methane dynamic in lotic ecosystems consists of (1) production of the methane within hyporheic sediments, which are place of anaerobic metabolism and formation of methane, (2) subsequent diffusion of methane to the surface water, where the methane is (3) transported downstream or (4) emitted to the atmosphere. The methane is also a subject of significant (5) oxidation by methane oxidizing bacteria during its transport in lotic ecosystems. Methanotrophs are often found at the anoxic/oxic interface of various habitats including freshwater sediments, where they consume the methane arising from methanogenesis and are thus able to reduce the most of the potential methane flux to surface water (Segers 1998, Trimmer et al. 2010). Last but not least, methane has been recently recognised as a potentially important carbon and energy source for freshwater food webs due to conversion of methane to microbial biomass by methane oxidation bacteria, which can be highly productive (Jones and Grey 2011). In rivers, grazing

methane-oxidizing bacteria could provide the caddis larvae (genuses *Agapetus* and *Silo*) up to 30 % of their carbon (Trimmer et al. 2009).

Basically, there are three main sources of methane to surface water of fluvial ecosystems: drainage of surrounding methane rich habitats, groundwater input and river sediments, while exact role of each of these sources is not yet quantified in the overall river C budget. Moreover, water inflows enriched in methane from wastewater treatment plants are significant source of methane in the human influenced rivers (Alshboul et al. 2016). The largest and most ecologically significant pathways of methane efflux from natural environments to the atmosphere are diffusion, ebullition (escape of methane in gas bubbles directly from the sediments) and passage through vascular plants. In natural streams and rivers both methane diffusion and ebullition were recognized to contribute significantly to the total methane emissions from these ecosystems. Moreover, ebullition bypasses the importance of oxidation by methanotrophic bacteria and hence, ebullition of sediment gas bubbles is an important transport process accounting up to 60 % of the total methane emissions from the fluvial ecosystem (Wilcock and Sorrel 2008).

In man-altered rivers, important elements modifying the methane dynamics in lotic ecosystems are artificial impoundments, which reduced flow water velocity and thus increase water residence time that allows organic matter sedimentation and development of anoxic conditions suitable for methane production (e. g. Maeck et al. 2013, Wilkinson et al. 2015, Crawford et al. 2016). Moreover, it was found that smaller impoundments have greater sediment accumulation rates per unit area than the large ones, while small impoundments create a significant part of the total area of impoundments (Downing et al. 2006, Downing et al. 2008). Existence of these artificial barriers play important role in resulted contribution of different methane evasion pathways. Molecular diffusion is usually dominant pathway in rivers (with exceptions described above), while ebullitive emissions are the dominant way for methane emissions from the surface of tropical reservoirs, and it is less significant way for methane emission at the air–water interface in the temperate reservoirs, where the diffusive fluxes are prevailing (Yang et al. 2014).

Generally, the most of the methane produced in nature originates from acetate, however, the relative amounts of methane produced from the methyl group of acetate or reduction of CO<sub>2</sub> can vary depending on the presence of other metabolic groups of anaerobes and the environment (Ferry 1993). Hydrogen should theoretically account for



33 % of total methanogenesis when carbohydrates or similar forms of organic matter are degraded (Conrad 1999). Many methanogenic environments show both much lower and much higher contributions of H<sub>2</sub> to methane production than is considered normal. The relative contribution of the two main methanogenic pathways to total methane production can be calculated due to the sufficient difference in isotopic fractionation during both the hydrogenotrophic and acetoclastic methanogenesis. However, experimental determination of fractionation factors in environmental samples is difficult, since either the hydrogenotrophic or the acetoclastic methanogenic pathway must be suppressed in order to determine the isotope fractionation by one of the two pathways specifically. For that reason, it was important milestone, when the methyl fluoride (CH<sub>3</sub>F) was reported to be a specific inhibitor of acetoclastic methanogens, while hydrogenotrophic methanogens were not affected (Janssen and Frenzel 1997).

## **2. Aims of dissertation thesis**

The aims of this thesis were:

- 1) To characterize the methane production and oxidation potential of the river sediments in longitudinal profile of the Elbe River and to reveal the contribution of individual methanogenic pathways to the total methane production using the stable carbon isotope analysis.
- 2) To compare the rate of the methane related processes (methane production, oxidation, emissions to the atmosphere) between weir impoundments and free flowing river sections of Morava River. Part of it was also quantification of methane ebullition and extent of methane degassing in the spillways.
- 3) To determine the proportion of methane production pathways in the sediments of the examined weir impoundments and river sections.
- 4) To characterize the spatial variability of methane production and consumption by the river sediments including the proportion of the methanogenic pathways in the cross-section profile of weir impoundment.

### 3. Material and methods

The samples collection and field measurements have taken place at the 11 locations along the Elbe River from river km 8 to river km 948 and in the cascade of three weirs in the 16 km long stretch of Morava River. More detailed characterization of studied sites is included in the Paper I and Paper II, respectively. Determination of individual components of methane dynamic in river demands involvement of many different methods, which are individually described in detail in the attached papers.

Briefly, methane production and oxidation potentials of sediments were measured during incubation experiments in the laboratory. Sediment for measurement of methane production potential was incubated in anoxic conditions (headspace of bottle was flushed with nitrogen) approximately one month. Gas samples from headspace were taken repeatedly during the course of incubation (4-6 weeks) and analysed for concentrations of methane. The rate of methane production was calculated from the slope of the linear regression given by the graph of methane concentration increase over time. Sediment for measurement of methane oxidation potential was incubated under the oxic conditions (ambient air in a headspace) with addition of methane. Potential methane oxidation rates were obtained from the slope of the methane concentration decrease over time.

Determination of the contribution of individual methanogenic pathways to total methane production was carried out using the stable carbon isotope measurements. It is based on effect of methylfluorid ( $\text{CH}_3\text{F}$ ), which completely inhibits acetate-dependent methanogenesis. Methane was then exclusively produced by hydrogenotrophic methanogenesis and thus allowed determination of the fractionation factors specific for this methanogenic pathway.

Methane emissions from the surface water to the atmosphere were detected by three different methods fully described in Paper II. First, methane emissions across the air-water interface were directly measured by a floating chamber method. Second, methane diffusion fluxes (i.e. without contribution of the ebullition) to the atmosphere were determined using calculations derived from recent studies and based on the gas transfer velocity and the methane concentration gradient between the river water and the atmosphere (Striegl et al. 2012, McGinnis et al. 2014, Borges et al. 2015, Bodmer et al. 2016). Third, ebullition measurements were carried out using submerged gas funnel traps. Moreover, degassing at weirs was estimated on the basis of methane concentration differences and water discharge. From comparison of first and second

method for measurement of methane emissions in our conditions followed, that we did not observe the increased methane emissions caused by the additional induced turbulence arising from application of anchored chambers described by Lorke et al. (2015). Diffusive fluxes calculated from gas transfer velocity and directly measured emissions by chambers were not different in sites with the marginal contribution of the ebullition.

## **4. Main results**

### **4.1. Sediment methane dynamics along the Elbe River**

The methane production was detected in six sediment samples (from 11 of total) along the Elbe River, while the methanotrophy was found in all examined sediment samples. The methane production and oxidation differed considerably in the river longitudinal profile without any clear trend and without any correlation with other studied environmental parameters. Moreover, the *mcr-A* and *pmo-A* gene copy numbers (genes showing the presence of methanogens and methanotrophs, respectively) were similar and quite stable among the sediment samples. It follows that while it was found hotspots of the measured methane processes, the molecular data showed no spatial characteristics. Incubation experiments with isotopic analyses of CO<sub>2</sub> and methane revealed that the hydrogenotrophic pathway of methane formation (CO<sub>2</sub> reduction) was dominant for all examined methane productive sites accounting for 52 to 78 % of total methane release, thus implying the most probably the incomplete degradation of organic matter.

### **4.2. Effect of weir impoundments on methane dynamics in a river**

We found that river methane dynamics might be highly influenced by weirs, especially by increased methane production and consumption by sediments, followed by increasing methane emissions to the atmosphere. Both methane production and oxidation potential of sediments were higher upstream of the weirs compared to downstream of the weirs or usual river reaches. The total methane emissions to the atmosphere reached the highest values upstream of the weirs, while the ebullition accounted for ~96 % of the total methane emissions. Methane consumption in the sediments together with the microbial methane oxidation in the water column substantially contributed to the methane removal from surface water. Thus, the contribution of the ebullition to the methane emissions in these shallow impoundments was enhanced by bypassing microbial methane oxidation, compared to relatively slow diffusion fluxes. Overall, methane emissions from average weir impoundment can reach up to 42 times higher values than from the river section of an equal area. In spite of such high emission fluxes including further methane release by degassing in the spillways of the weirs and high methane oxidation, considerable 7.5 times increased of methane concentration in the surface water was observed in the 16 km long examined section, pointing to important methane sources in such a short river reach.

The contribution of H<sub>2</sub>/CO<sub>2</sub>-dependent methanogenesis to total produced methane tended to be higher for sediments upstream of the weirs, compared to the sediments from river sections or downstream of the weirs. More precisely, hydrogenotrophic methanogenesis contributed 37 to 89 % of the total methane production and was dominant (more than 50 %) in sediments upstream of the weirs, while acetoclastic methanogenesis was probably prevailing in remaining sediments.

#### **4.3. Methane formation and consumption by sediments in the cross-channel profile of the impoundment**

We hypothesized that more uniform sedimentation in cross-channel profiles compared to river due to the overall decrease of the flow velocity upstream of the weir will result in more identical rates of methane-related processes through the transect. Instead, we found frequently observed pattern in lakes and rivers, that littoral zones are main sites of the methanogenic activity. In our study, sediments near the bank zones and in the mid-channel were characterised by the highest organic carbon content (6.9 %) as well the highest methanogenic activity (2.5 mmol g<sup>-1</sup> DW d<sup>-1</sup>). Samples from the surface sediment layer (0-10 cm) reached higher methane production than sediments from the deeper layer (10-20 cm) during the incubation experiments. The methane oxidation potential of sediments showed the same spatial pattern as observed for the methane production.

Stable carbon isotopes analysis, used for determination of individual methanogenic pathways, confirms our previous findings that the methane production is dominated by H<sub>2</sub>/CO<sub>2</sub> dependent methanogenesis upstream of the weir. However, it was more evident in the surface sediment layer (0-10 cm), while the proportion of acetoclastic and hydrogenotrophic methanogenesis in deeper sediment layer (10-20 cm) was more balanced. This slight shift in the contribution of methanogenic pathways was not caused by the lability or availability of organic substrate because the similarity of these parameters between examined sediment layers.

## **5. Conclusions**

Altogether results presented in this study confirm that river sediments are important place of anaerobic degradation of organic matter with methane as a final product. Moreover, the methane dynamics in a river system show a high local variability, indicating that multiple measurements are needed to characterize the sources and fates of the methane. The methane dynamics along the river continuum is dramatically impacted by the building of small impoundments, which contribute significantly to the total methane production and its subsequent emission disproportionately to their area. Therefore, sampling carried out regardless of occurrence of small impoundments can considerably underestimate methane emissions to atmosphere from lotic ecosystems and the importance of these barriers for river methane dynamic. We found out that the most productive sites in the impounded river zones are littoral sediments as was previously reported for different freshwater habitats including lakes and rivers. Modifications of methane related processes in impounded river zones are reflected also by different contribution of individual metabolic pathways of methane production compared to usual river sections.

Recent studies clearly show that rivers may emit considerable amount of methane to the atmosphere. Hence, rivers should be included into the future estimations and models of a global methane budget. Namely, high spatial variability of methane related processes is remarkable in rivers compared to other ecosystems and deserves to be considered in greater details in the further studies. In any case, it is necessary to consider the sampling location of particular water habitats, because sediment samples taken only in the particular zones of water habitats can significantly misrepresent the further extrapolation of obtained results. Considering the considerable sediment methane production upstream of the weirs, more studies focusing on quantification of direct methane fluxes from sediments to surface water and conducted in-situ would also be of great importance.

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## 7. List of author's papers

The thesis is based on the following papers:

### Paper I:

**Bednařík A**, Blaser M, Matoušů A, Tušer M, Chaudhary PP, Šimek K, Rulík M (2019): Sediment methane dynamics along the Elbe River. *Limnologica* (*in review*)

### Paper II:

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## 8. Souhrn [Summary, in Czech]

Metan je spolu s oxidem uhličitým a oxidem dusným řazen mezi nejvýznamnější skleníkové plyny. Navzdory mnoha současným studiím vyzdvihujícím význam říčních ekosystémů jako přírodního zdroje emisí metanu do atmosféry, jsou poznatky týkající se zdrojů metanu v řekách nedostatečné. Předložená práce se zabývá dynamikou metanu v říčních ekosystémech se zvláštním důrazem na vliv umělých příčných bariér (jezů) na procesy spojené s koloběhem metanu. Kromě koncentrací metanu, jeho oxidace, produkce a emisí do atmosféry byl determinován příspěvek dvou hlavních metabolických cest vzniku metanu do jeho celkové produkce, a to s využitím analýzy obsahu stabilních izotopů uhlíku. Na základě provedených měření v říčním kontinuu řeky Labe byly detekovány místa s nezvykle vysokou produkcí metanu v sedimentech, které odpovídaly změnám v lokálních faktorech prostředí spojených především s existencí příčných bariér na vodním toku. Další práce se proto blíže zaměřila na porovnání změn v jednotlivých složkách dynamiky metanu v kaskádě tří jezů a říčních úsecích mezi nimi. Bylo zjištěno, že jezy vyvolávají řadu změn v sedimentačních procesech (usazování jemnější frakce, vyšší obsah organického uhlíku), které se následně projevují v procesech spojených s koloběhem metanu, jako je zvýšená mikrobiální aktivita (produkce a oxidace metanu v sedimentech), vysoký podíl ebulice na celkové emisi do atmosféry (uvolňování metanu ve formě bublin) a rozdílný poměr metabolických cest vzniku metanu. To spolu s ostatními charakteristikami nadjezí vypovídá o tom, že mnoho procesů probíhajících v sedimentech jezových zdrží včetně tvorby bublin a metabolismu uhlíku je lépe srovnatelných s prostředím sedimentů jezer než s říčními sedimenty. Mimo to byla zjištěna také značná variabilita v produkci metanu uvnitř vybraného nadjezí. Další výsledky ukázali, že jen část vzorků inkubovaných v anoxických podmínkách produkuje metan, přestože v nich bylo detekováno srovnatelné množství metanogenních archaea jako v aktivních vzorcích, což bylo s největší pravděpodobností dáno nedostatkem vhodného substrátu. Toto zjištění naznačuje, že pozorovaná variabilita v mikrobiální aktivitě stejně jako výsledné koncentrace metanu ve vodě jsou jen nepřímo řízeny přítomností určitého mikrobiálního společenstva, ale jsou spíše ovlivněny jeho aktivitou. Naše studie tak potvrzuje, že dynamika metanu v říčních ekosystémech vykazuje vysokou prostorovou variabilitu a z toho důvodu lze charakterizovat zdroje metanu a jeho další osud v ekosystému jen s využitím velmi komplexních měření. Získaná data mohou mimo jiné posloužit i jako

cenný údaj pro zpřesnění odhadů významu říčních systémů v bilanci metanu v rámci vnitrozemských vod a v kontextu globální dynamiky metanu.